

**Vertical Profiles of Reactive Gaseous
Mercury during TORERO
*plus a few other things about Hg in
the atmosphere***

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Atmospheric Hg

- Two stable oxidation states
- Elemental: Hg^0 (>95%): GEM
 - unreactive, insoluble, low deposition rate
 - long atmospheric lifetime: 6 month – 1 year
 - global background $[\text{Hg}^0] = 1 - 2 \text{ ng m}^{-3}$
 - Northern Hemisphere: 1.5 ng m^{-3}
 - Southern hemisphere: 1 ng m^{-3}
 - Atmospheric $[\text{Hg}^0]$ has risen by a factor of 3 in the last 100 years

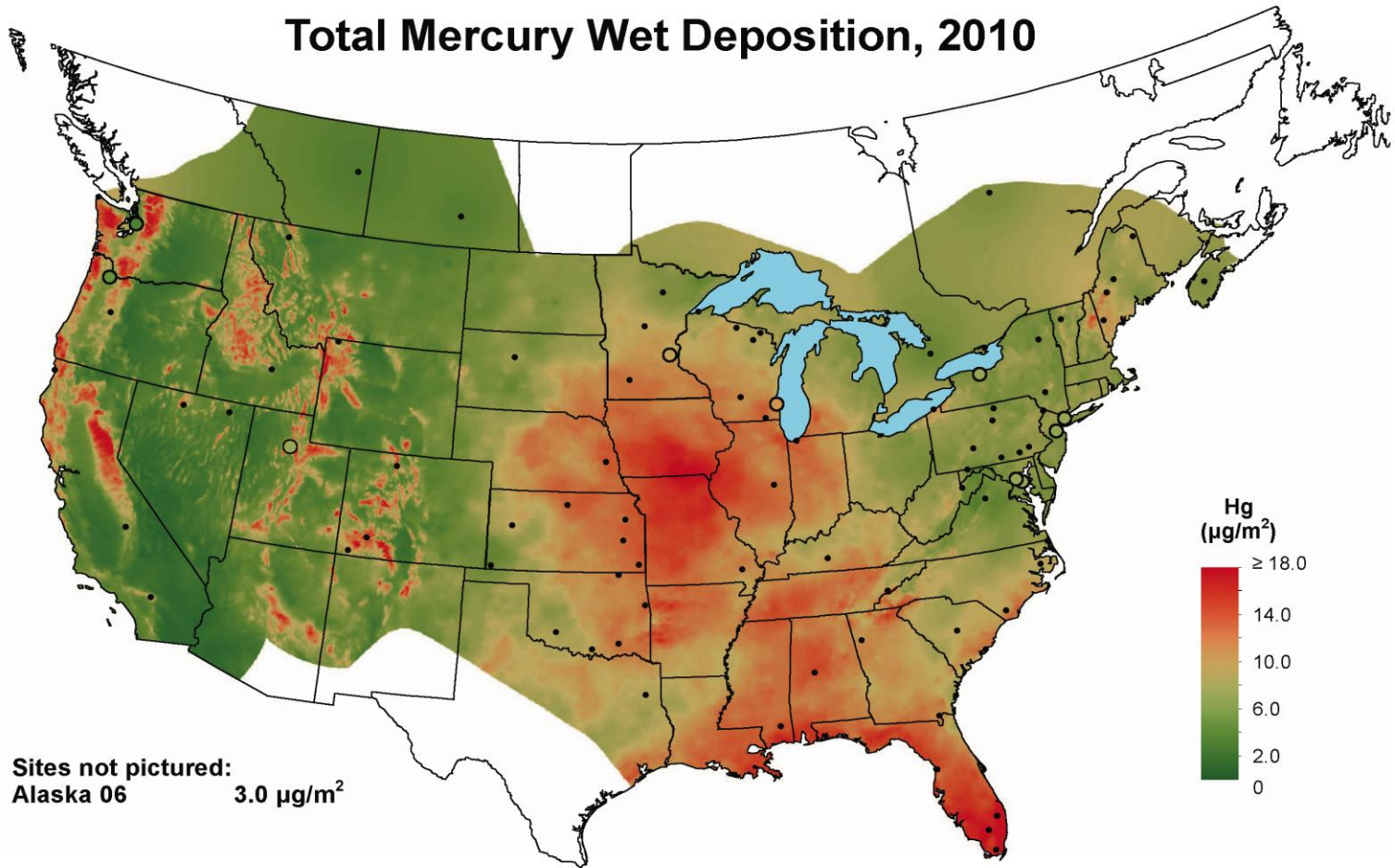
Reactive Gaseous Mercury (RGM)

- Operationally defined term: mercury species that collect on a KCl denuder
- Major species ? : HgCl_2 , HgBr_2
- Water soluble
- Easily washed out of the atmosphere resulting in increased local deposition
- Concentrations are normally low ($3 \pm 11 \text{ pg m}^{-3}$ in MBL : Sorensen et al. 2010)
- BUT higher values observed in subsiding air (Mt. Bachelor : Reno) and AMDE's

ATMOSPHERIC Hg CHEMISTRY

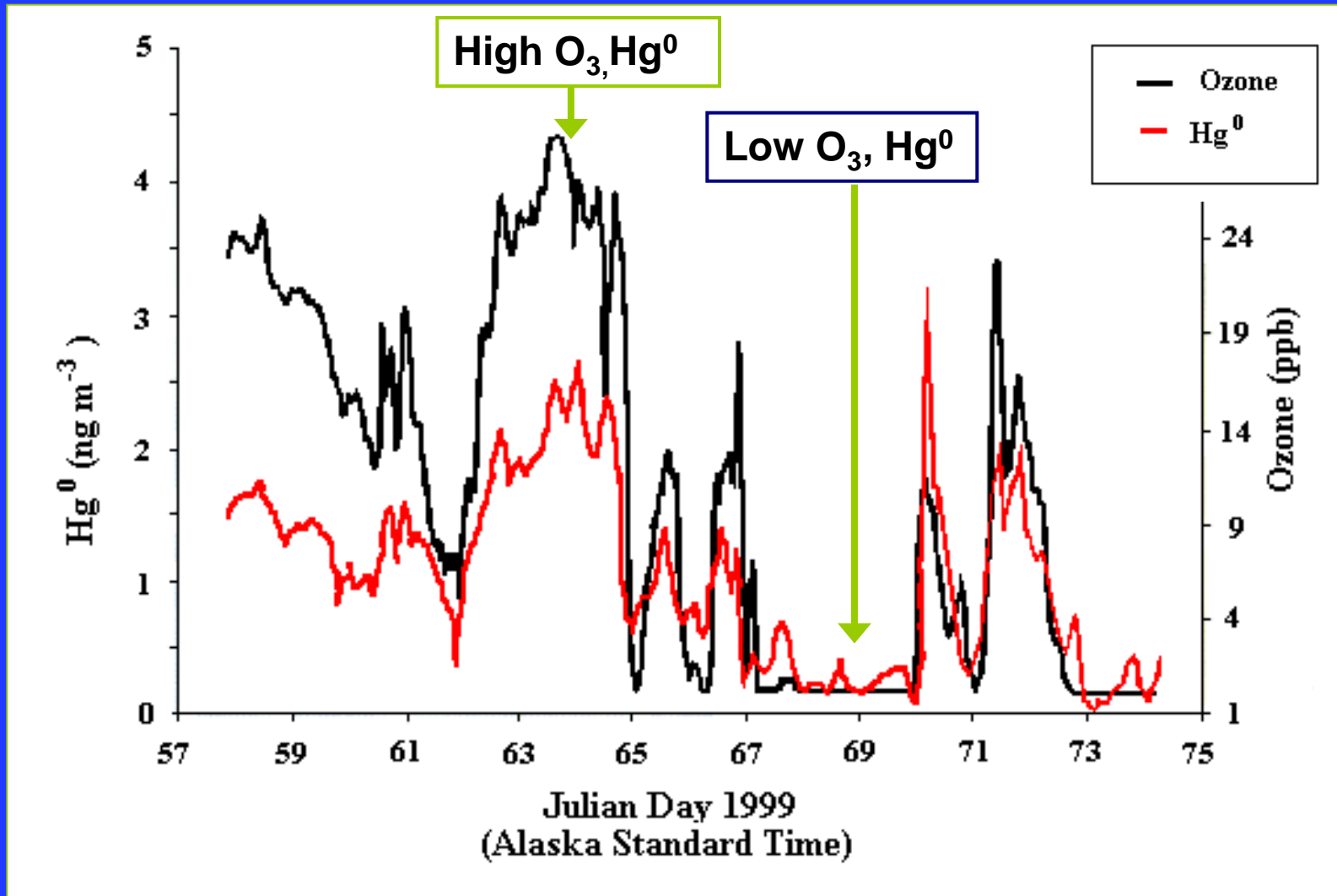
- Not well understood!!
- Deposition: Peculiar pattern of wet deposition with very high values in the SE USA
- Atmospheric Mercury Depletion Events (AMDE's). Sporadic examples of very fast oxidation chemistry at poles and elsewhere

Total Mercury Wet Deposition, 2010



National Atmospheric Deposition Program/Mercury Deposition Network
<http://nadp.isws.illinois.edu>

Barrow Arctic Mercury Study



Lindberg et al (2002)

Arctic Hg depletion

- ◆ No Hg depletion before sunrise

Strong correlation between O_3 and Hg depletion

**Reactive gaseous mercury (RGM) is rising
during depletion of Hg(0)**

Depletion event is triggered by sunrise

Halogen (Br) chemistry is likely candidate

Non-Arctic Chemistry

Peleg et al. [3] diurnal cycles of mercury, ozone, and BrO near the Dead Sea, Israel.

Mona Loa, Mount Bachelor, Reno evidence for elevated RGM in the free and upper troposphere.

MBL observations of diurnal variation in production of RGM

What is the mechanism of Atmospheric Hg⁰ Oxidation?

- HOx (OH; O₃) vs Halogen initiated oxidation
- Global oxidation of background pool vs local anthropogenic emissions?

