

Preliminary Measurements from TORERO with the Trace Organic Gas Analyzer (TOGA)

Eric Apel

Becky Hornbrook – July, 2012

Alan Hills, D. Riemer



NCAR

Key Topics

Atmospheric Chemistry

Emissions, transformations, and redistribution of trace gases affect the global environment

- Oxidative capacity (ability of atmosphere to cleanse itself)
- Ozone (health and ecosystem effects, radiative - climate)
- Radiative and health effects from nucleation and growth of ultrafine particles
- stratospheric/trop ozone depletion from VSL halogenated species

Recent experiment that address unresolved or outstanding issues relevant to these topics and the results of which are globally significant

- **TORERO** – GV/ ship study – CR and Chile – release and transport of trace species from coastal and open ocean waters

Tropical Ocean Troposphere Exchange of Reactive halogen species and Oxygenated VOC

R. Volkamer PI

The scientific objective of the TORERO project is to study the release and transport of halogenated gases and oxidized VOCs in the Eastern Tropical Pacific during the season of high biologic productivity.

TORERO Science Hypothesis

Hypothesis #1: Ocean sources of oxygenated VOC (OVOC) and reactive halogen species (RHS) impact atmospheric composition in the MBL, and in the FT as a result of deep convective transport.

Hypothesis #2: The gas fluxes across the air-sea boundary vary between the oligotrophic and mesotrophic ocean, and coastal upwelling.

Hypotheses #3: Reactive gases released from the ocean are relevant to chemistry and climate.

TORERO: measures for the first time simultaneously bromine and iodine oxide radicals, and organic precursor molecules.

Halogens destroy heat trapping tropospheric ozone, modify oxidative capacity, and oxidize mercury.

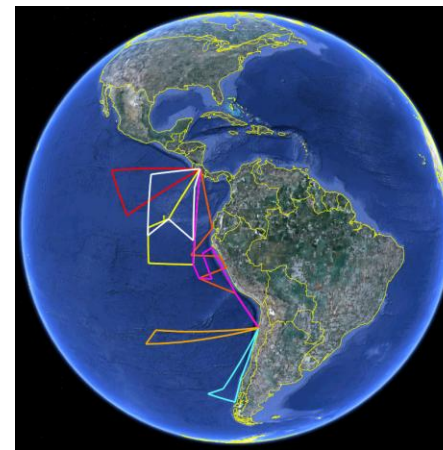
NSF/NCAR GV

CU AMAX-DOAS on HIAPER



Zenith
Forward
(NADIR + 2x slant
down not shown)

NOAA RV Ka'imimoana



The Trace Organic Gas Analyzer (TOGA)

Eric Apel (PI), Alan Hills, Rebecca Hornbrook (ACD/NESL/NCAR)
Dan Riemer (Co-PI; University of Miami)

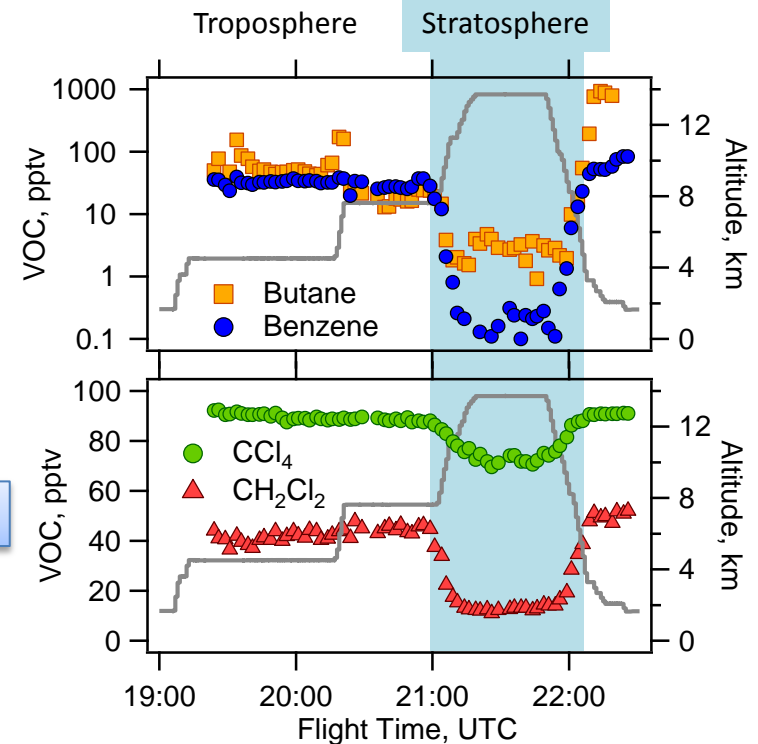
- VOCs needed to understand chemistry leading to trop O_3 and aerosols. Halogenated species can impact both trop and lower strat
- Designed specifically for the G-V
- Maiden research voyage – TORERO 2 min time res
- Designed to have very low LOD ppt to sub – pptv detection limits, over 45 VOC measured simultaneously



Sample data:
DC3 Test Flight 2

Pollutants

Greenhouse gases

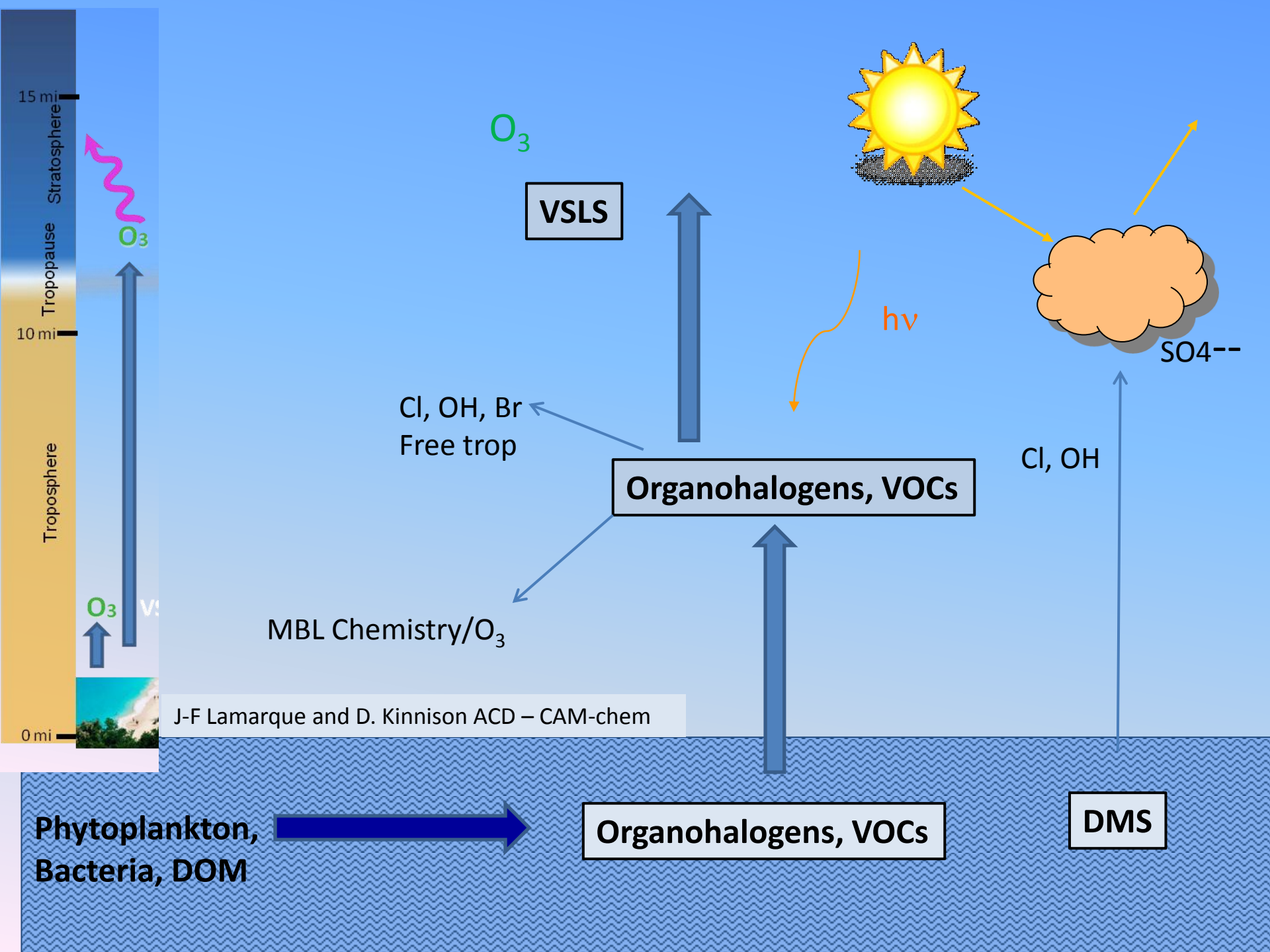


TOGA compounds

Hydrocarbons	Propane 1-Butene <i>i</i> -Butene Butane <i>i</i> -Butane Benzene Toluene Ethyl Benzene <i>t</i> -2-Butene <i>c</i> -2-Butene Pentane 1,3-Butadiene Limonene	Isoprene <i>t</i> -2-Pentene <i>c</i> -2-Pentene <i>i</i> -Pentane <i>o</i> -Xylene <i>m/p</i> -Xylene 1,3,5-Trimethylbenzene 1,2,4-Trimethylbenzene α -Pinene β -Pinene Camphene Myrcene	
Oxygenates	Acetaldehyde Propanal Butanal Pentanal Methacrolein Methyl Vinyl Ketone Methyl Butenol	Methanol Ethanol Acetone Butanone 2-Pentanone 3-Pentanone Methyl t-Butyl Ether	
Halocarbons	Chloroform (CHCl ₃) Methylene chloride (CH ₂ Cl ₂) Methyl chloride (CH ₃ Cl) Methyl bromide (CH ₃ Br) Tetrachloroethane (CH ₂ Cl ₄) Tetrachloroethylene (C ₂ Cl ₄) Bromoform	Tetrachloromethane (CCl ₄) CFC-113 HCFC-141b HCFC-134a 1,2-Dichloroethane (C ₂ H ₄ Cl ₂) Methyl Iodide (CH ₃ I) iodoform	dibromomethane diodomethane bromocjhloromethane bromiodomethane chloriodomethane
Nitrogen and sulfur compounds	Acetonitrile Dimethyl Sulfide (DMS)	DMSO?	