Halogen recombination



$\text{X} = \text{Br}, \text{Cl}$

Three body recombination,

P dependent

Measurable rate coefficient

HgX stable molecule, monitor
spectroscopically, study chemistry

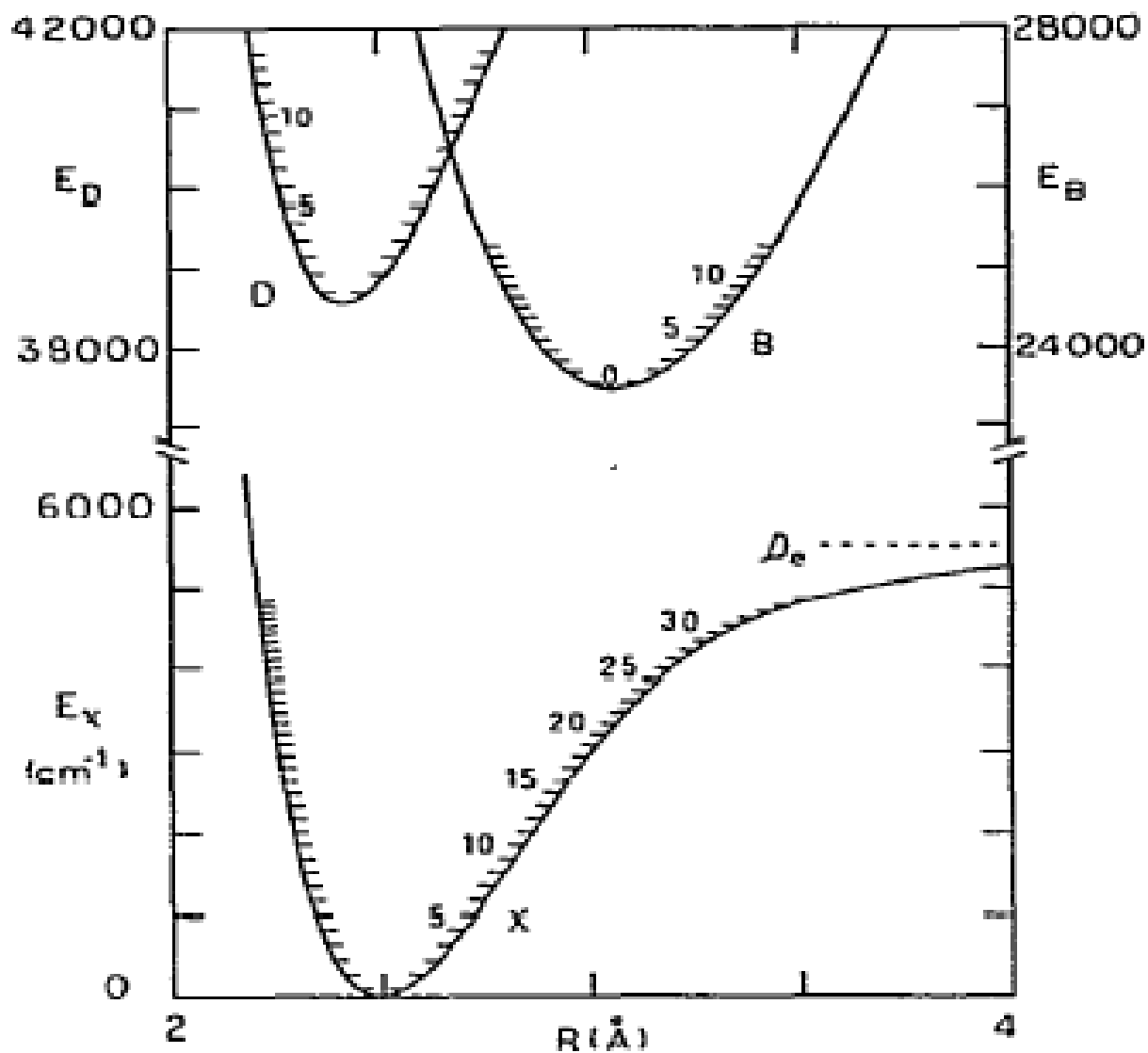
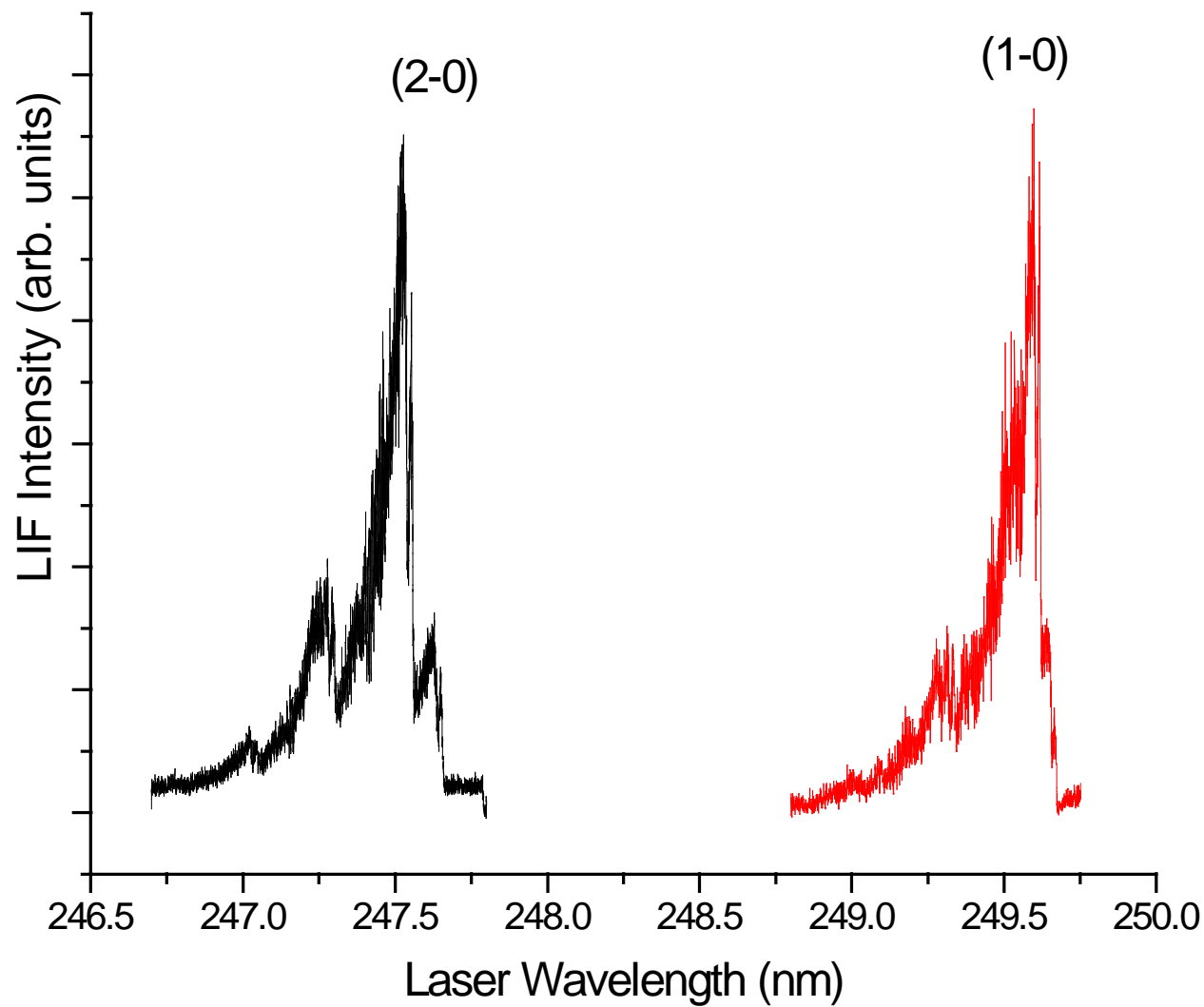
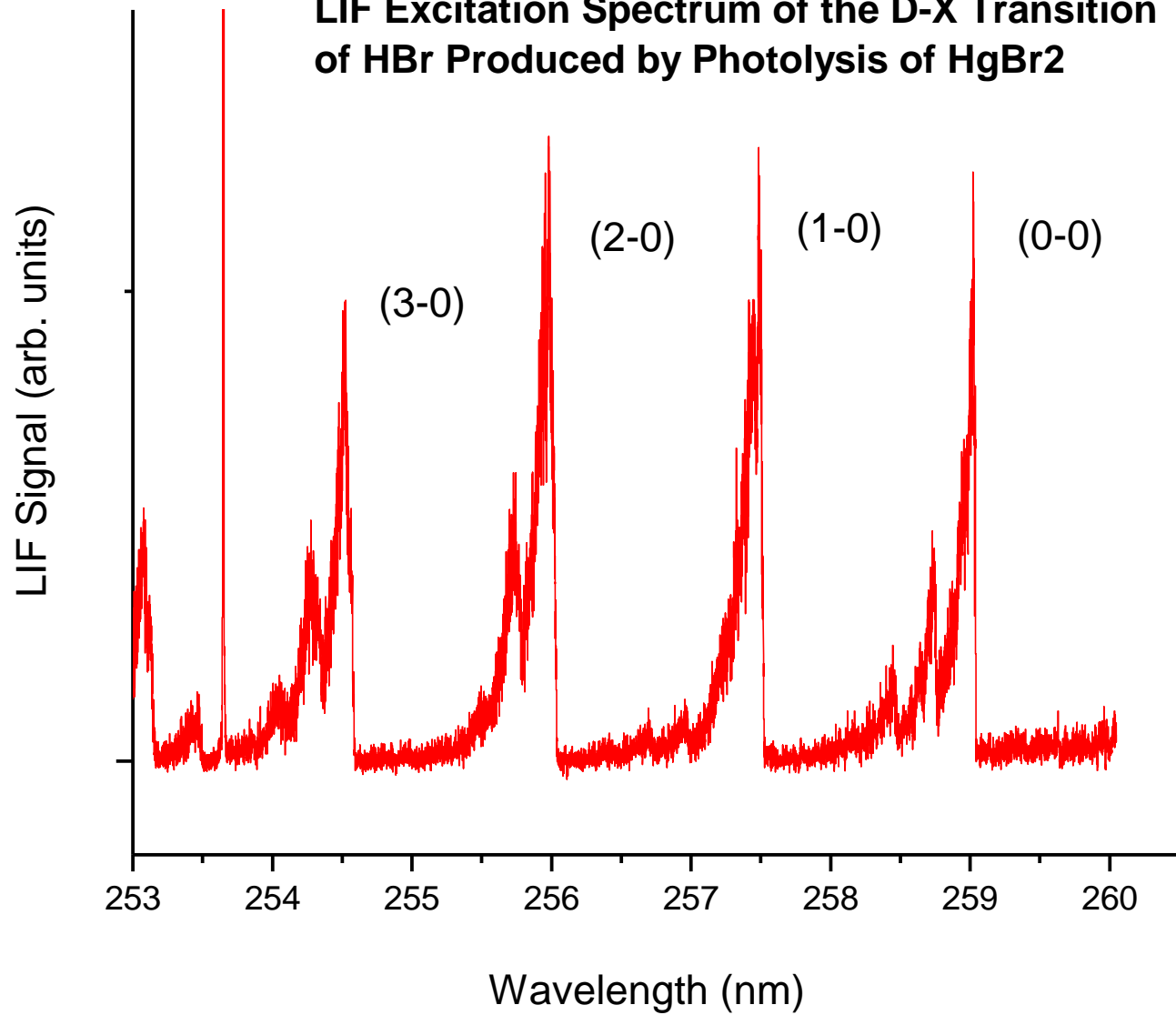


Fig. 1. Potential diagram for HgBr, showing levels of the X, B, and D states spanned by the present analysis. Note the different energy scales for all three states.

LIF Spectra of the $D^2\Pi$ - $X^2\Sigma$ (2-0) and (1-0) Bands of HgCl after Photolysis of HgCl_2



LIF Excitation Spectrum of the D-X Transition of HBr Produced by Photolysis of HgBr₂





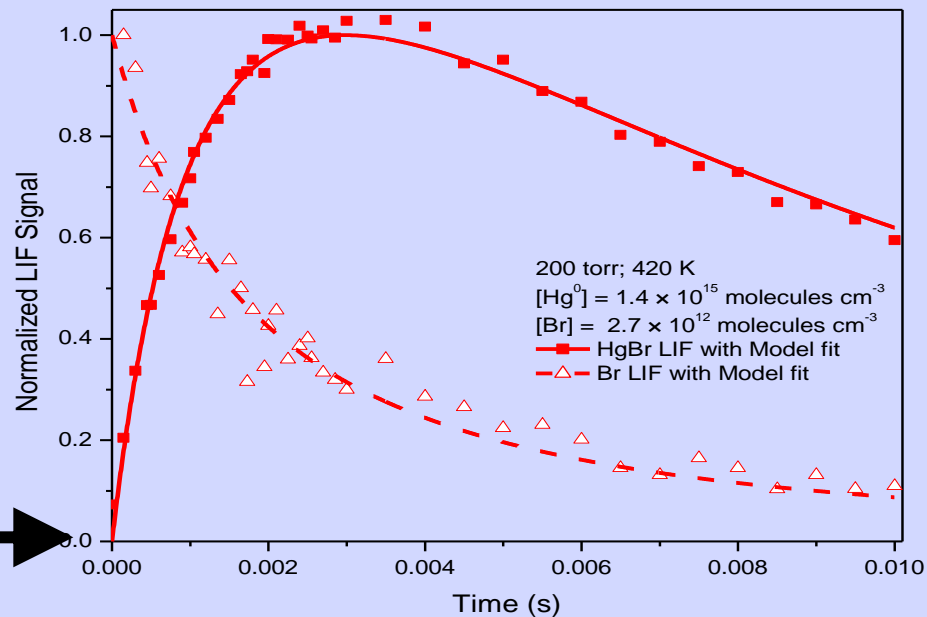
Important in AMDE' s?

Global Oxidant for Hg(0)?