TORERO





Tropospheric Halogen Chemistry

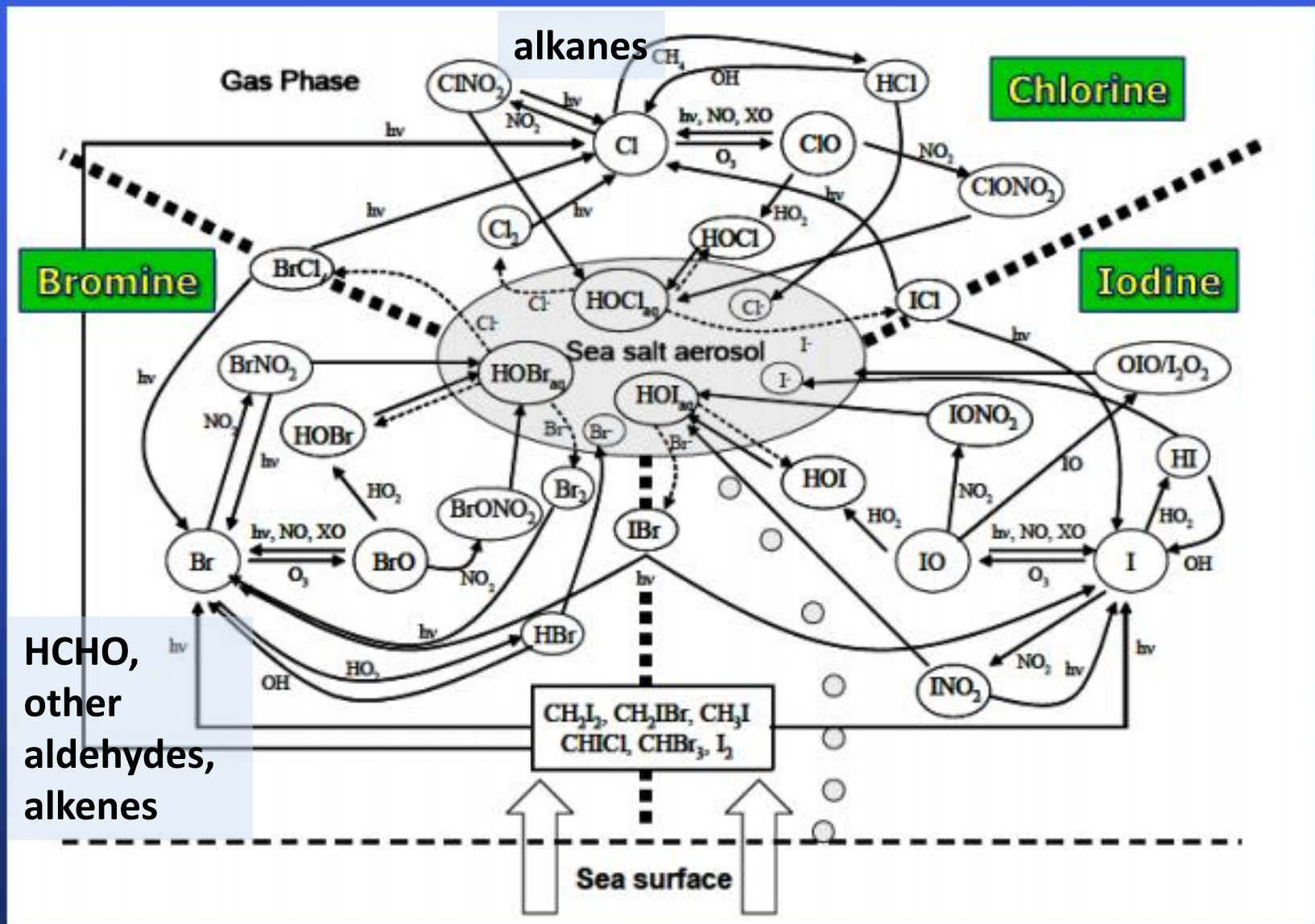


Table 1-4. Lifetimes for very short-lived halogenated source gases.

Montzka, Reimann, et al.

Compound	Local Lifetime from Previous Assessments (τ_{local}), days	OH Lifetime ¹ (τ_{OH}), days	Photolysis Lifetime from Previous Assessments (τ_{local}), days	New Local Lifetime, (τ_{local}), days	Notes
Chlorocarbons					
CH ₂ Cl ₂	140	144	> 15000	144	2, 8
CHCl ₃	150	149	> 15000	149	2, 8
CH ₃ CH ₂ Cl	30	39		39	2
CH ₂ ClCH ₂ Cl	70	65		65	4
CH ₃ CH ₂ CH ₂ Cl		14		14	5
CHClCCl ₂		4.9	> 15000	4.9	3, 8
CCl ₂ CCl ₂	99	90		90	3
CH ₃ CHClCH ₃		18		18	5
Bromocarbons					
CH ₂ Br ₂	120	123	5000	123	2, 8
CHBr ₃	26	76	36	24	2, 8
CH ₂ BrCl	150	137	15000	137	2, 8
CHBrCl ₂	78	121	222	78	6, 8
CHBr ₂ Cl	69	94	161	59	7, 8
CH ₃ CH ₂ Br	34	41		41	2
CH ₂ BrCH ₂ Br	55	70		70	4
n-C ₃ H ₇ Br	13	12.8	> 1200	12.8	3, 8
iso-C ₃ H ₇ Br		16.7		16.7	3
Iodocarbons					
CH ₃ I	7	158	7 (4–12)	7	4, 8
CF ₃ I	4	860	not determined	4	2
CH ₂ ClI	0.1		0.1	0.1	2 hrs 8
CH ₂ BrI	0.04		0.04	0.04	8
CH ₂ I ₂	0.003		0.003	0.003	5 min 8

Preliminary Lucy Carpenter/Stephen Andrews analysis vs. various standards

Compound	vs. NCAR Lab	vs. NOAA Air spike	vs. NOAA gravimetric	vs. NCAR in-flight calcs
CH ₃ I	3.50	2.75		3.71
CH ₂ Br ₂	2.25	2.04	2.39	2.13
CHBr ₃		5.14	5.89	4.60
CH ₂ I ₂	0.83	1.12	0.50	0.72
CH ₂ BrCl	7.43	5.87		6.71
CHBr ₂ Cl		3.53		1.64
CH ₂ IBr	1.79	1.49	2.03	1.72

NOAA Gases

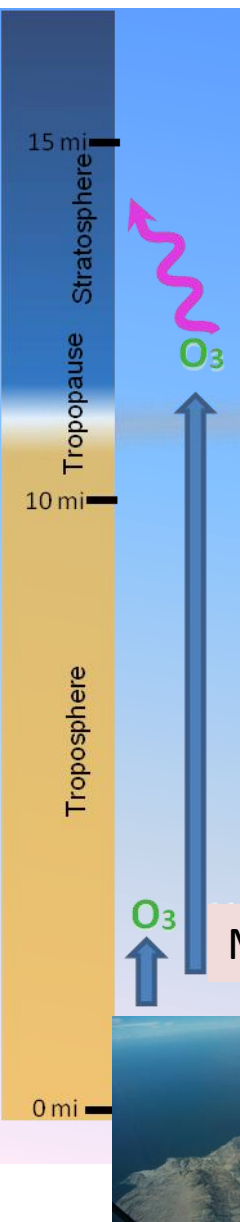


NCAR Gases



Canister Gases





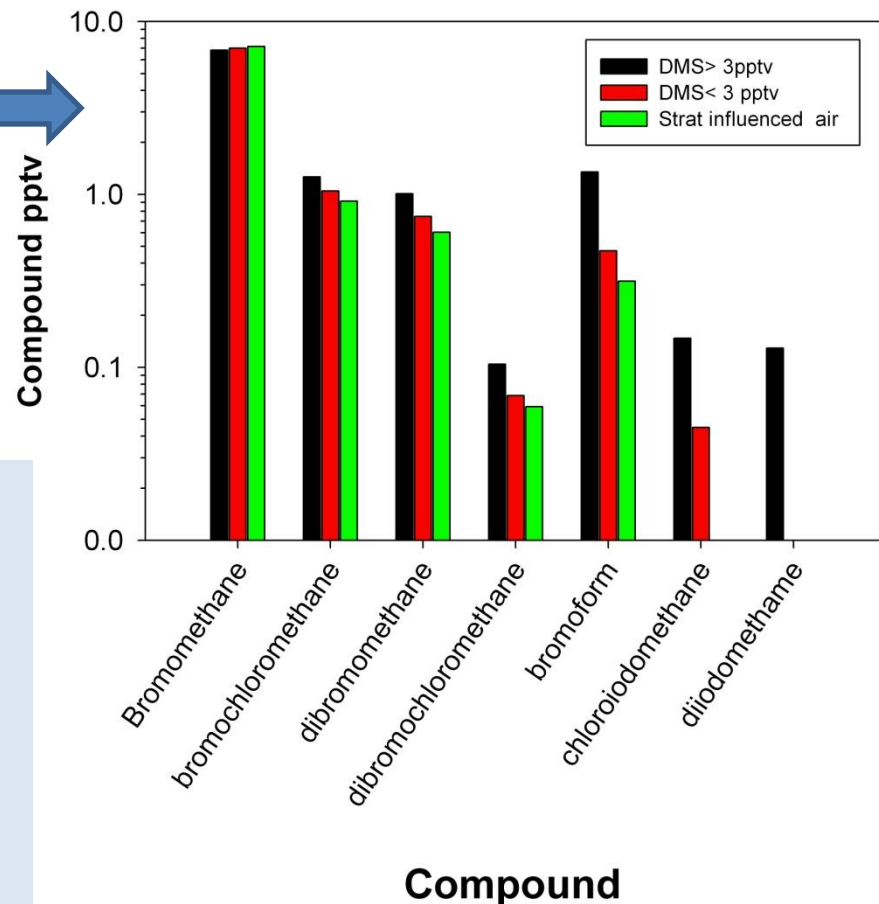
Ultra-high sensitivity needed to investigate some chemical processes such as the inorganic halogen/organo-halogen species – parts per quadrillion sensitivity required (see Carpenter, Atlas, etc.)

Relatively stable organic halogens such as bromomethane, bromoform (CHBr_3) and dibromomethane (CH_2Br_2), emitted predominantly from the oceans, can impact the MBL and be transported to the lower stratosphere and make a contribution to total bromine levels and thus to stratospheric ozone depletion.

Build on previous studies

J-F Lamarque and D. Kinnison ACD – CAM-chem

TORERO TOGA Organohalogen Measurements



VLSL – defined as less than 6 mo.

Bromomethane ≈ 1 year

Bromoform ≈ 1 month

Dibromomethane ≈ 4 months

Chloriodomethane ≈ 2 hours
(LOD TOGA = 0.03 pptv)

Diiodomethane ≈ 5 minutes (LOD TOGA 0.03 pptv)

DMS

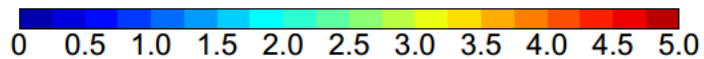
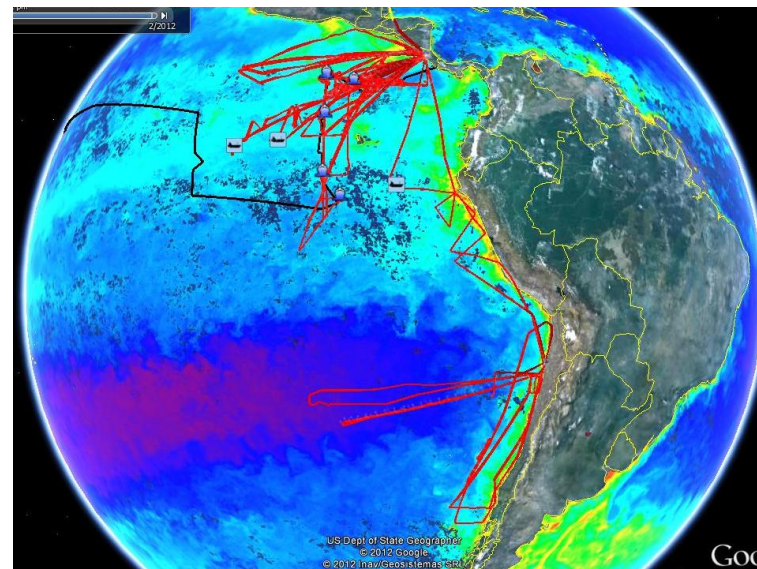
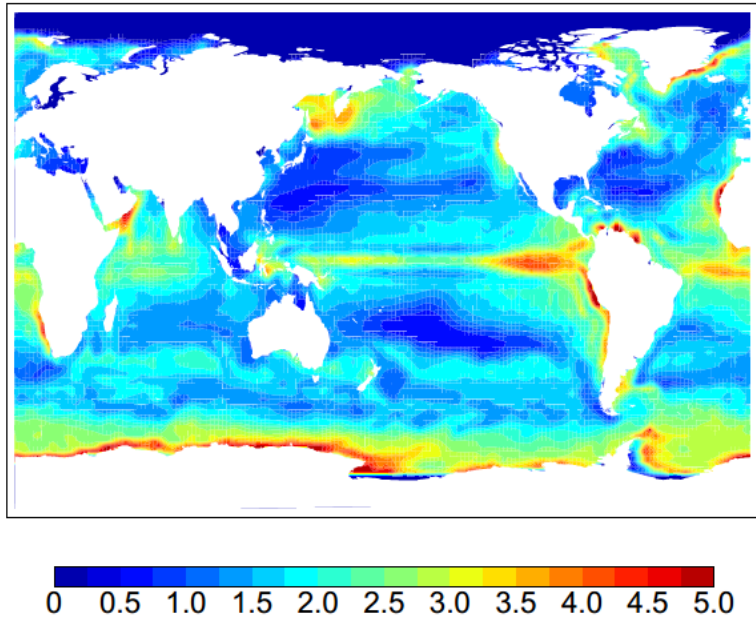
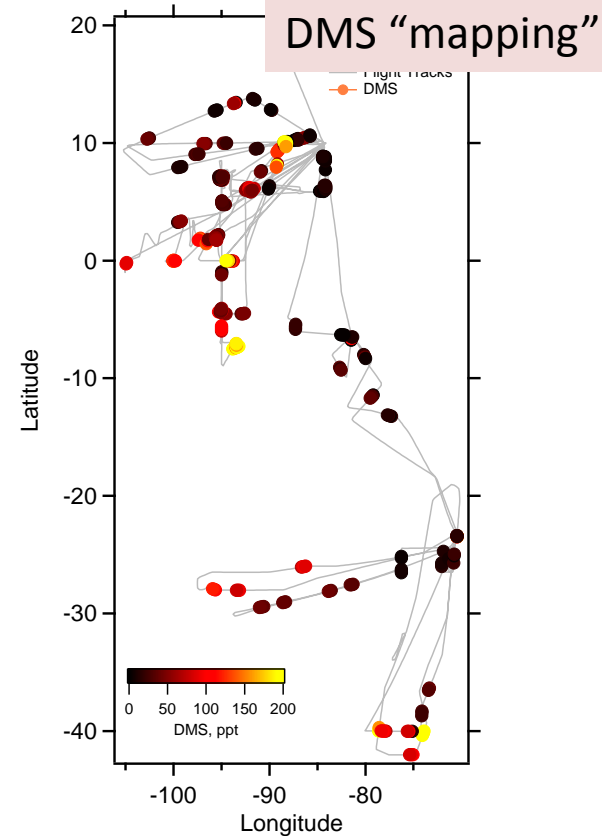
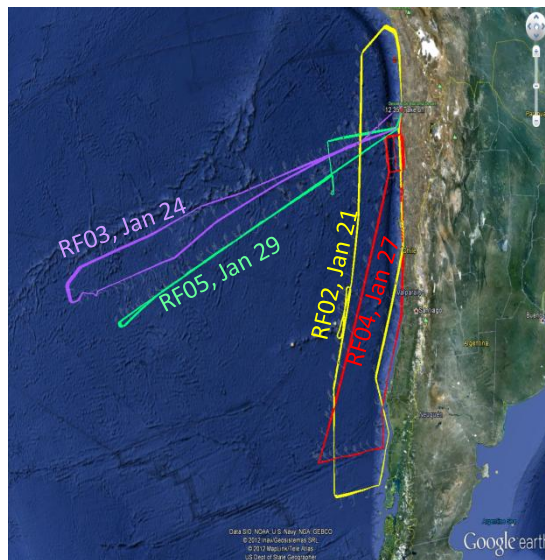
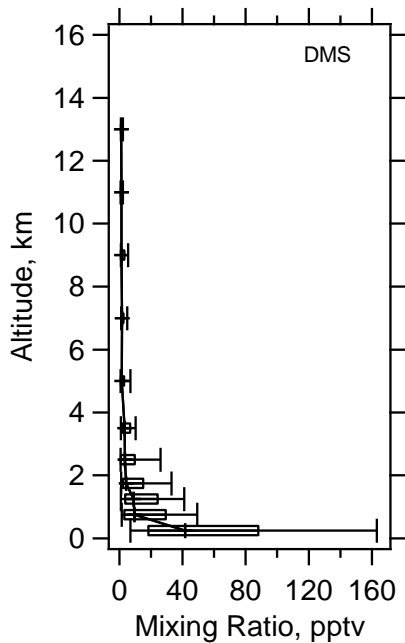


Fig. 1. Modeled annual mean DMS sea surface concentration. Units are nmol/l.

Kloster et al., 2005, Biogeosciences Discussions

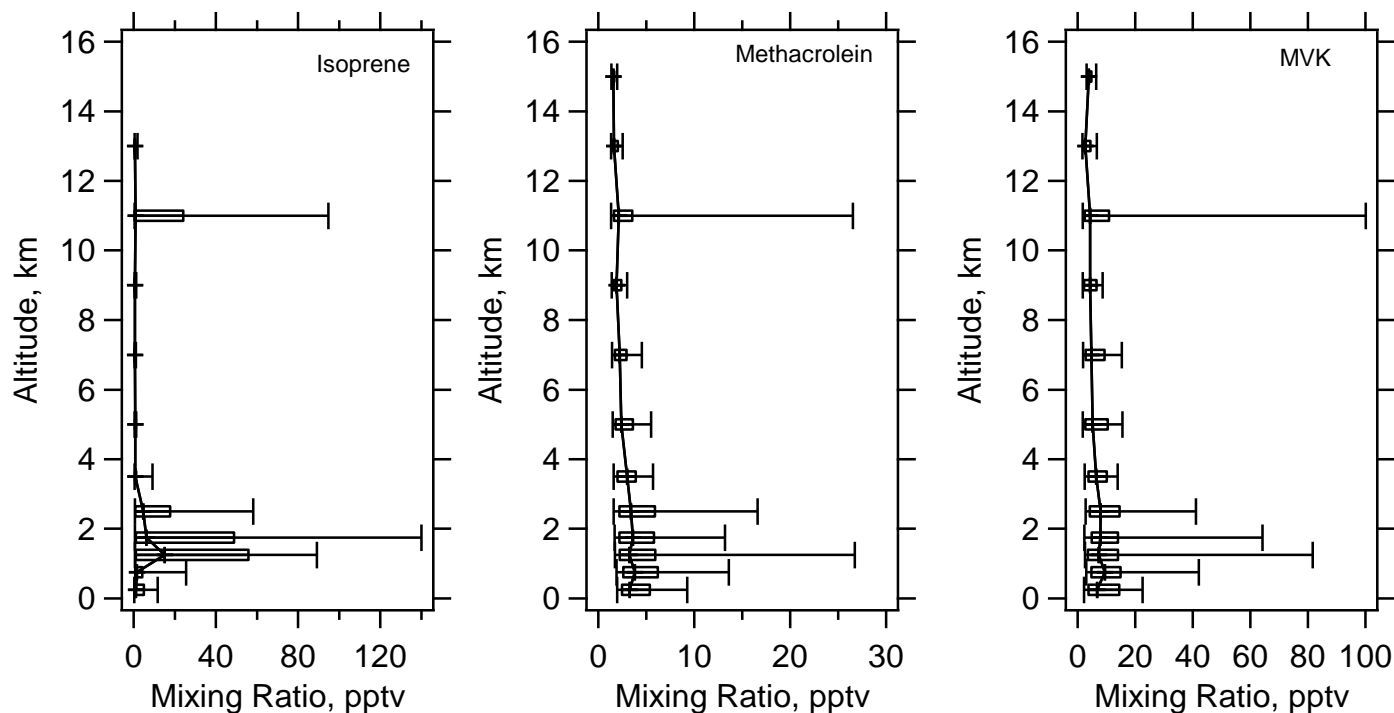


Biogenic compounds from ocean – impact on SOA etc.

very little isoprene observed in this study from oceans

Open question: 8 Tg/yr global source of organic marine aerosol (Spracklen et al., 2008)

Virtually no terpenes observed –very low MRs



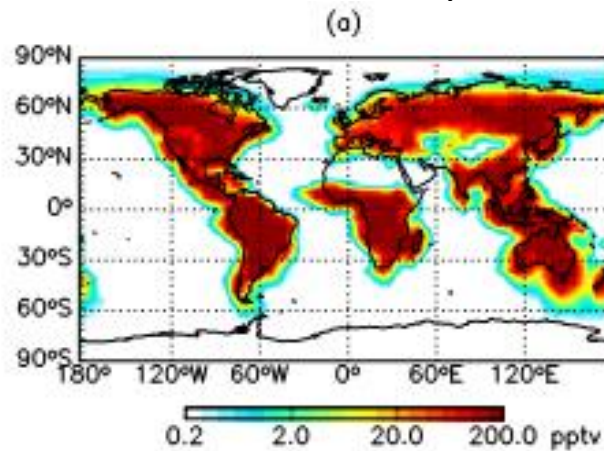
Avg. isoprene w/ DMS > 20 = 1.1 pptv

Avg. terpenes w/ DMS > 20 = 0.4 pptv

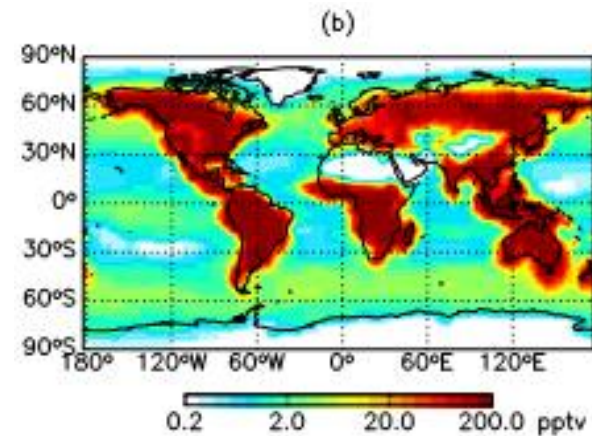
Isoprene consistent with bottom-up GEOS-Chem w/ isoprene source