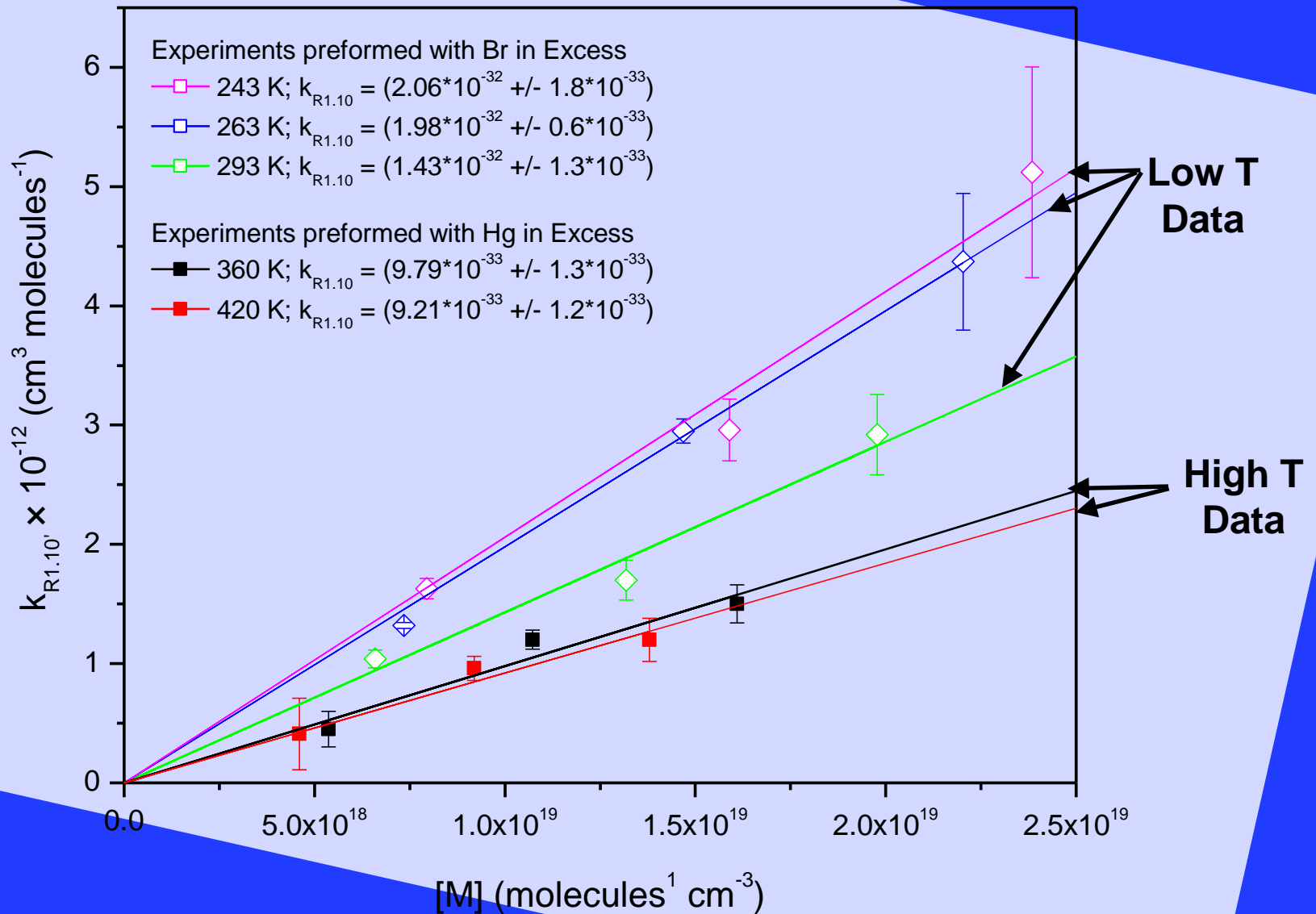
**Perhaps the best defined Hg(0)
rate coefficient, uncertainty less
than a factor of 10 !**

Numerical Integration

$$\frac{d[Br]}{dt} = -k_1 [Br][CF_3] - k_2 [Br] - k_3 [Br][Hg] - k_5 [HgBr][Br] + k_4 [HgBr]$$



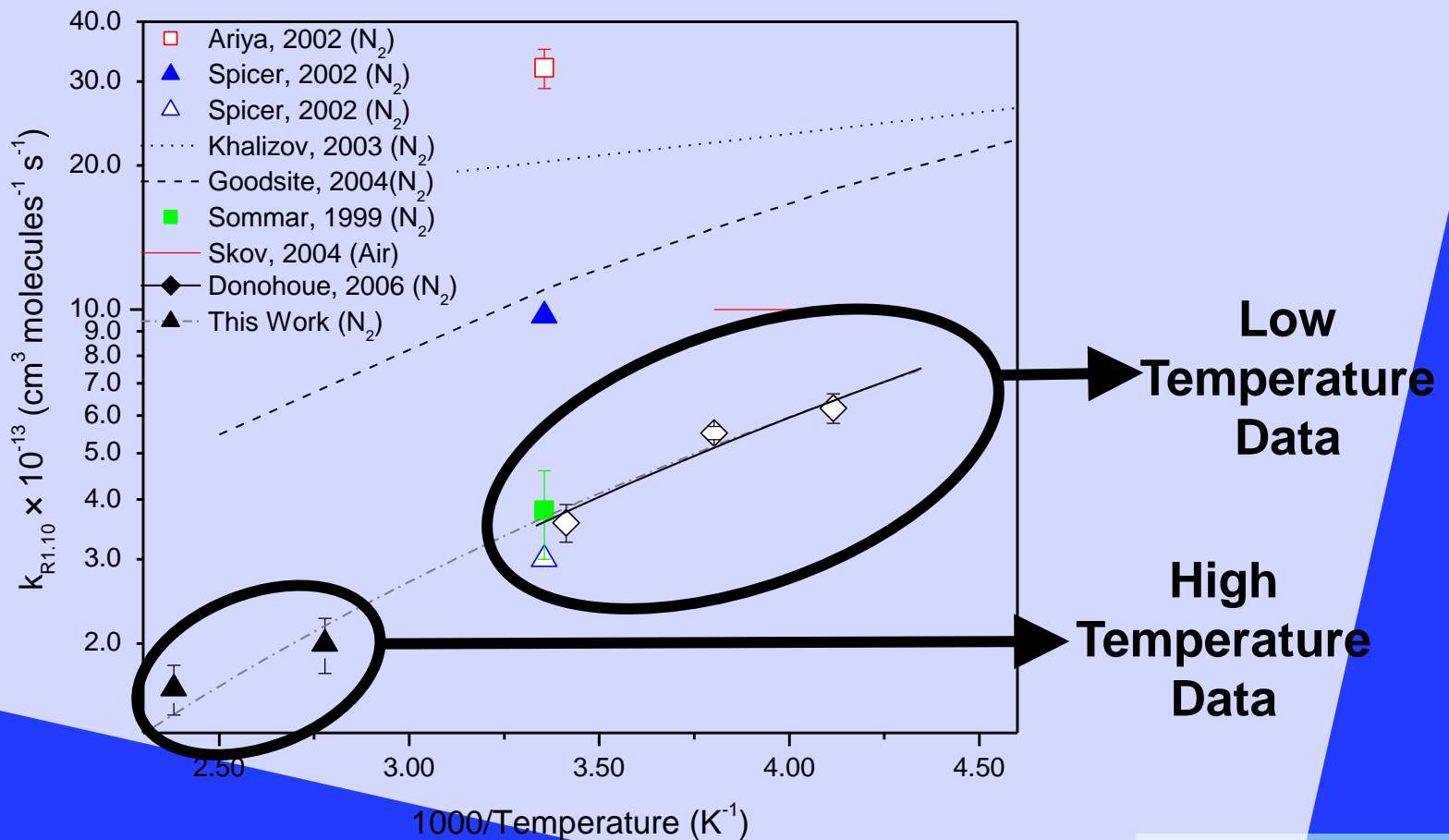
$$\frac{d[HgBr]}{dt} = k_3 [Br][Hg] - k_5 [HgBr][Br] - k_4 [HgBr]$$

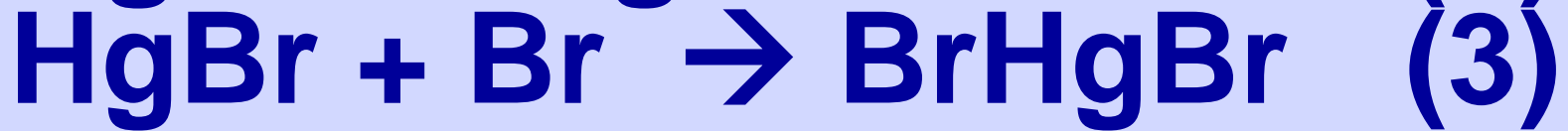
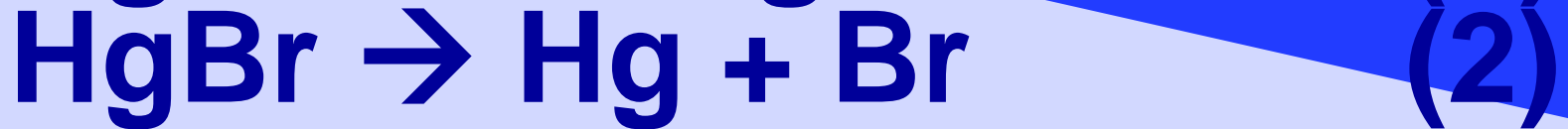




3rd Order Plot

$$k_{3,N_2} (243 - 420K) = (1.49 \pm 0.34) \times 10^{-32} \times \left(\frac{T}{298} \right)^{-(1.76 \pm 0.5)}$$





τ due to reaction with Br =

$$t_{\text{Hg}} = \frac{\frac{k_2}{k_3} + [\text{Br}]}{k_1 [\text{Br}]^2}$$

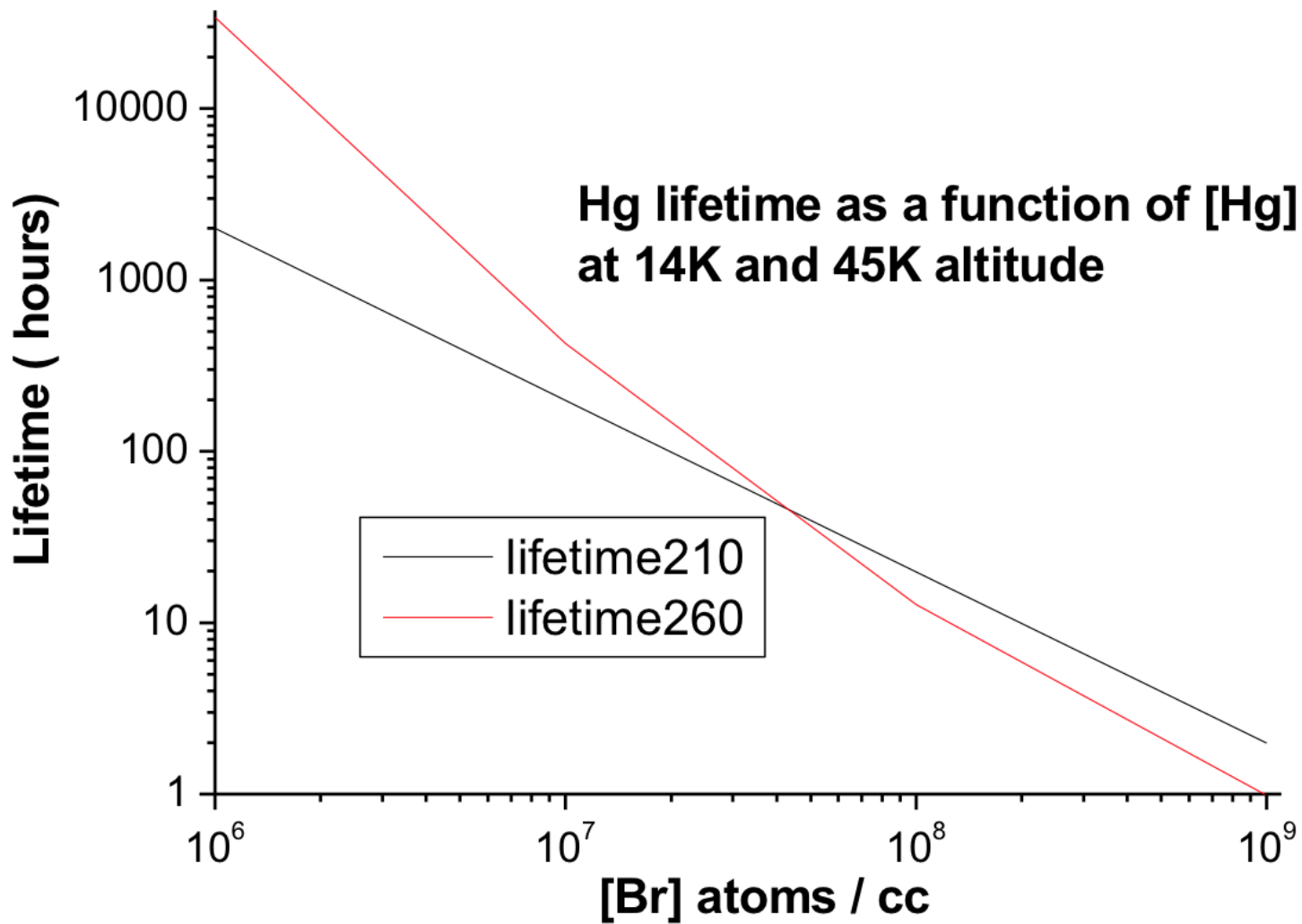
At 210K, 110 Torr

τ due to reaction with Br =

$$t_{Hg} = \frac{1}{k_1 [Br]}$$

At [Br] = 1e9 : 2 hours

1e8 : 20 hours



Hg on TOREO

- Late, “bootlegged” addition to project
- Challenges from fiscal and personnel issues
- Thanks to Rainer, RAF staff for getting us onboard