Arnold et al., ACP, 2009

GEOS Chem – no isop source



Bottom-up GEOS Chem – w/ isop source



w/ top-down estimate

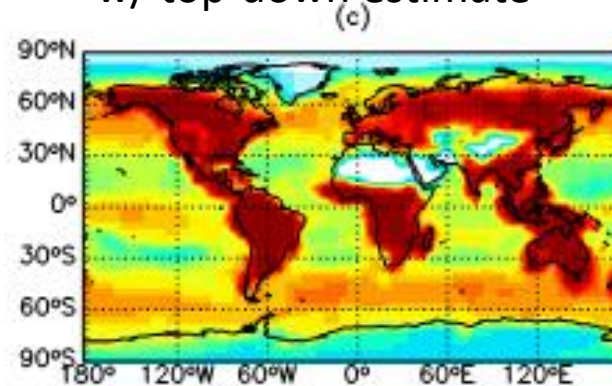


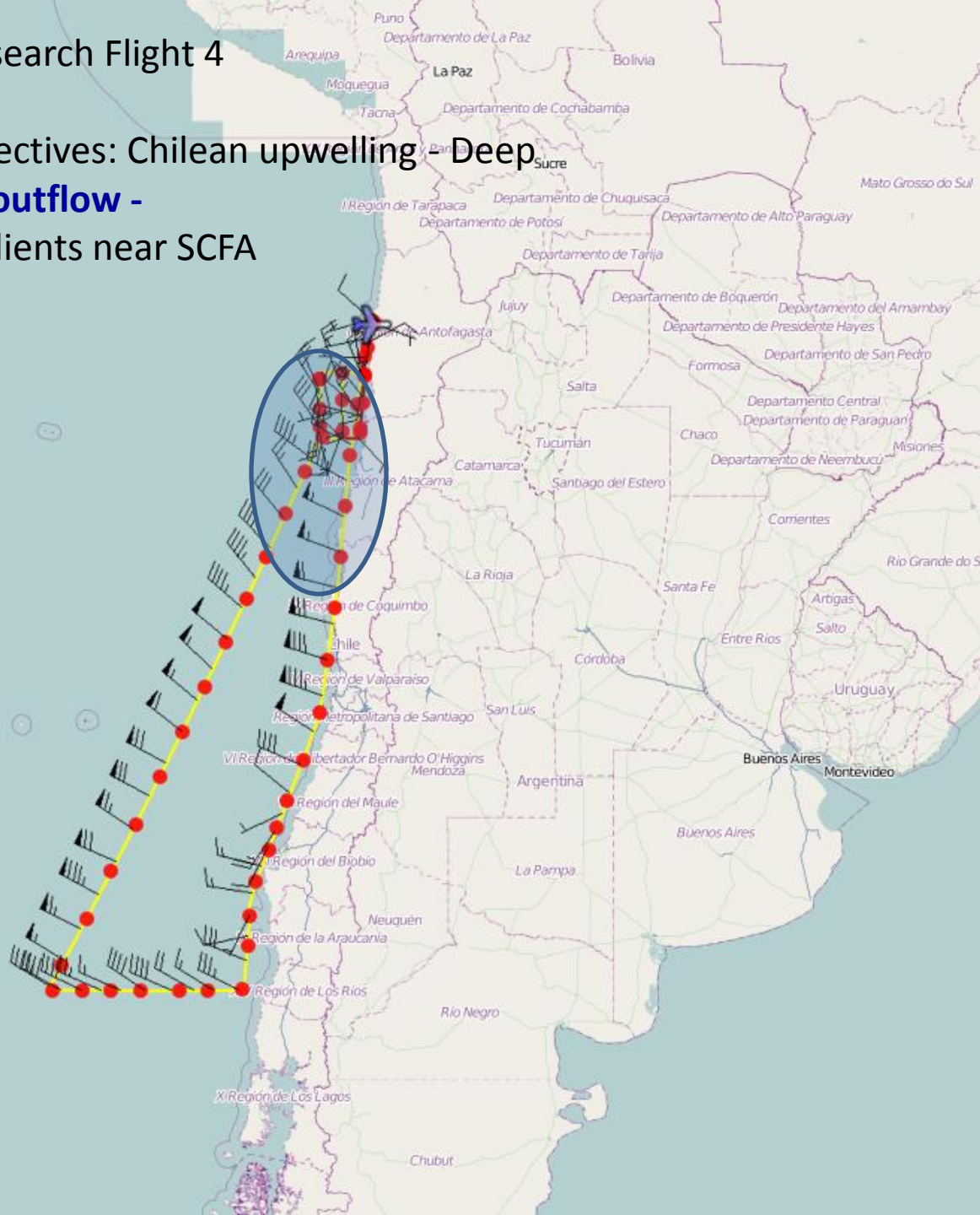
Fig. 2. (a–c) Annual mean surface atmospheric isoprene concentrations from the GEOS-CHEM model for year 2000 from simulations (a) without an oceanic isoprene source, (b) with the 0.31 Tg/yr “bottom-up” source estimate, and (c) with the 1.9 Tg/yr “top-down” source estimate. (d) Ratio of annual mean surface atmospheric isoprene concentrations from the 1.9 Tg/yr oceanic source to the no oceanic isoprene source simulations (c/a). Note colour scale is saturated at highest and lowest colour-bar values.

TORERO Research Flight 4

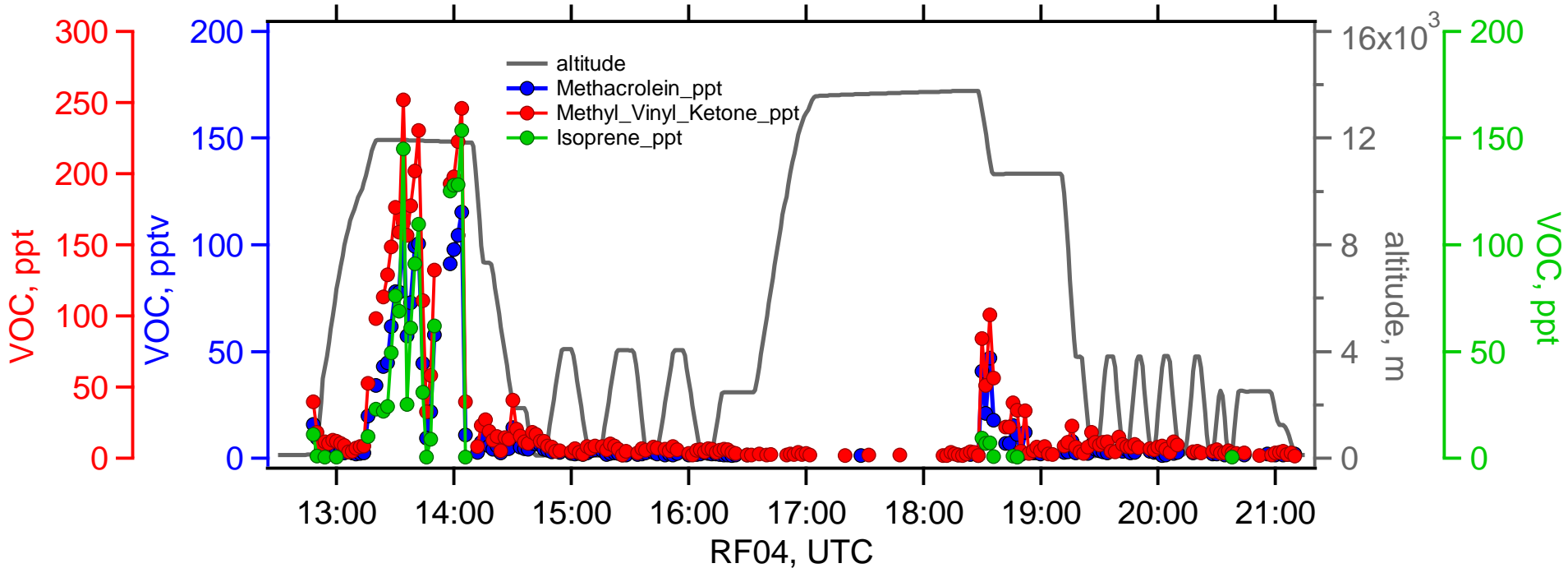
Science Objectives: Chilean upwelling - Deep

convective outflow -

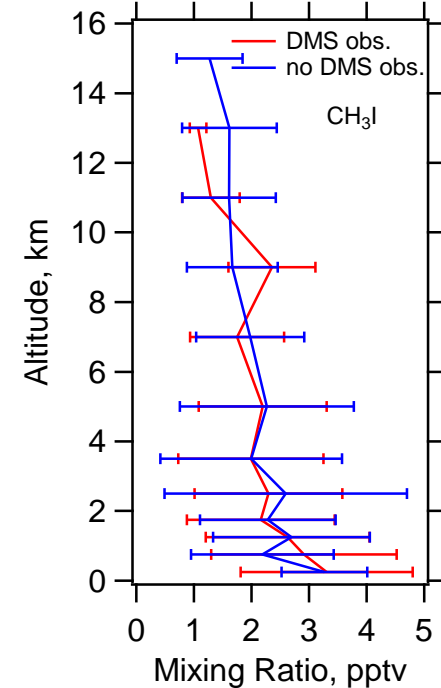
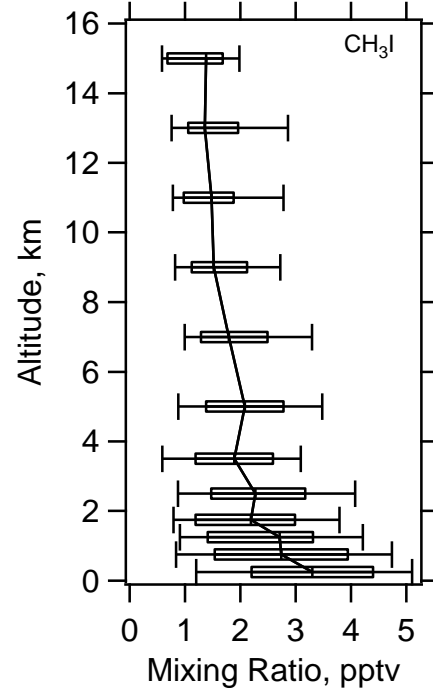
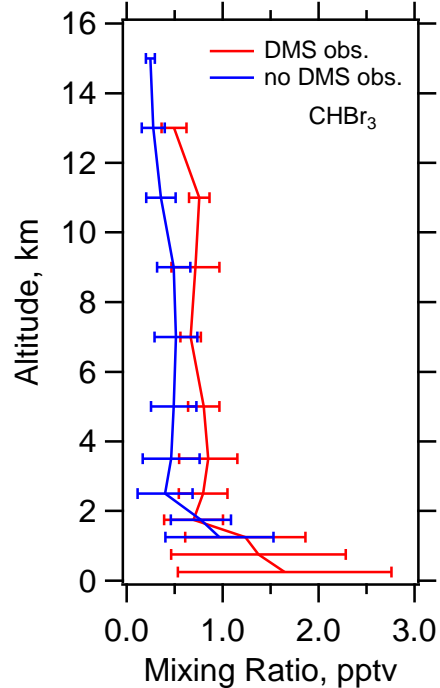
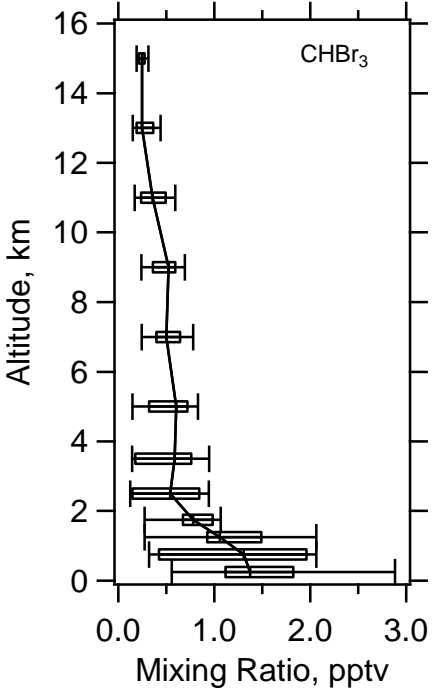
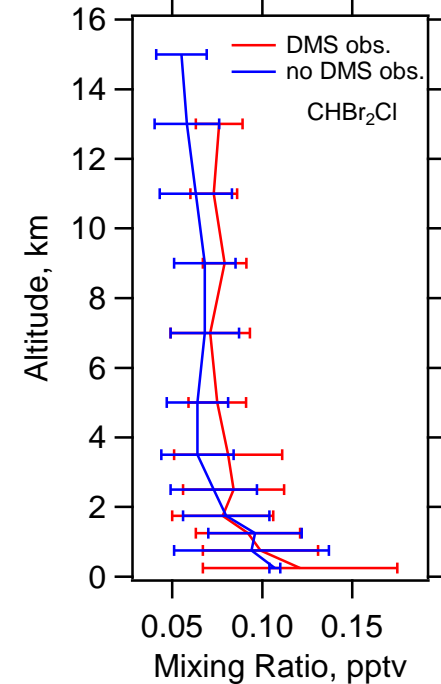
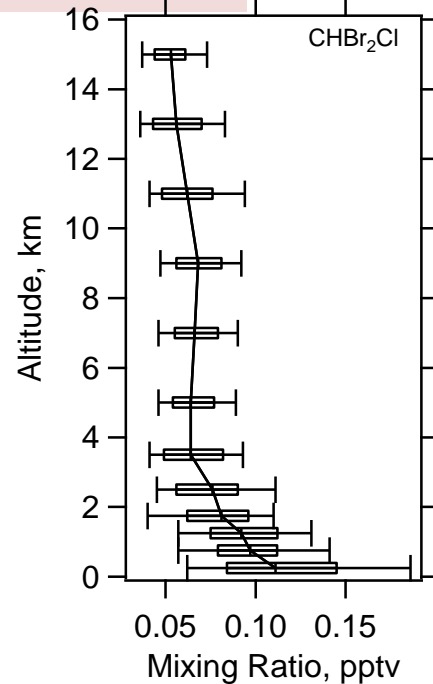
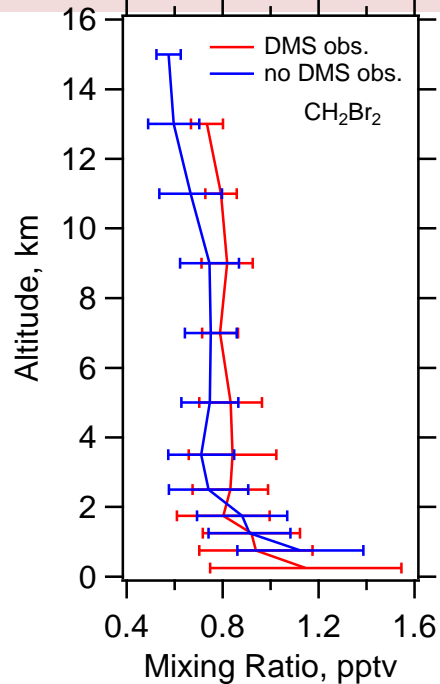
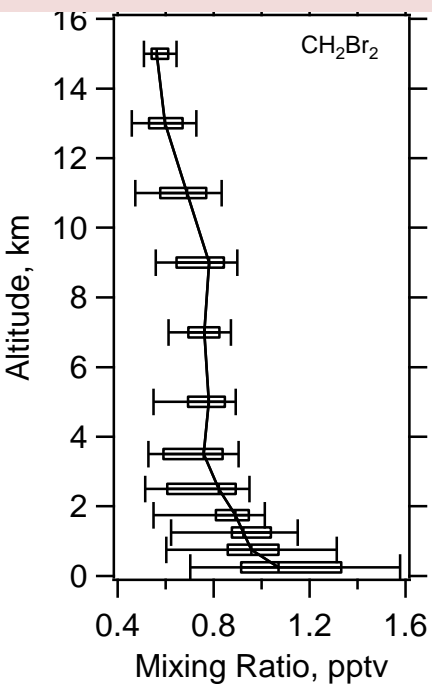
Coastal gradients near SCFA



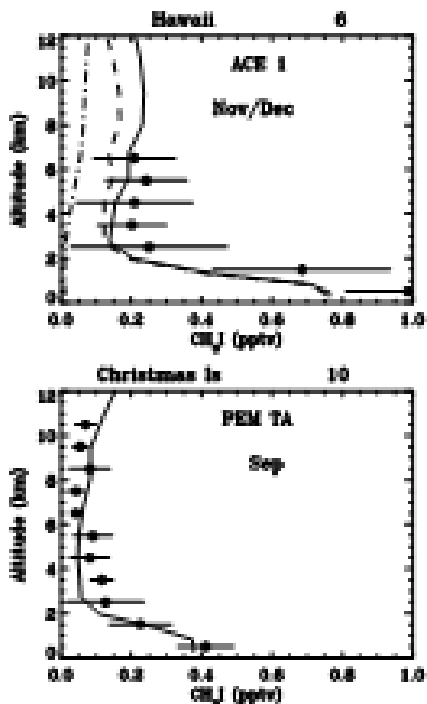
TOGA Measurements of convected species off coast of South America



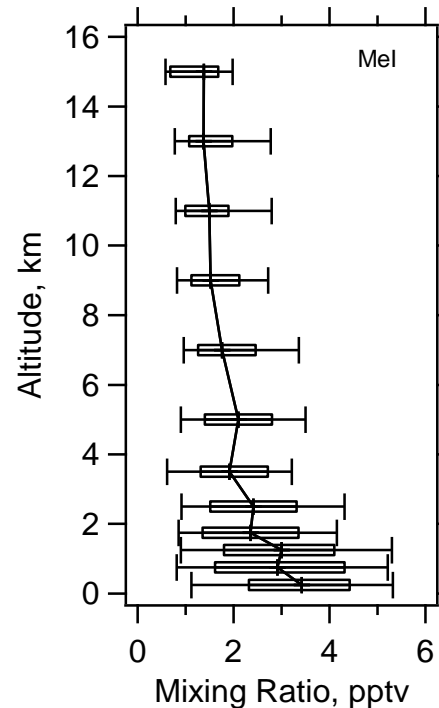
Impact of convection on brominated VSLs and methyl iodide



Some Previous Measurements



TOGA TORERO



Bell et al.,

Long-lived semi-soluble species

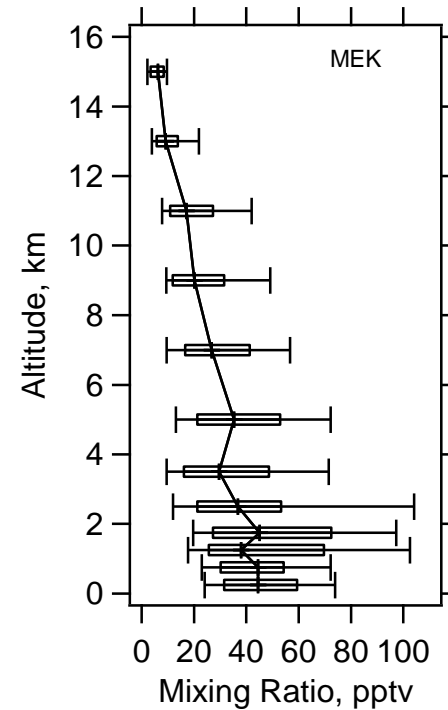
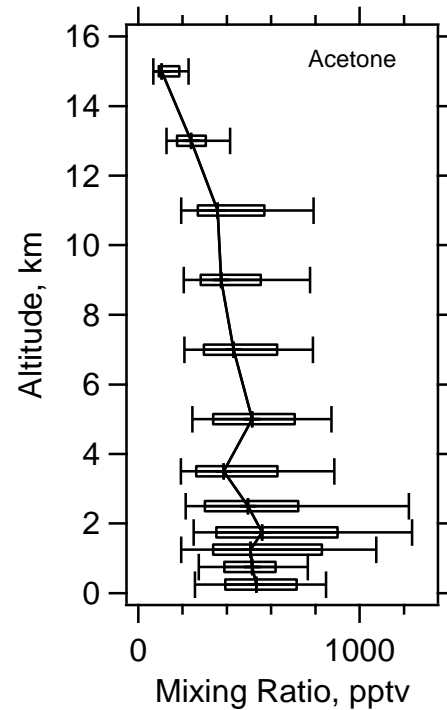
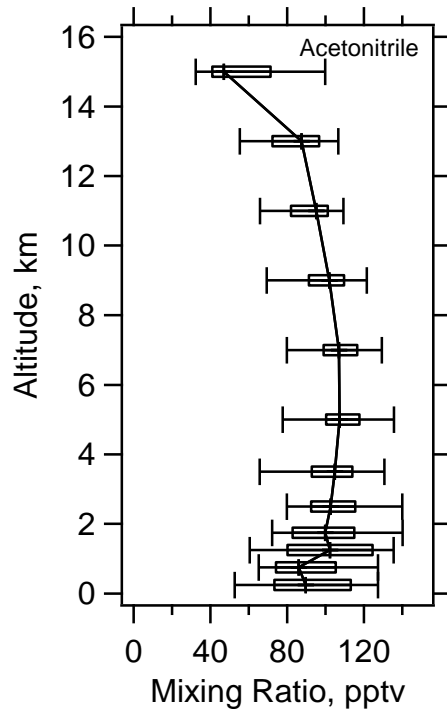
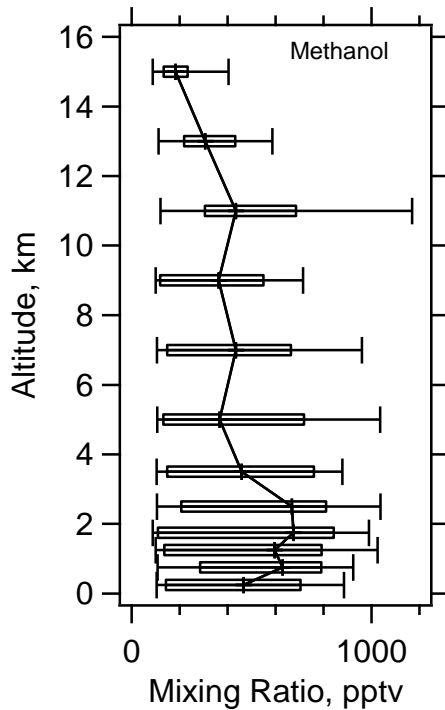
Methanol ~ 10 days, biogenic, BB, anthro, photochem