TORERO Hg Objectives

- **Measure RGM at a variety of altitudes**
- **Look for evidence of halogen initiated oxidation of GEM**

Approach:

Sample on KCl coated annular denuders

Standard operating procedure is to sample at 10 SLPM

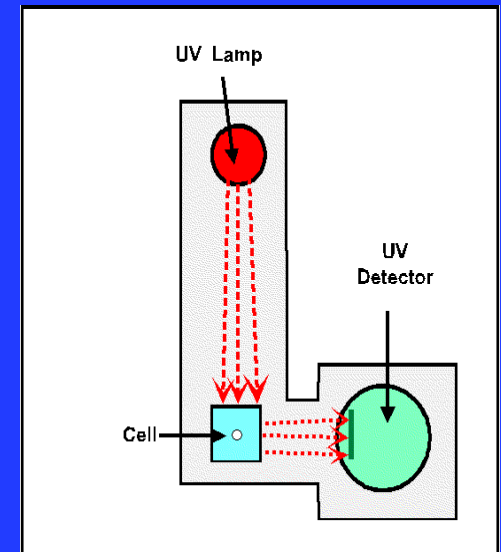
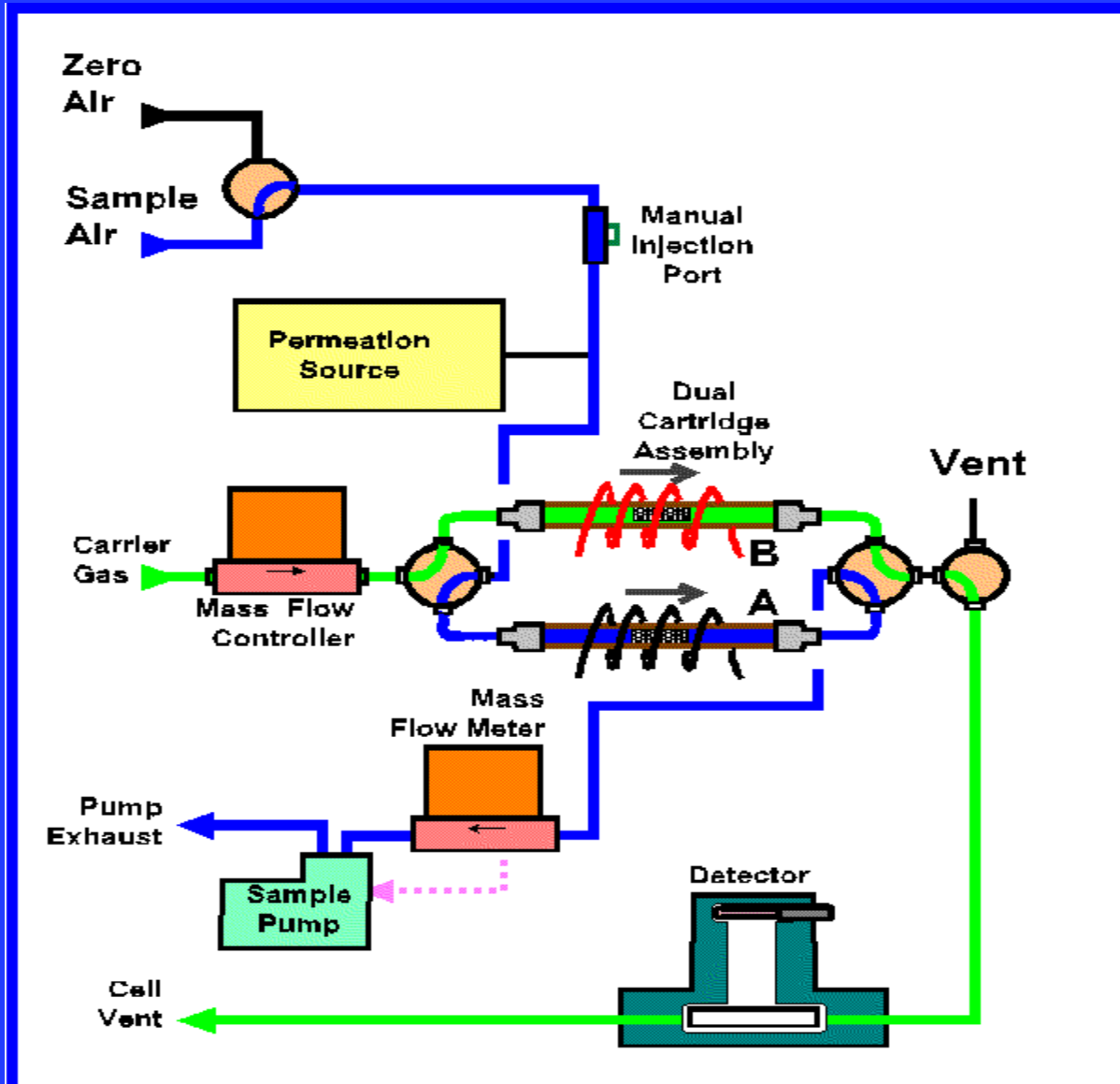
During TORERO we were limited to 10 LPM volumetric so low mass sampling at high altitudes

Most flights: 5 samples, 2 blanks, 1 bypass

Denuders analyzed immediately post flight using Tekran 2537A Hg analyzer (gold amalgamation / CVAFS)

Mass flow calculated by integrating flight pressure / temperature profile

CVAFS instrument with presampling on gold (TEKTRAN)



CVAFS instrument with presampling on gold

Preconcentrate to enhance sensitivity

Remove very efficient quenching gases

N_2 and O_2

**Monitor resonance fluorescence in
inefficient quencher (Ar)**

Annular Denuder Methodology for the Measurement of RGM

Environ. Sci. Technol., 36 (13), 3000 -3009, 2002

Development and Characterization of an Annular Denuder Methodology for the Measurement of Divalent Inorganic Reactive Gaseous Mercury in Ambient Air

Matthew S. Landis

U.S. EPA, National Exposure Research Laboratory, Research Triangle Park, North Carolina 27711

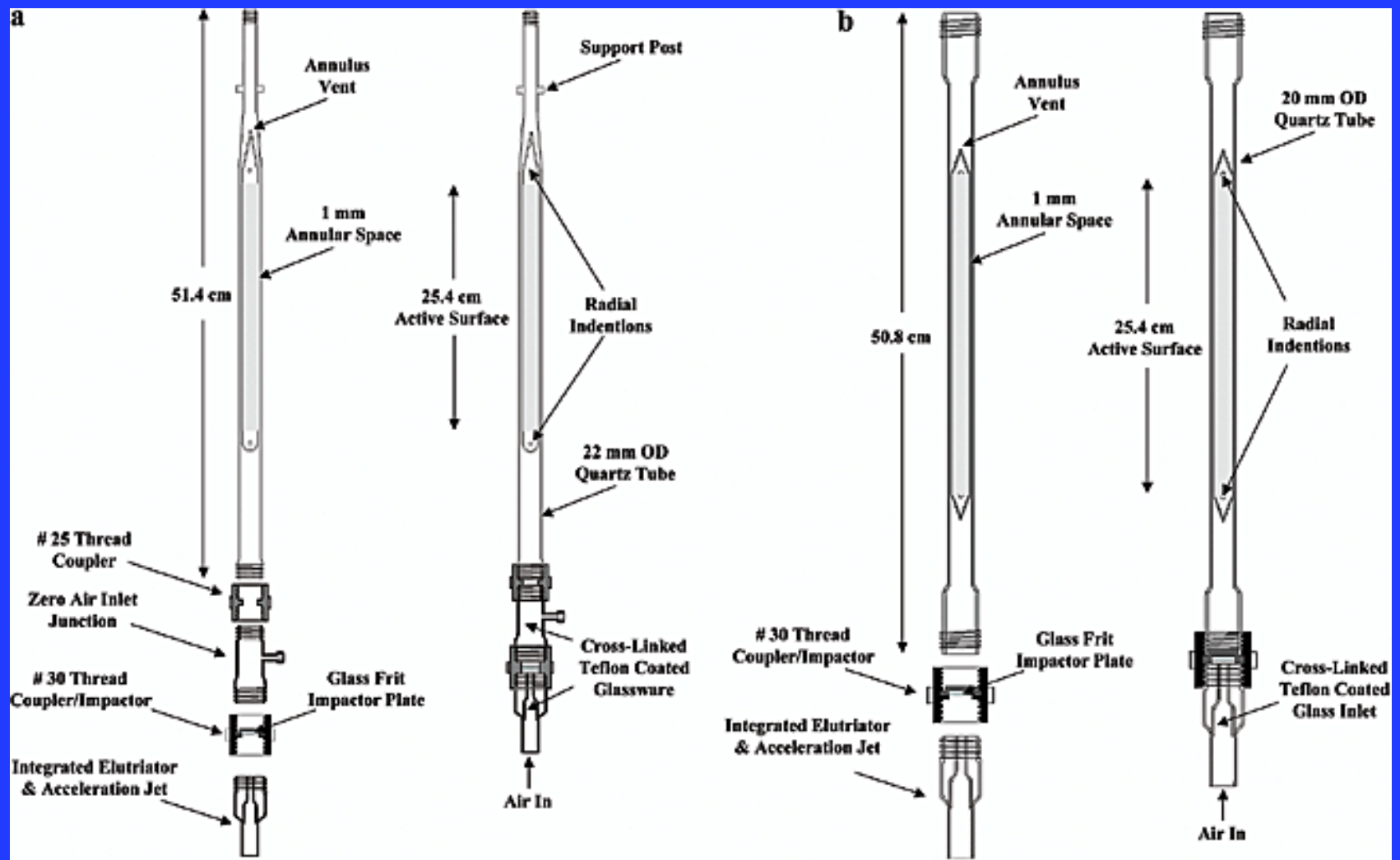
Robert K. Stevens

Florida Department of Environmental Protection, Tallahassee, Florida 32399

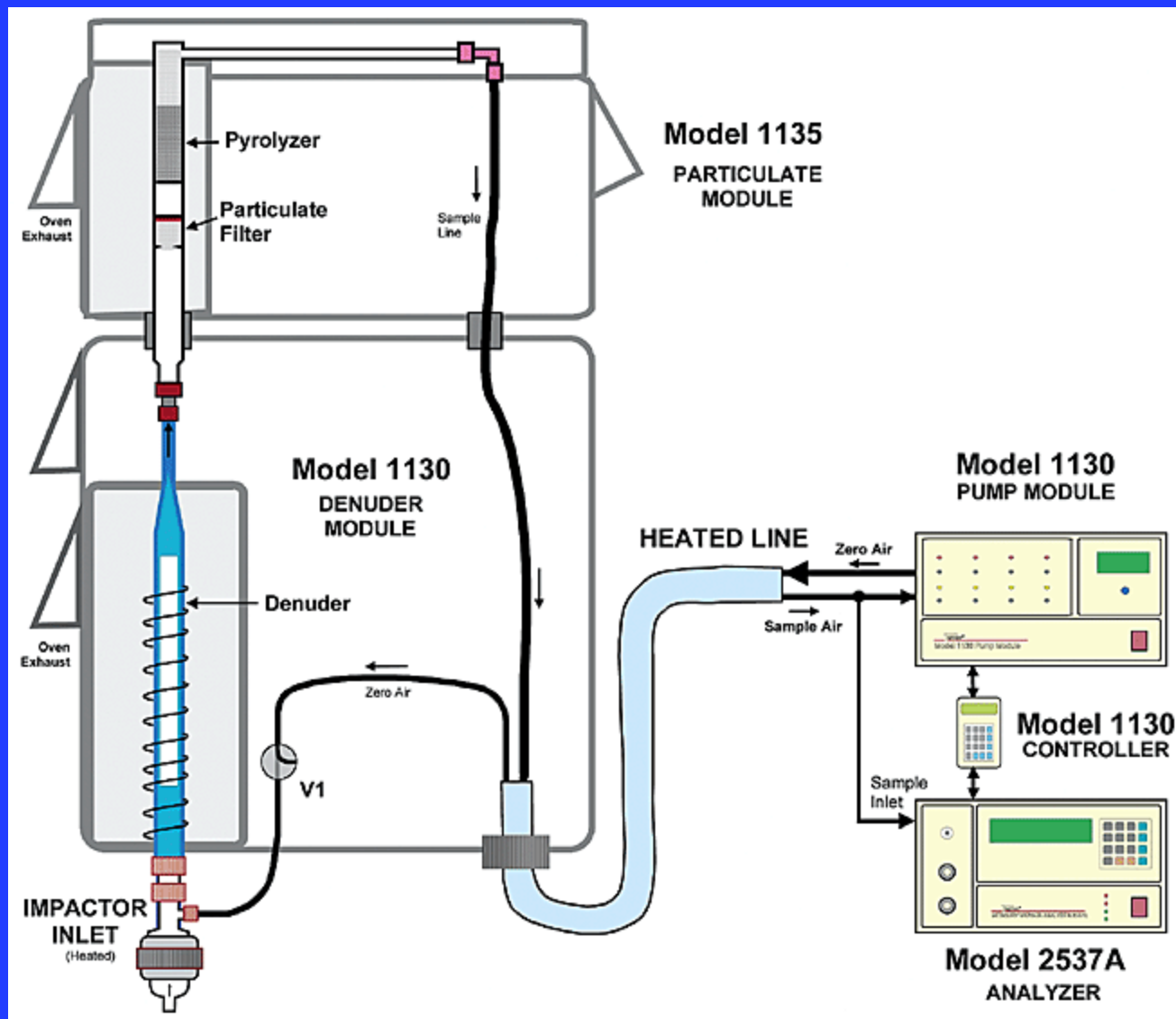
Frank Schaedlich

Tekran Inc., 1-132 Railside Road, Toronto, Ontario M3A 1A3, Canada

Eric M. Prestbo



Diagrams of disassembled and assembled quartz annular denuder systems for (a) automated and (b) manual methods, respectively.



Schematic diagram of Tekran automated mercury speciation instrumentation showing the configuration of the model 1130 denuder module and the model 1135 particulate mercury unit.

Research Flight 12:

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		Average height	vol (Std m ³)	Denuder load	pg/m ³	pg/Std m ³	2 sig error
1.....	high 35-38K	37796	0.07	9.88	-24.52	-160.38	79.4
2.....	low 0.3-19K	11587	0.54	23.48	1.87	2.95	19.2
3.....	high 43-44K	42746	0.05	16.41	-12.19	-105.28	104.8
4.....	high 44-45.5K	44715	0.05	57.31	78.71	679.80	104.8
5.....	low 0.3-23K	10139	0.62	38.47	17.83	26.34	17.9

Research Flight 13:

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		Average height	vol (Std m ³)	Denuder load	pg/m ³	pg/Std m ³	2 sig error
1	D 7 high	34910	0.09	4.72	-3.33	-17.76	18.7
2	D 6 high	39931	0.07	14.46	15.38	114.00	25.9
3	D 5 high	44806	0.04	5.76	-1.37	-13.32	34.0
4	D 3 high	45521	0.04	8.36	4.71	51.87	35.5
5	D 2 high	46524	0.03	5.50	-2.38	-25.67	37.8

Research Flight 14:

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		Average height	vol (Std m ³)	Denuder load	pg/m ³	pg/Std m ³	2 sig error
1	D 2 high	39819	0.16	18.82	10.69	78.61	25.7
2	D 3 low	10384	0.36	39.38	59.04	90.91	5.4
3	D 5 high	44595	0.05	9.24	5.75	55.31	33.7
4	D 7 high	47246	0.04	6.21	-0.22	-2.37	38.3

Research Flight 15:

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		Average height	vol (Std m ³)	Denuder load	pg/m3	pg/Std m3	2 sig error
1	D 5 high	38466	0.09	18.19	19.79	135.11	23.9
2	D2 6000-26000	14931	0.33	33.08	43.53	80.78	6.5
3	D 3 300-5000	1063	0.91	49.48	45.44	47.39	3.6
4	D 6 high	44697	0.12	12.79	5.81	56.11	33.8
5	D 7 high	47528	0.05	11.56	10.10	114.41	39.6

Research Flight 16:

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		Average height	vol (Std m ³)	Denuder load	pg/m ³	pg/Std m ³	2 sig error
1	D 2 < 7000ft	2563	0.5	27.46	38.49	42.55	3.9
2	D 3 300-5000	1544	0.59	45.42	62.94	66.83	3.7
3	D 5 >40000	43658	0.07	17.54	18.42	167.13	31.7
4	D 7 40-45Kft	47091	0.05	14.92	16.25	179.40	38.6
5	D 6 6-26Kft	19162	0.1	12.28	25.80	59.10	8.0

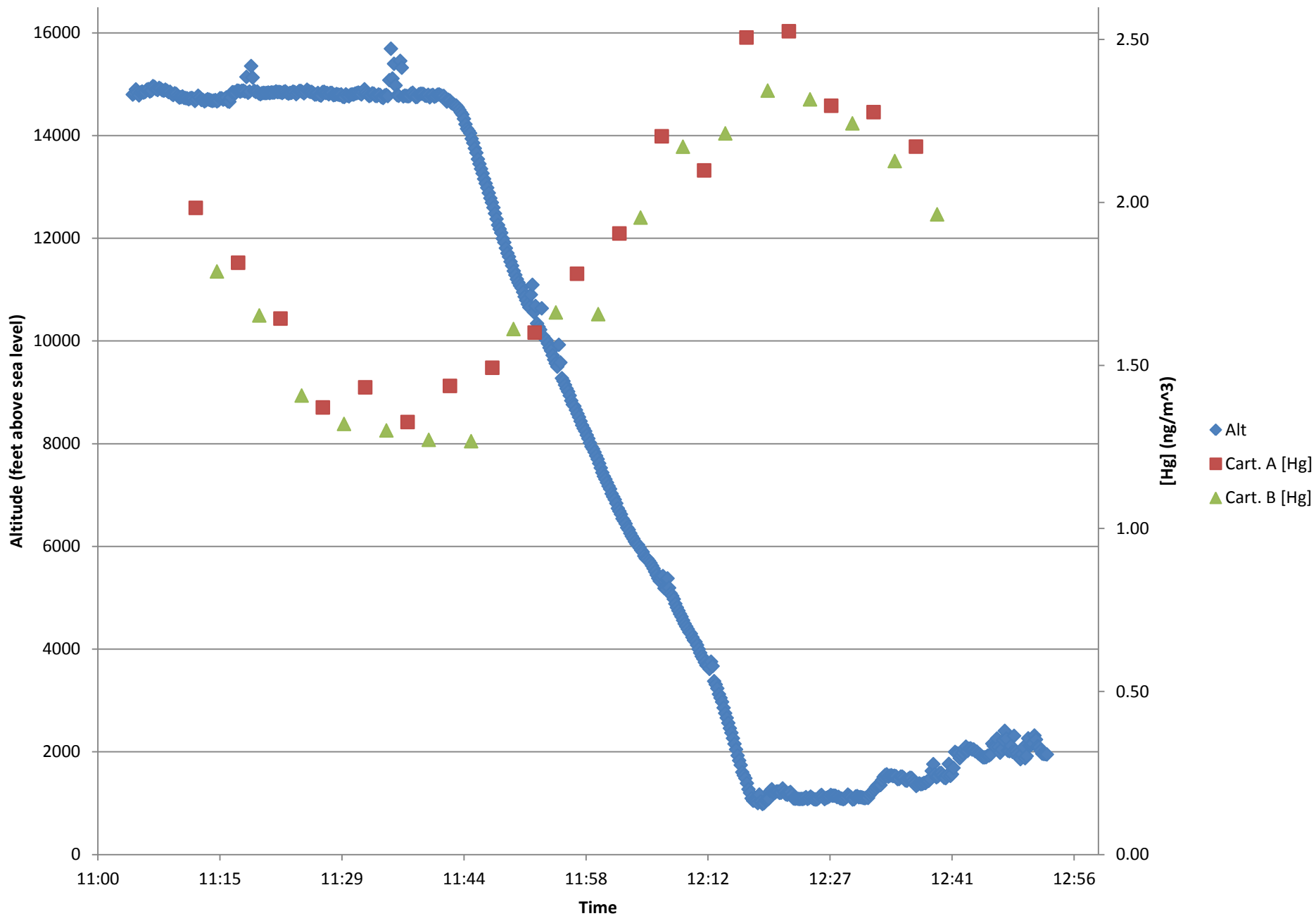
Research Flight 17:

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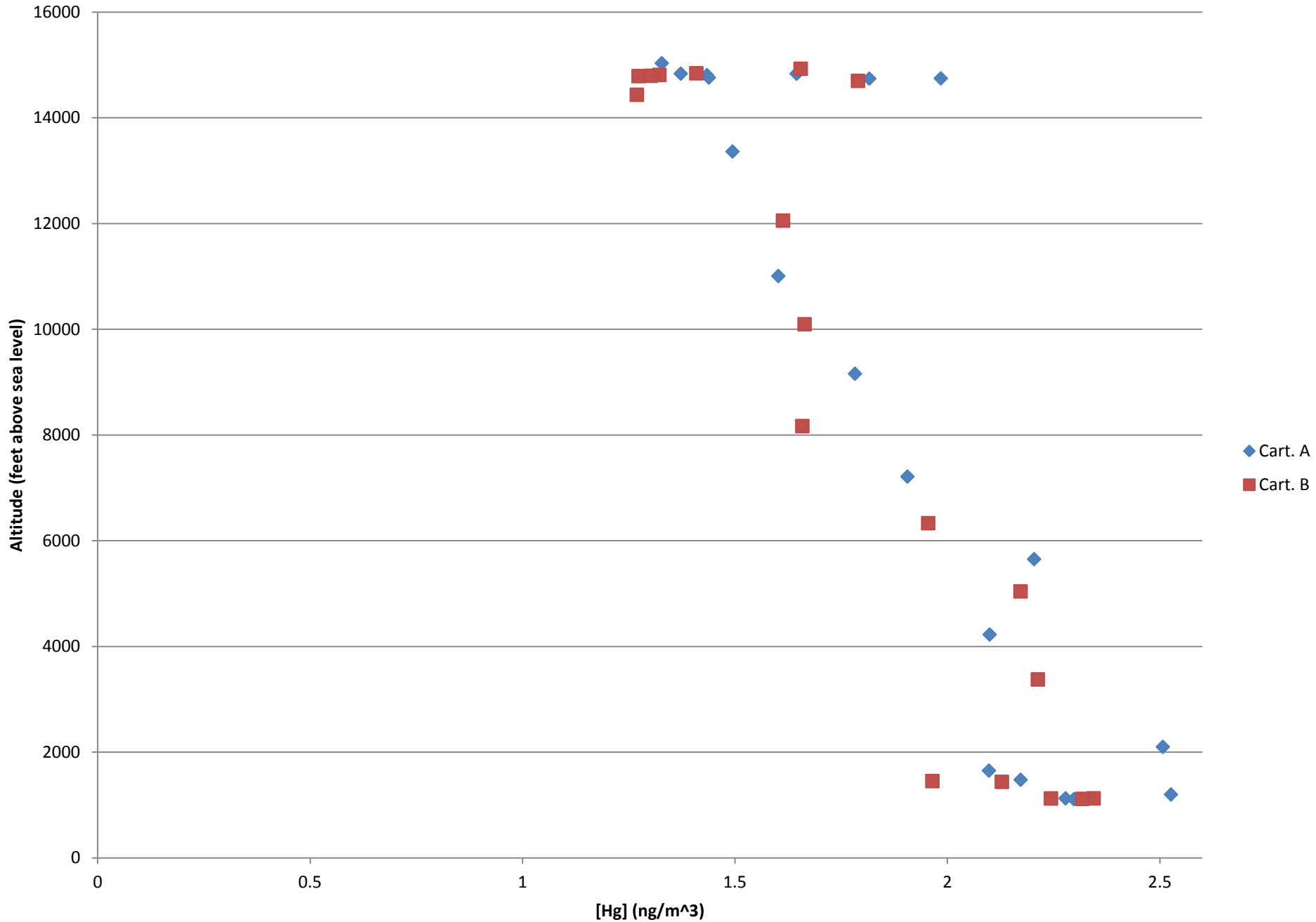
		Average height	vol (Std m ³)	Denuder load	pg/m ³	pg/Std m ³	2 sig error
1	D 5 33000	33500	0.10	17.20	21.77	110.83	17.8
2	D 2 6k-26kft	11685	0.22	15.43	28.48	45.48	5.6
3	D 3 300-5000	1102	0.32	21.76	45.44	47.34	3.6
4	plumbing	problem :		bypass	partially open		
5	D 7 40-45Kft	41260	0.04	8.55	7.71	59.79	27.1