CH₃CN ~ months, BB (not much here)

Acetone ~ 1 month (14 days)

MEK ~ 10 days

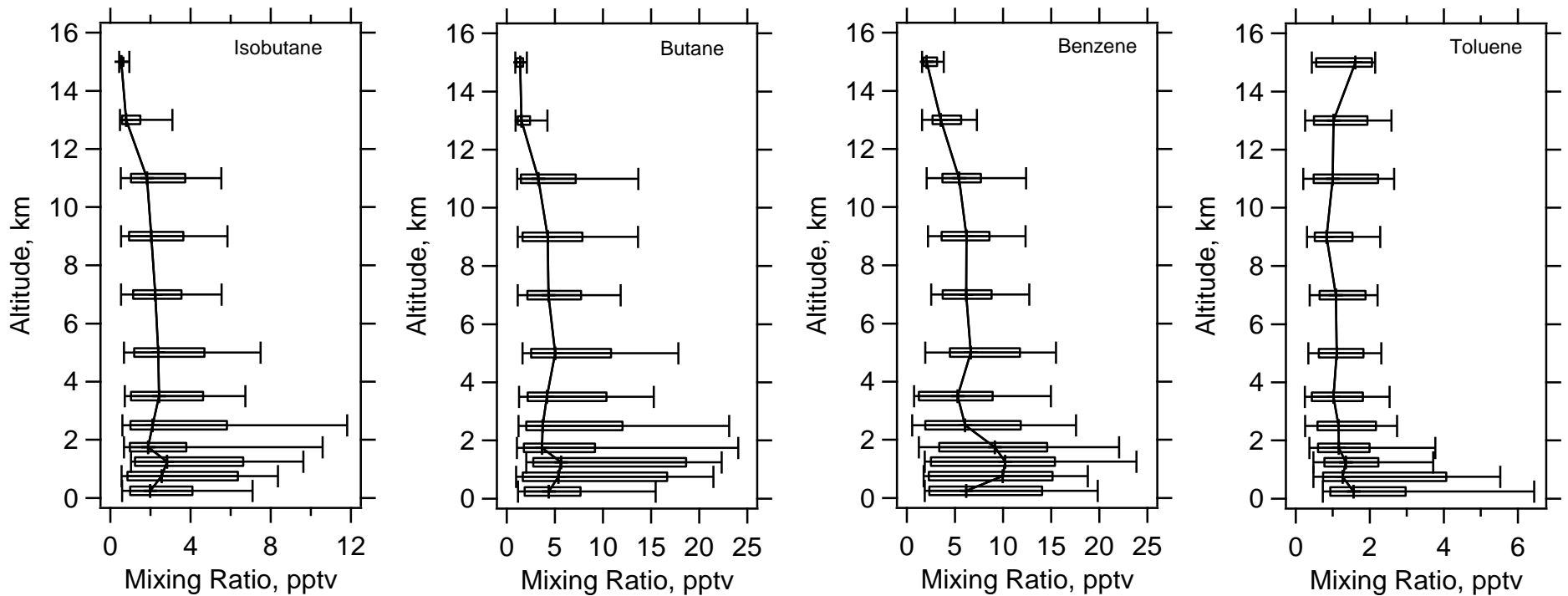
TORERO TOGA Measurements



Anthropogenic (BB) compounds – Tracers from over ocean-only data

Note low MRs of species – very little overall anthro (BB) influence

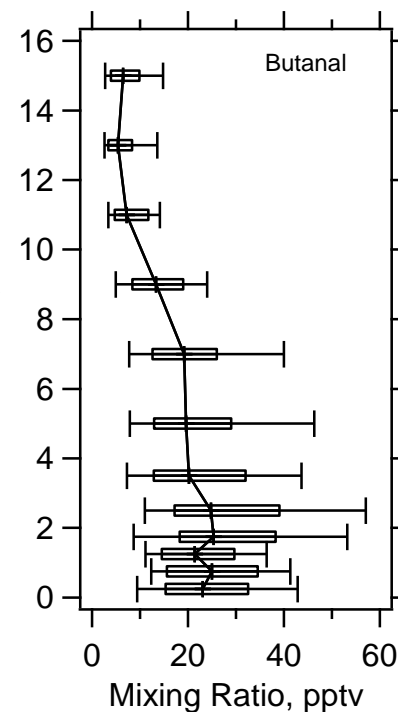
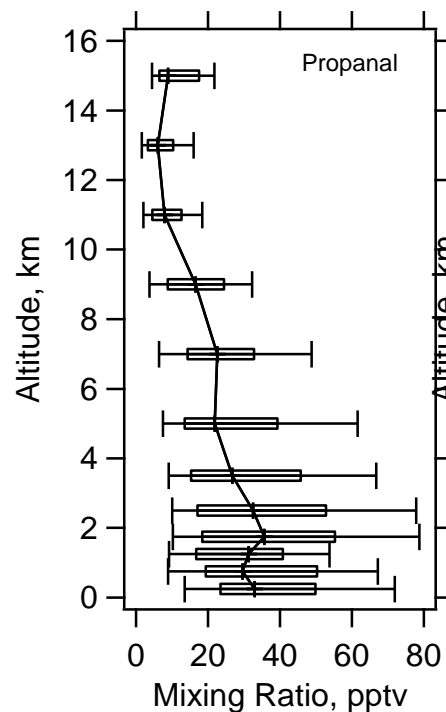
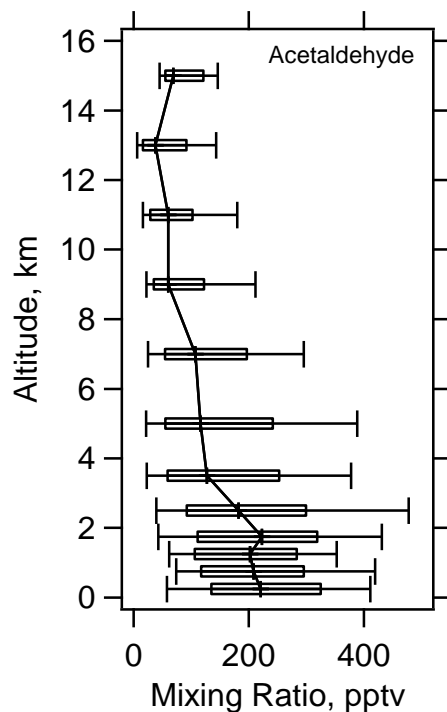
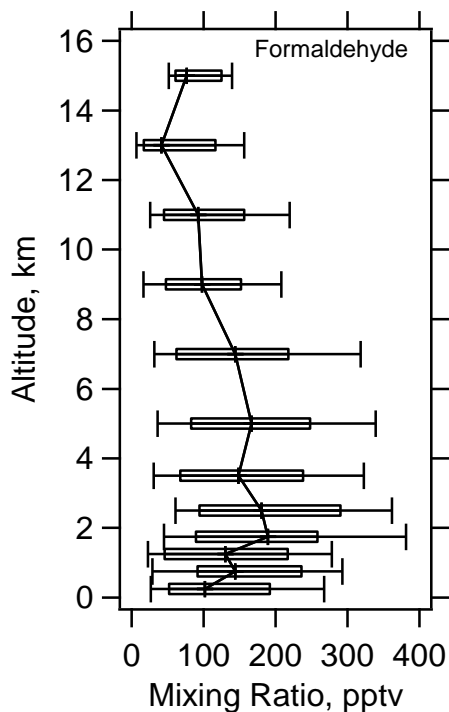
Butanes $\tau \sim$ week, Benzene \sim 1 month, toluene \sim days



Short-lived OVOCs

Aldehydes – formaldehyde – many sources incl. methane, methanol, MeOOH, CH₃CHO, etc
Acetone...