Airborne Hg Sampling in Grand Bay, MS

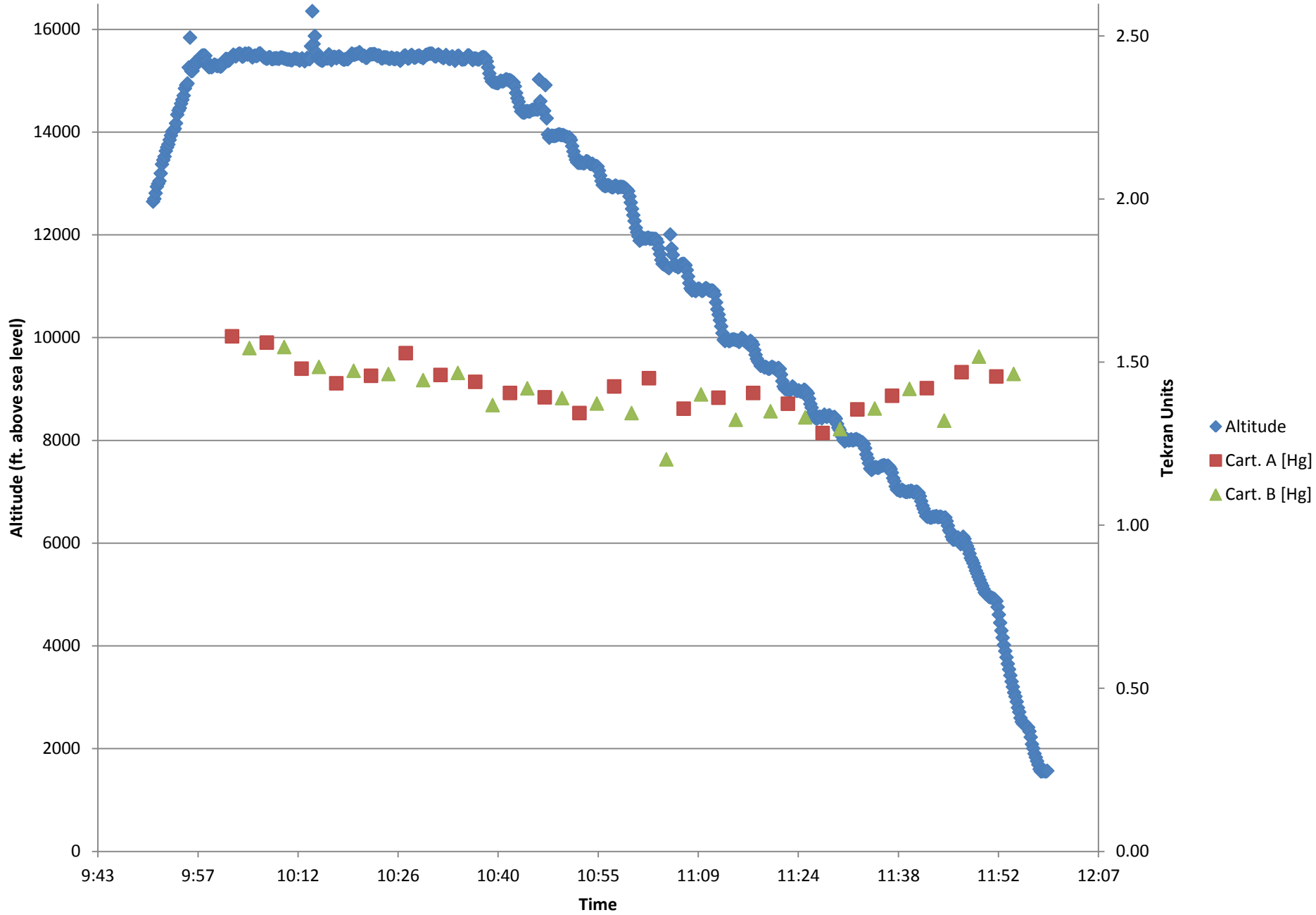
100804 (Flight #2)



100804 Altitude Profile



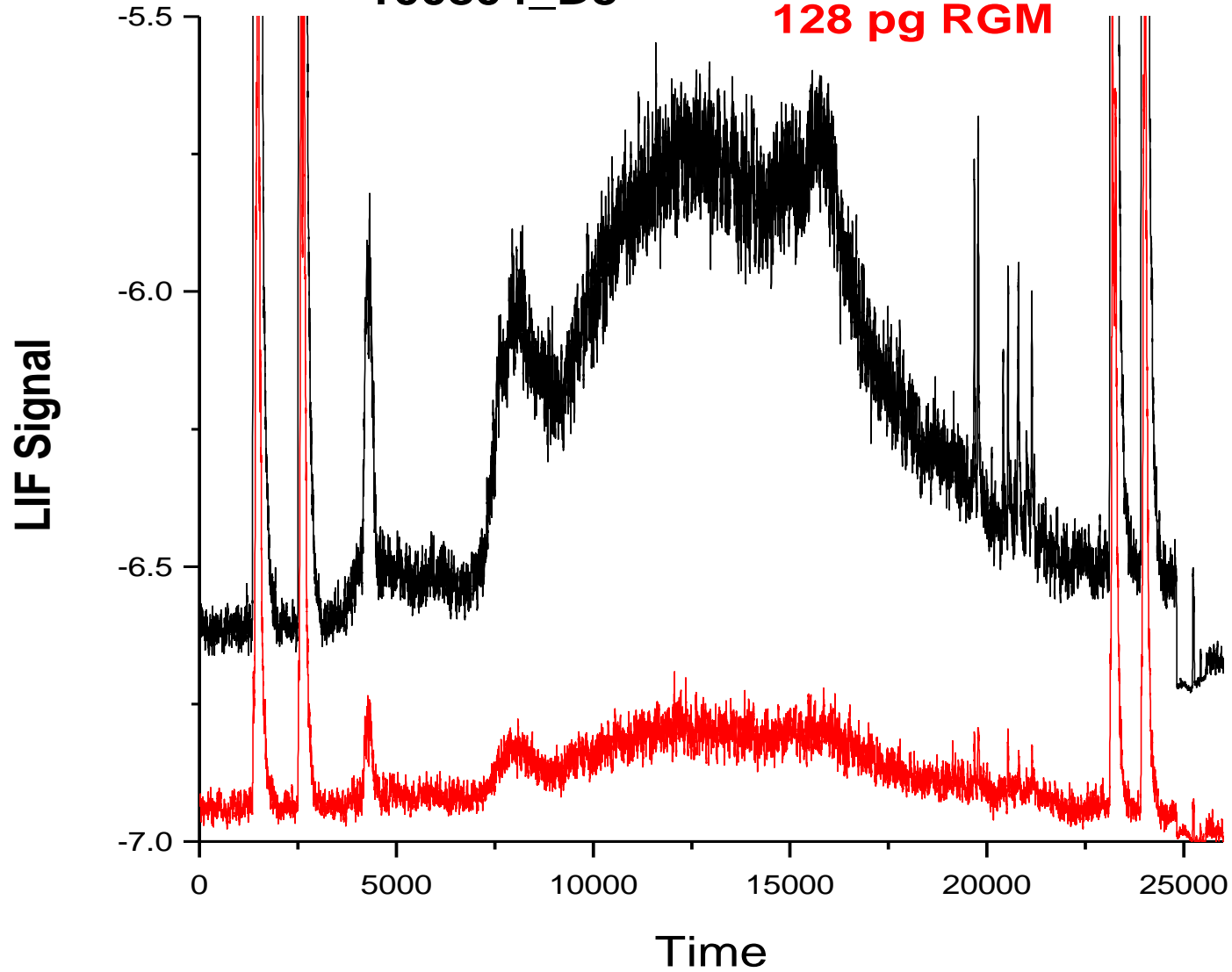
100811 (Flight #5)

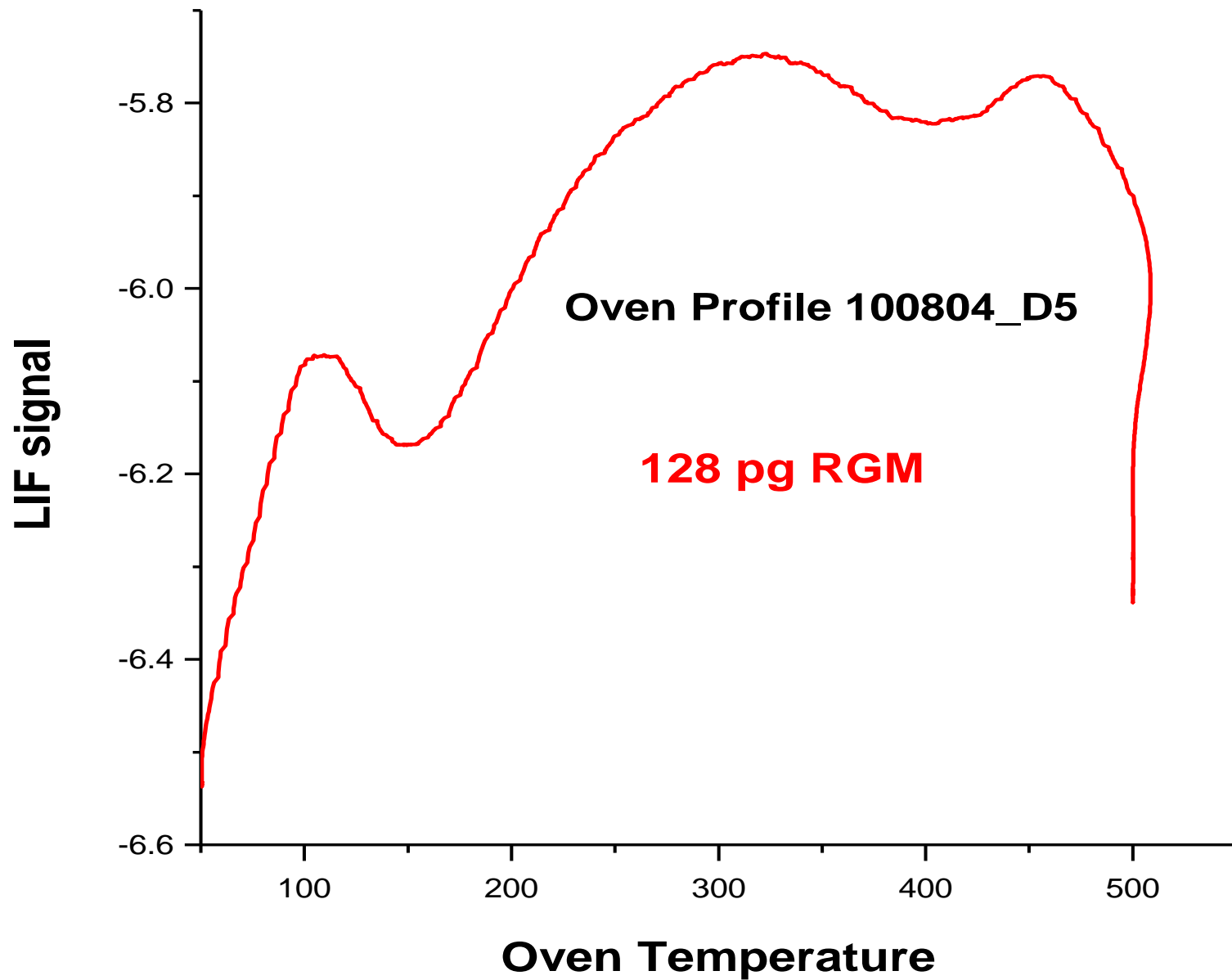


Denuder Analysis Profile

100804_D5

128 pg RGM

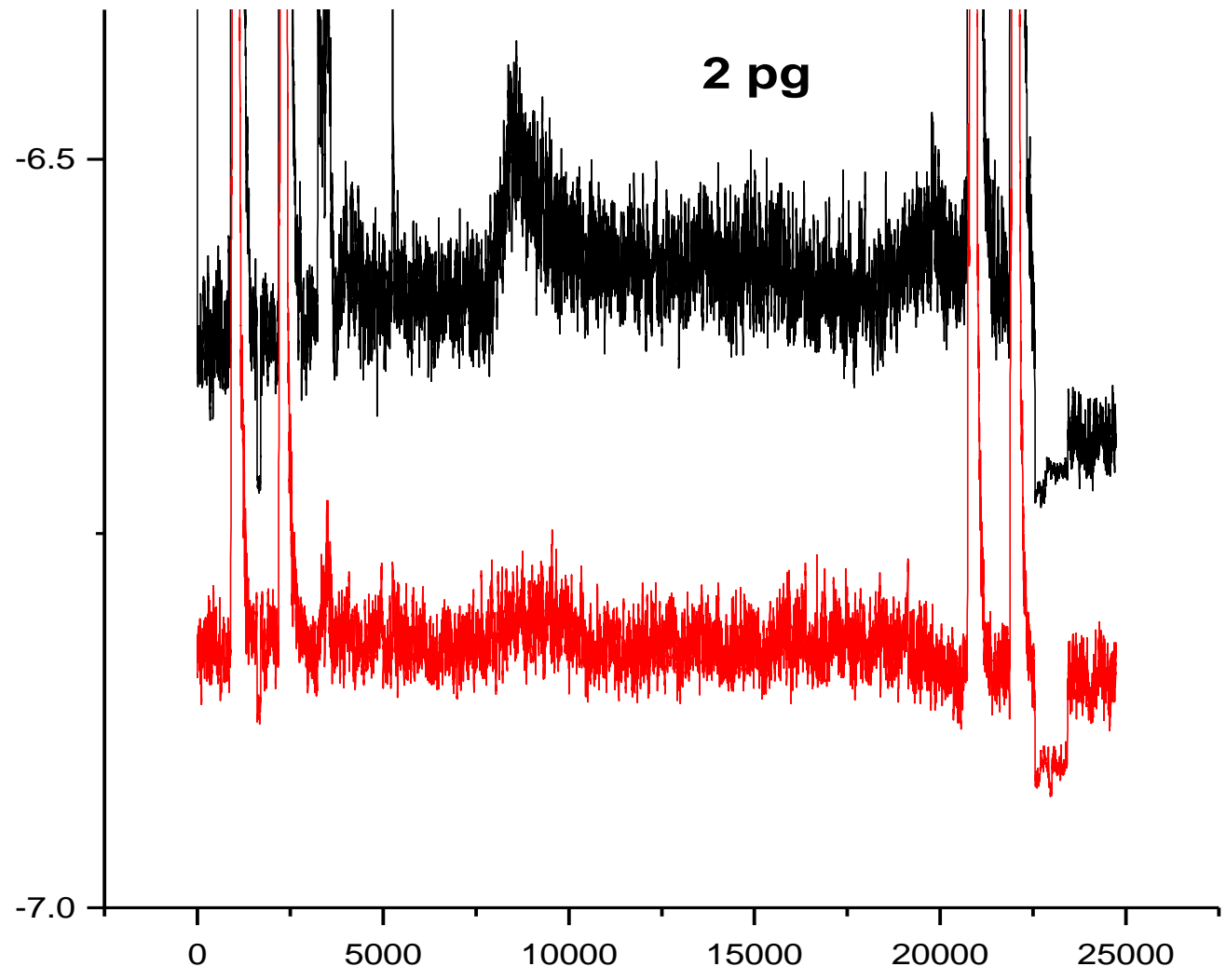




100804D6

2 pg

LIF Signal



-6.5

-7.0

0

5000

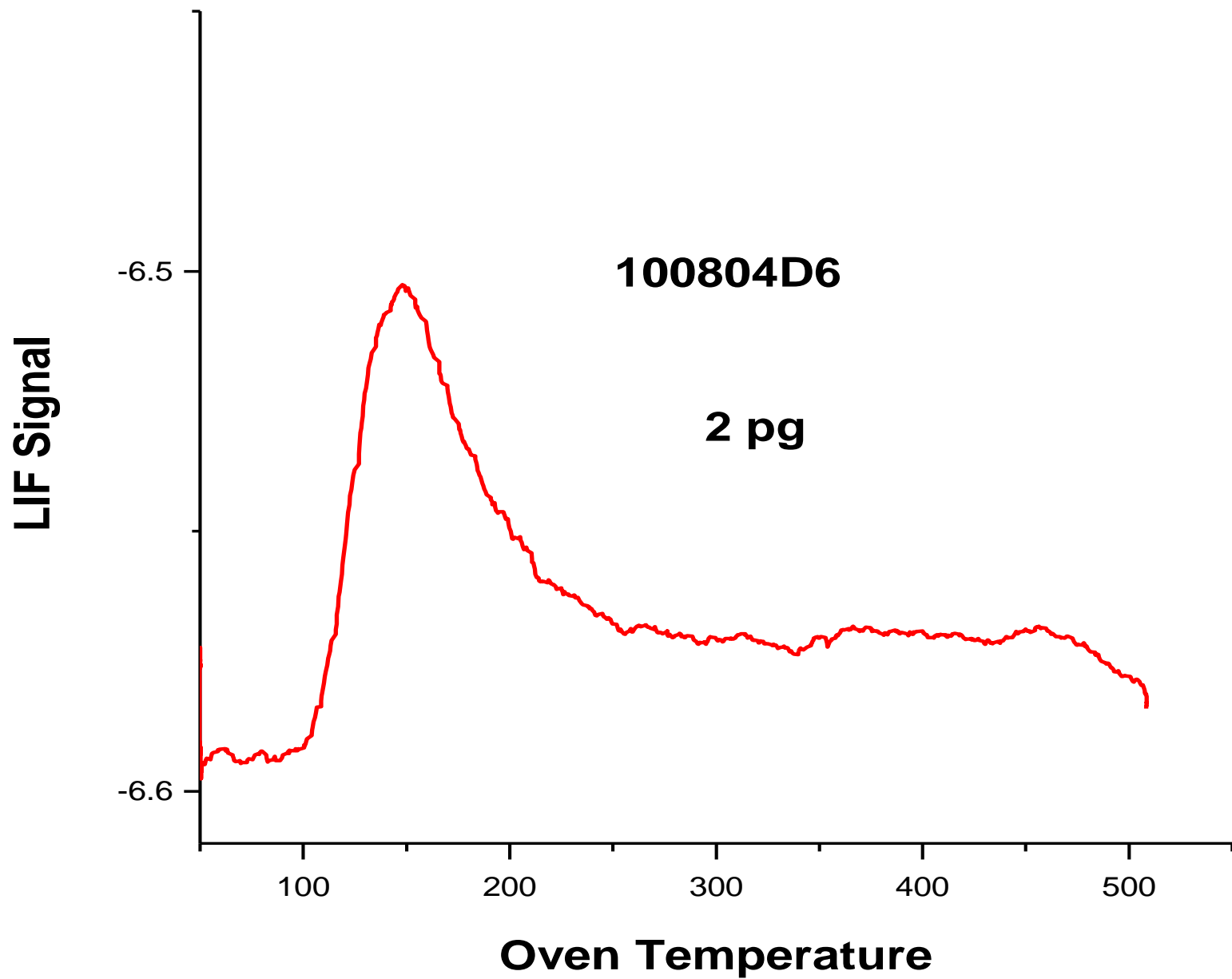