Others – all have short lifetimes



Specific flights

RF03, Jan 24

RF05, Jan 29

RF02, Jan 21

RF04, Jan 27

Desierto De Atacama Desert
12:35 Take off

Chile

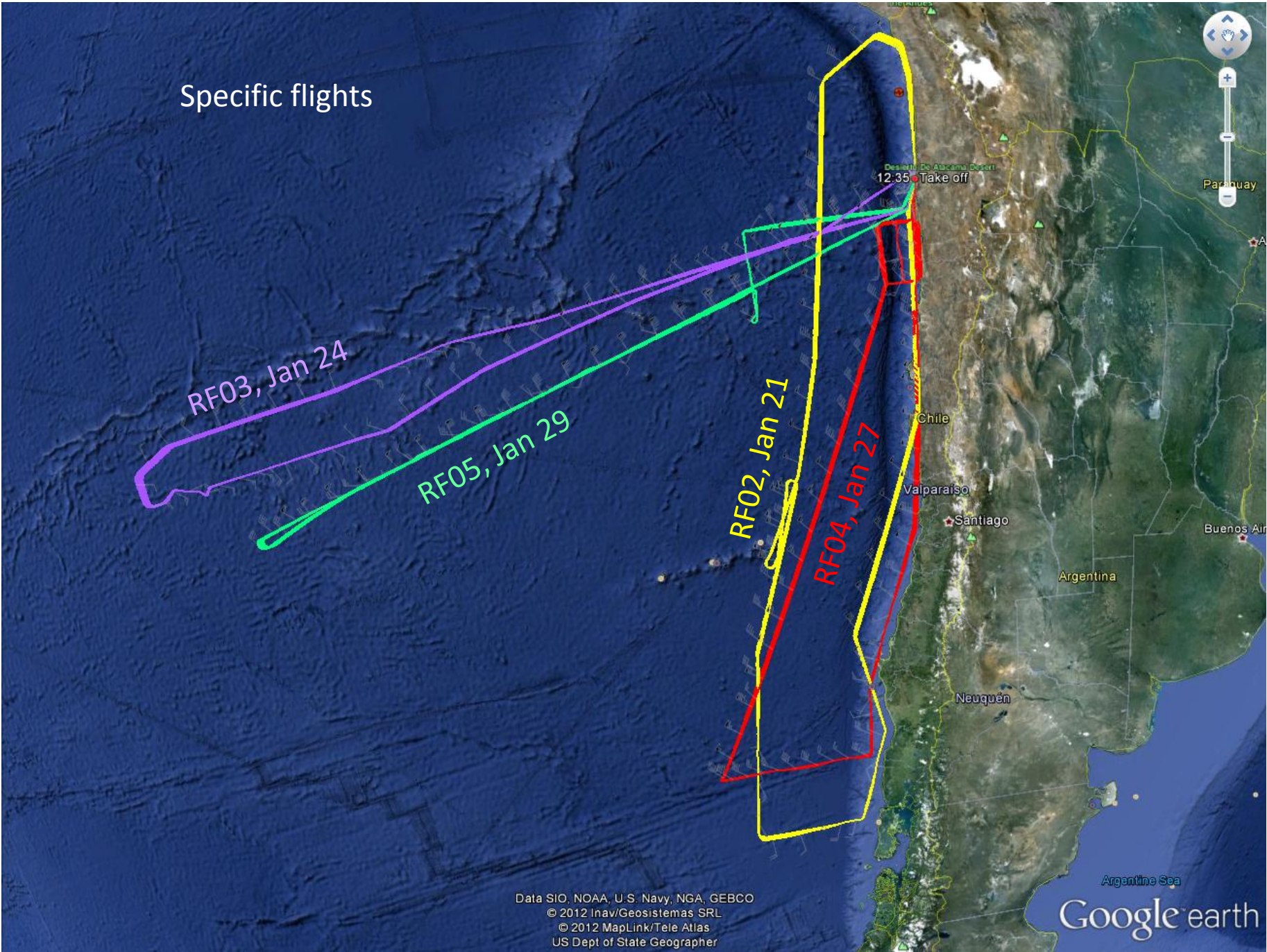
Vaiparaiso

Santiago

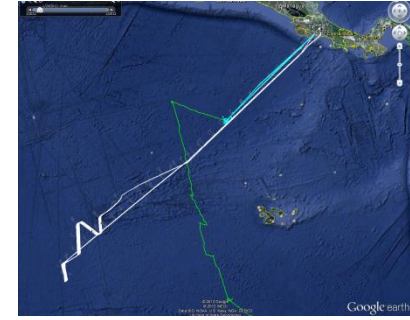
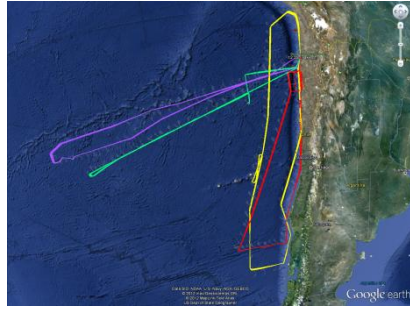
Argentina

Neuquén

Argentine Sea



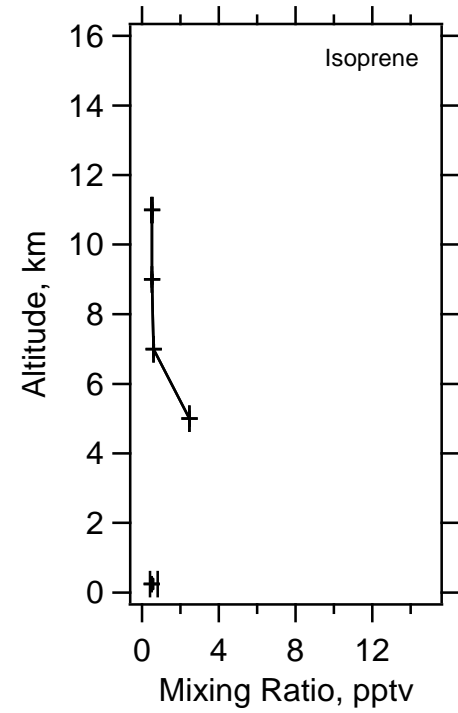
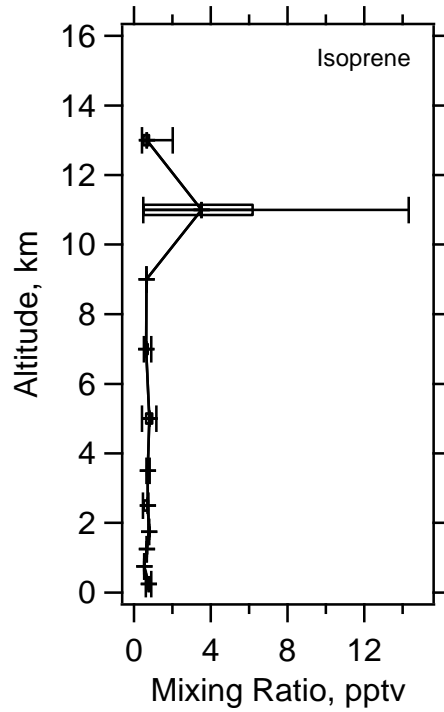
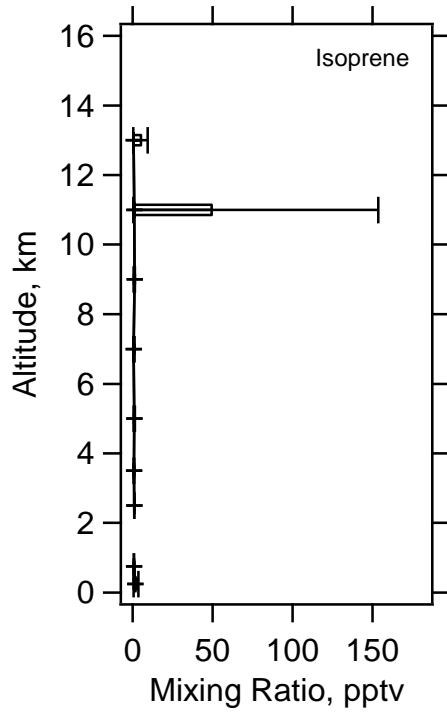
Convection - Isoprene



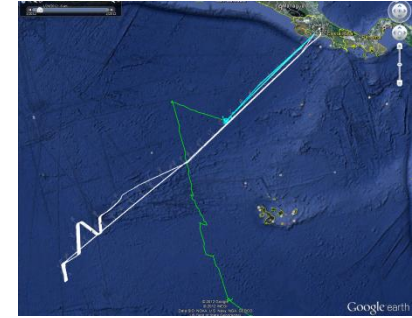
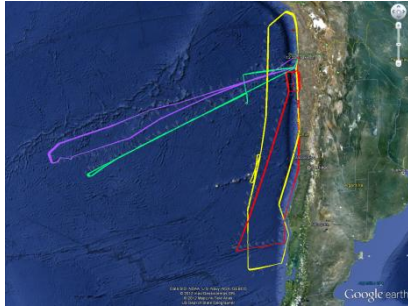
RF02 and RF04

RF03 and RF05

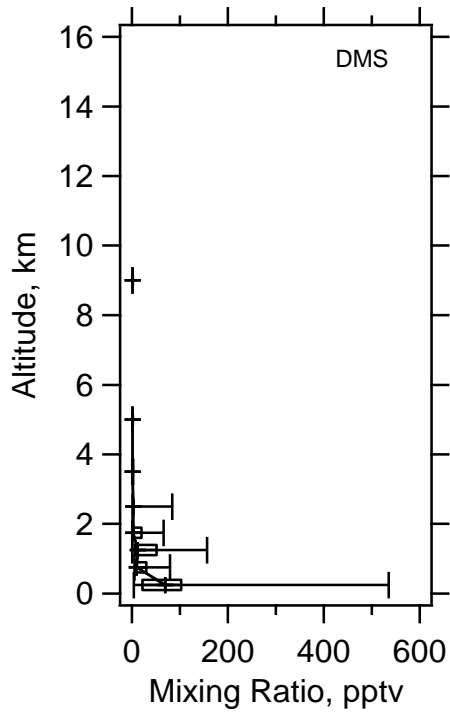
RF08 and RF17



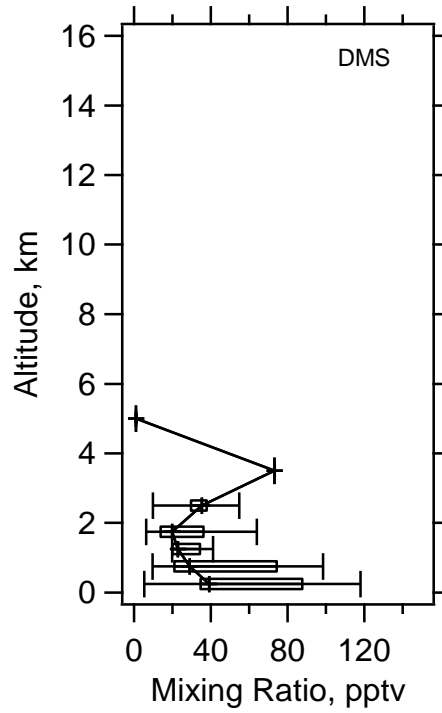
DMS



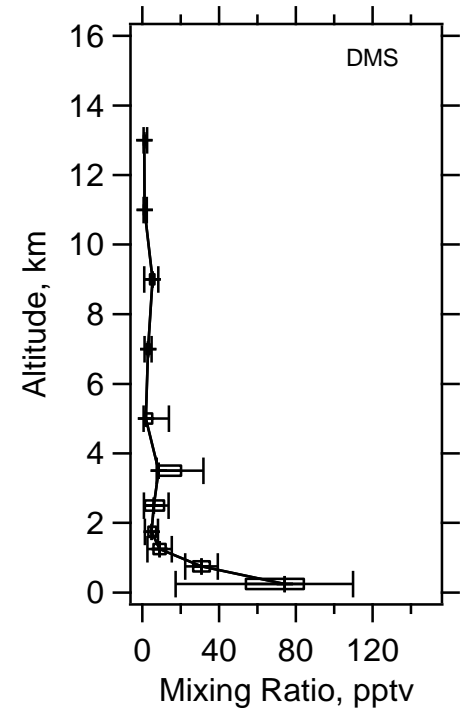
RF02 and RF04



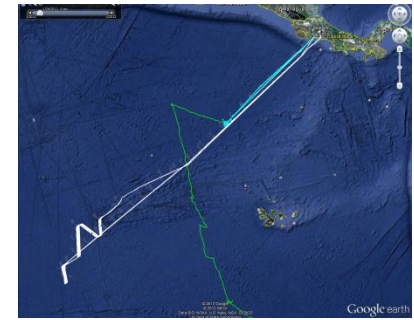
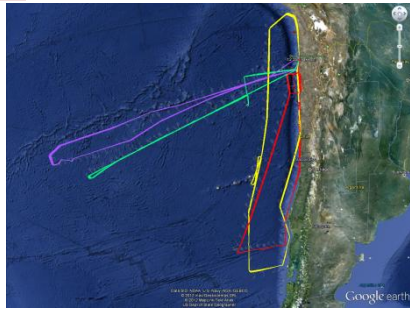
RF03 and RF05



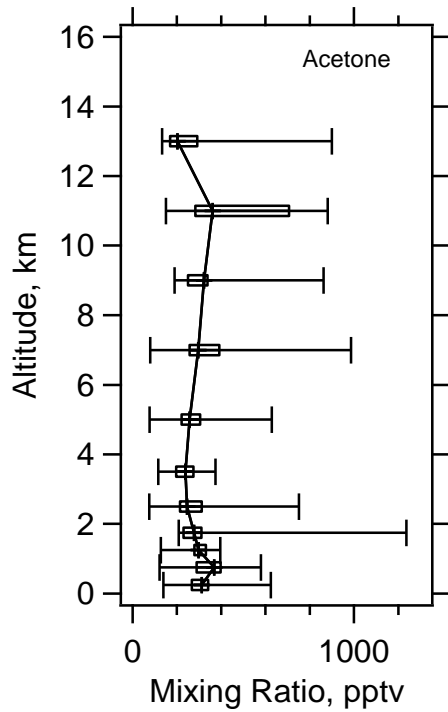
RF08 and RF17



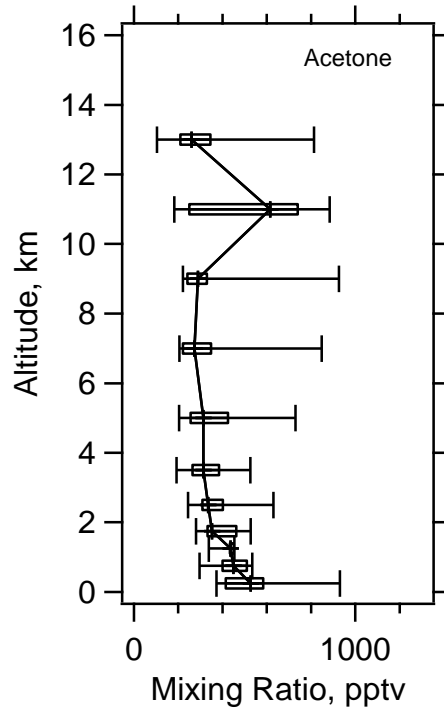
Acetone



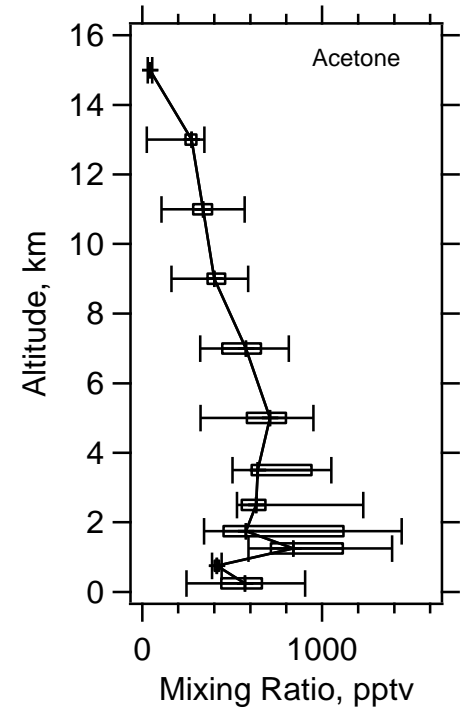
RF02 and RF04



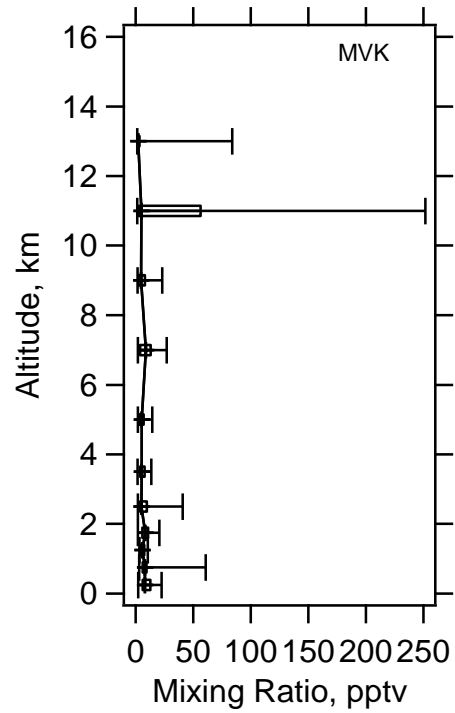
RF03 and RF05



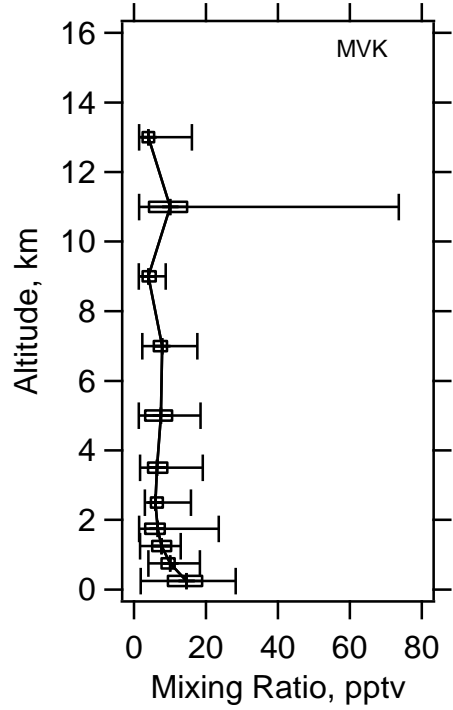
RF08 and RF17



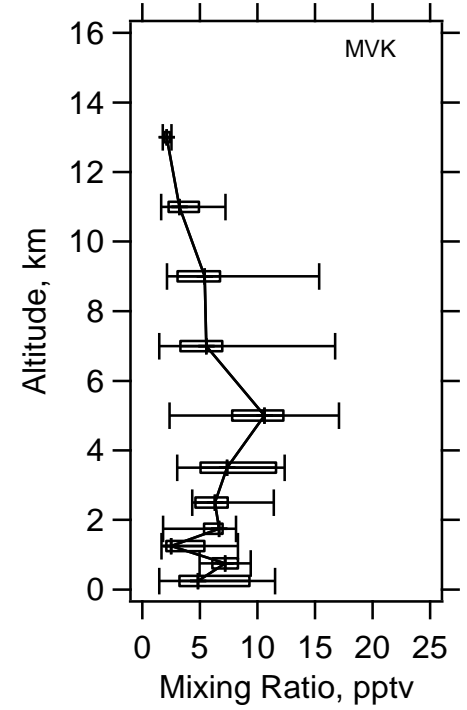
RF02 and RF04



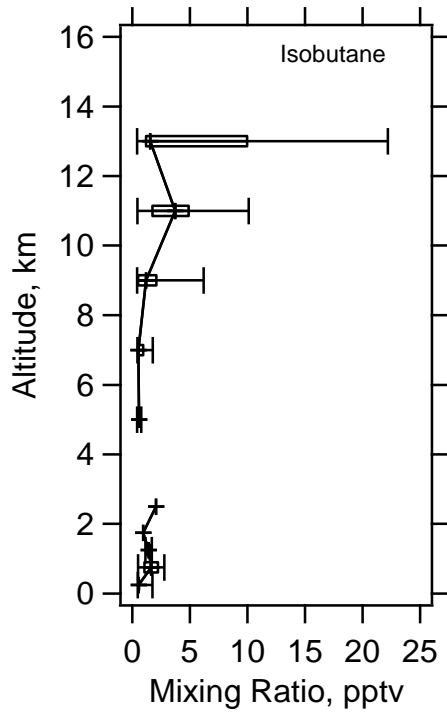
RF03 and RF05



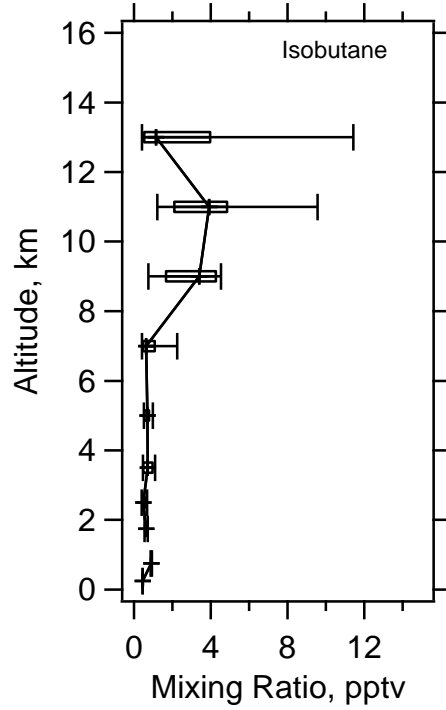
RF08 and RF17



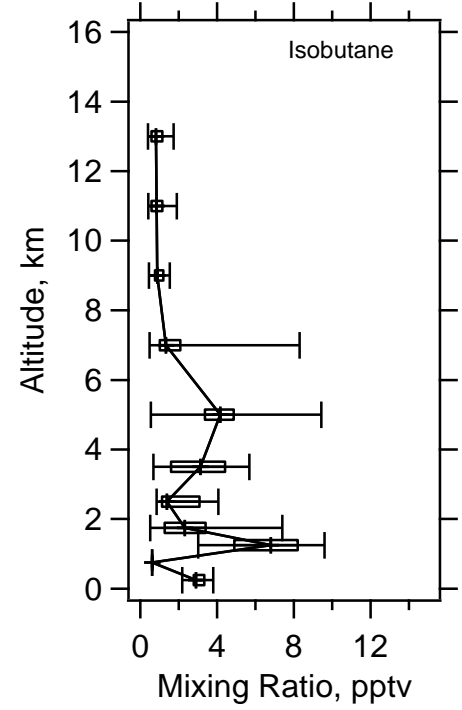
RF02 and RF04



RF03 and RF05

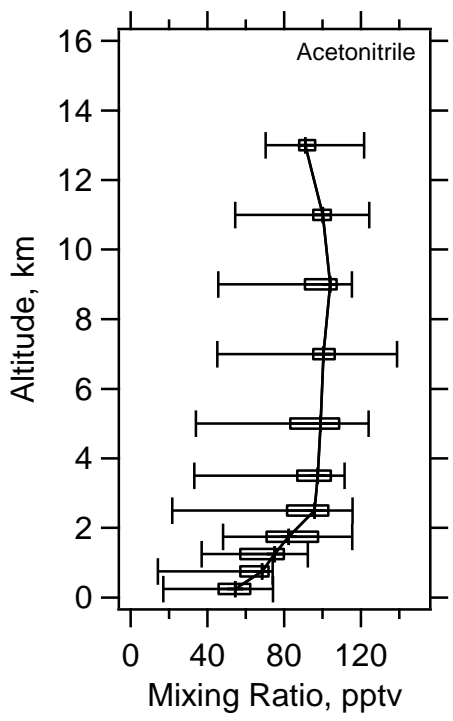


RF08 and RF17

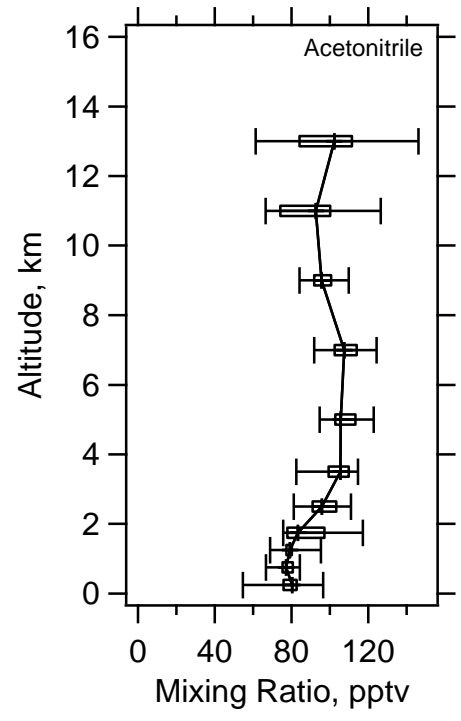


Biomass burning

RF02 and RF04



RF03 and RF05



RF08 and RF17