10000

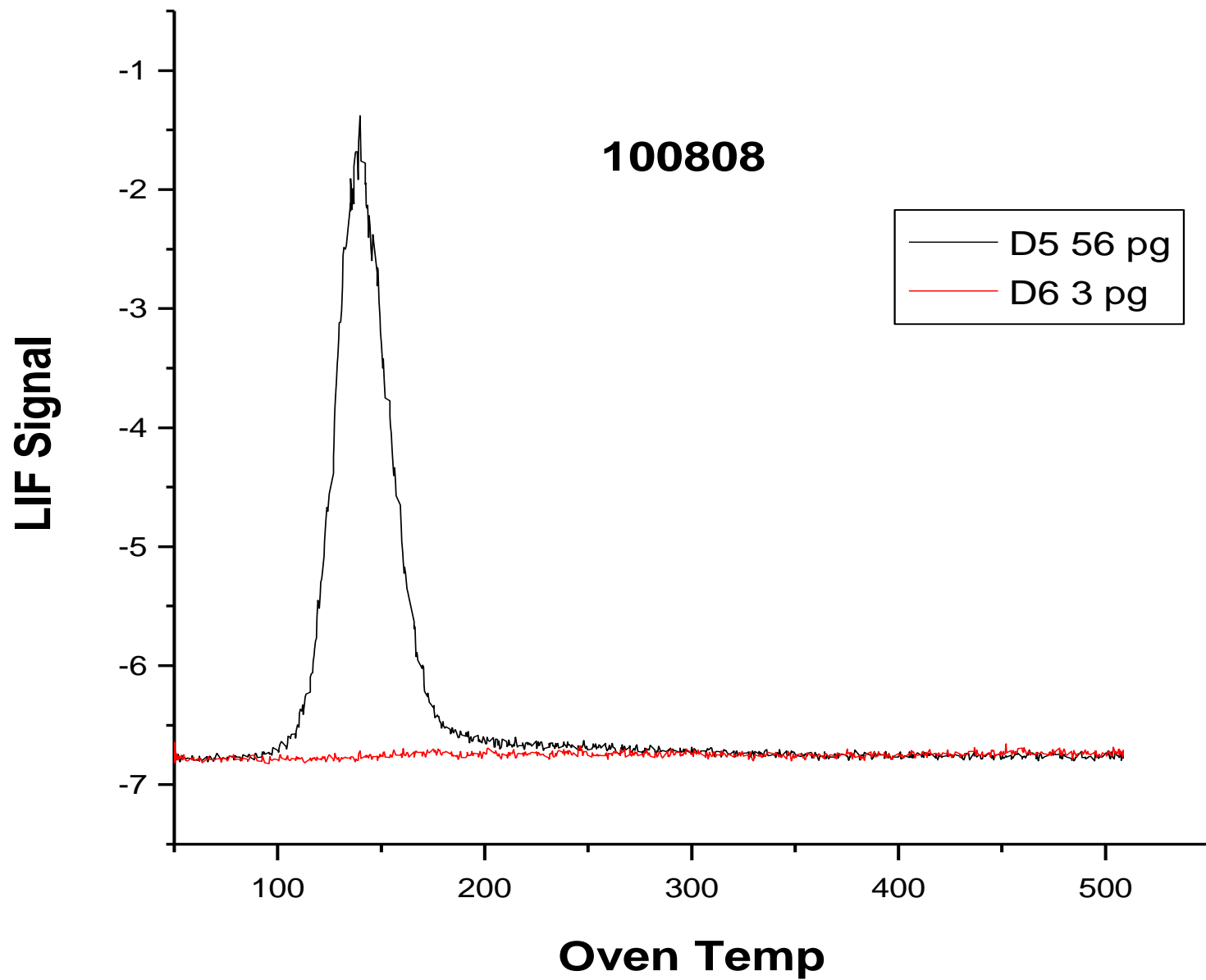
15000

20000

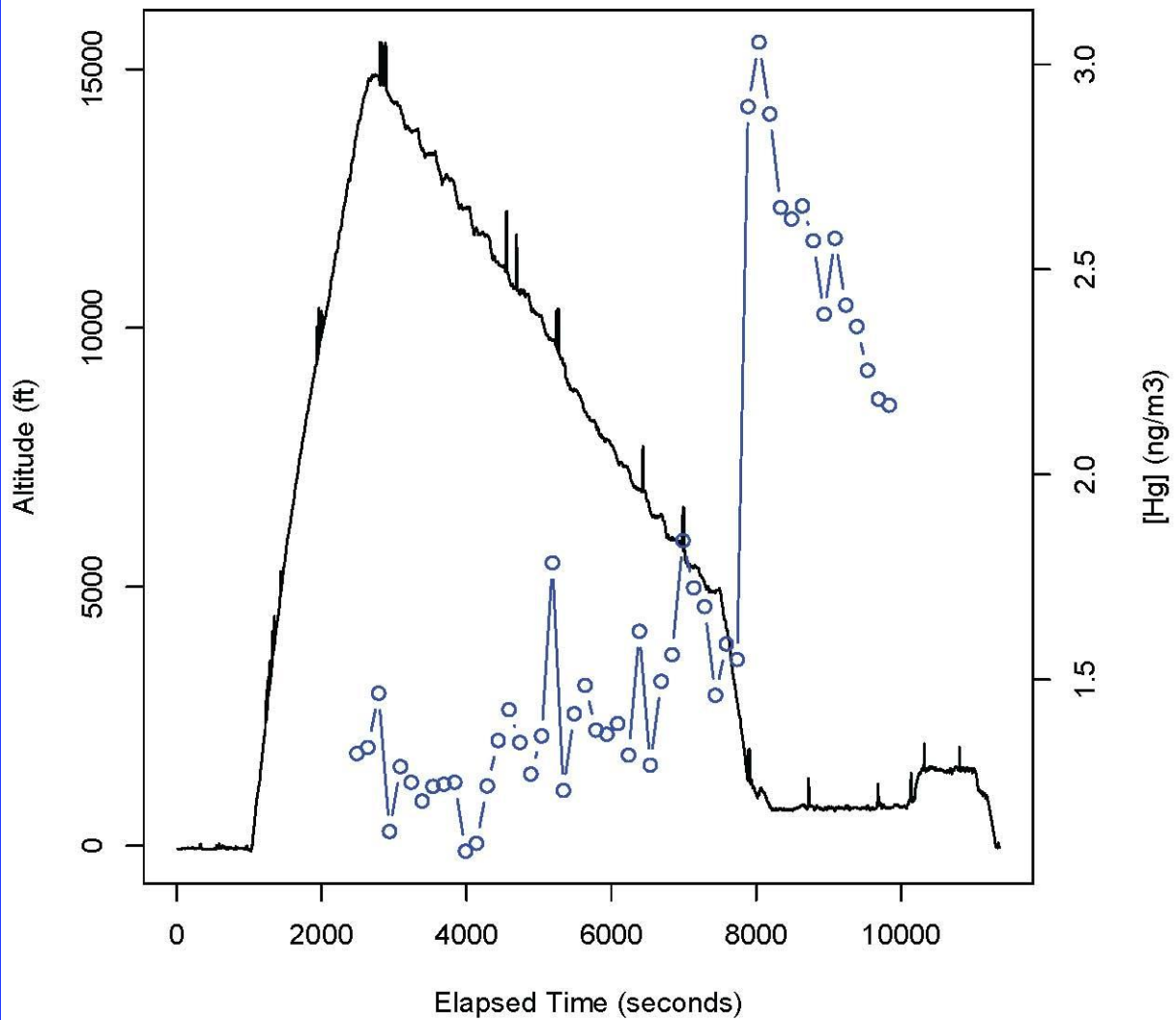
25000

Time

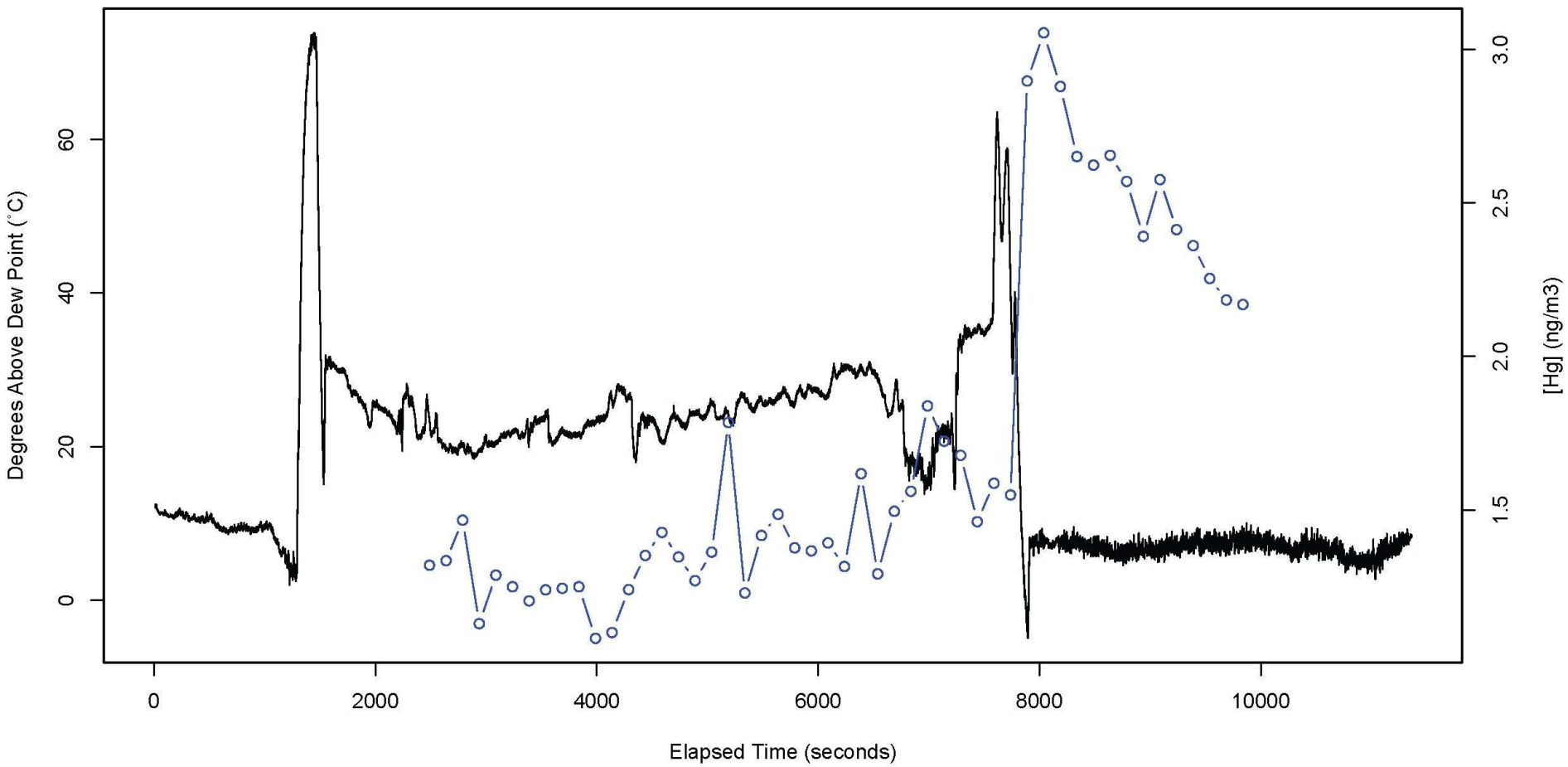




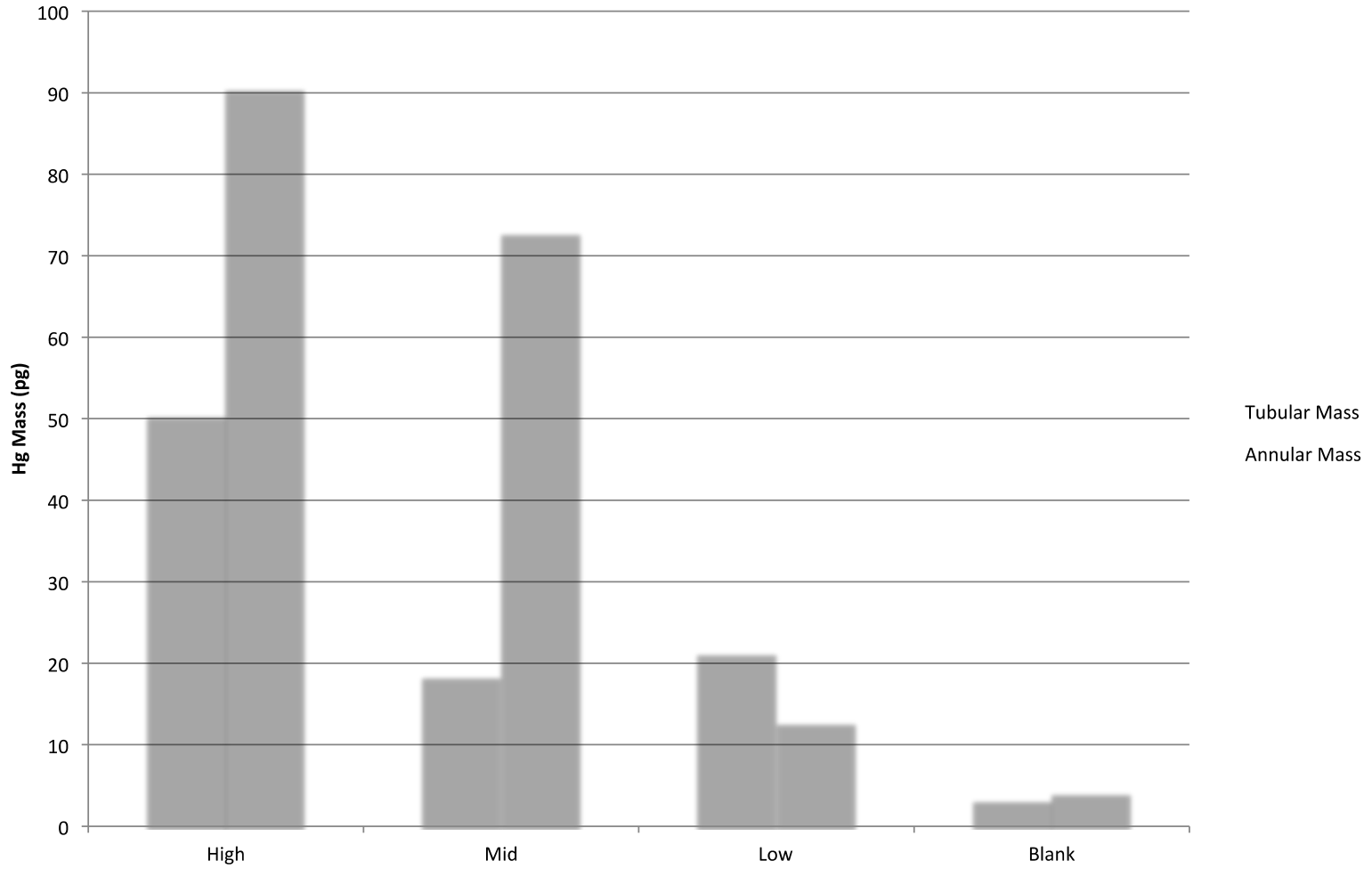
30 April 2011



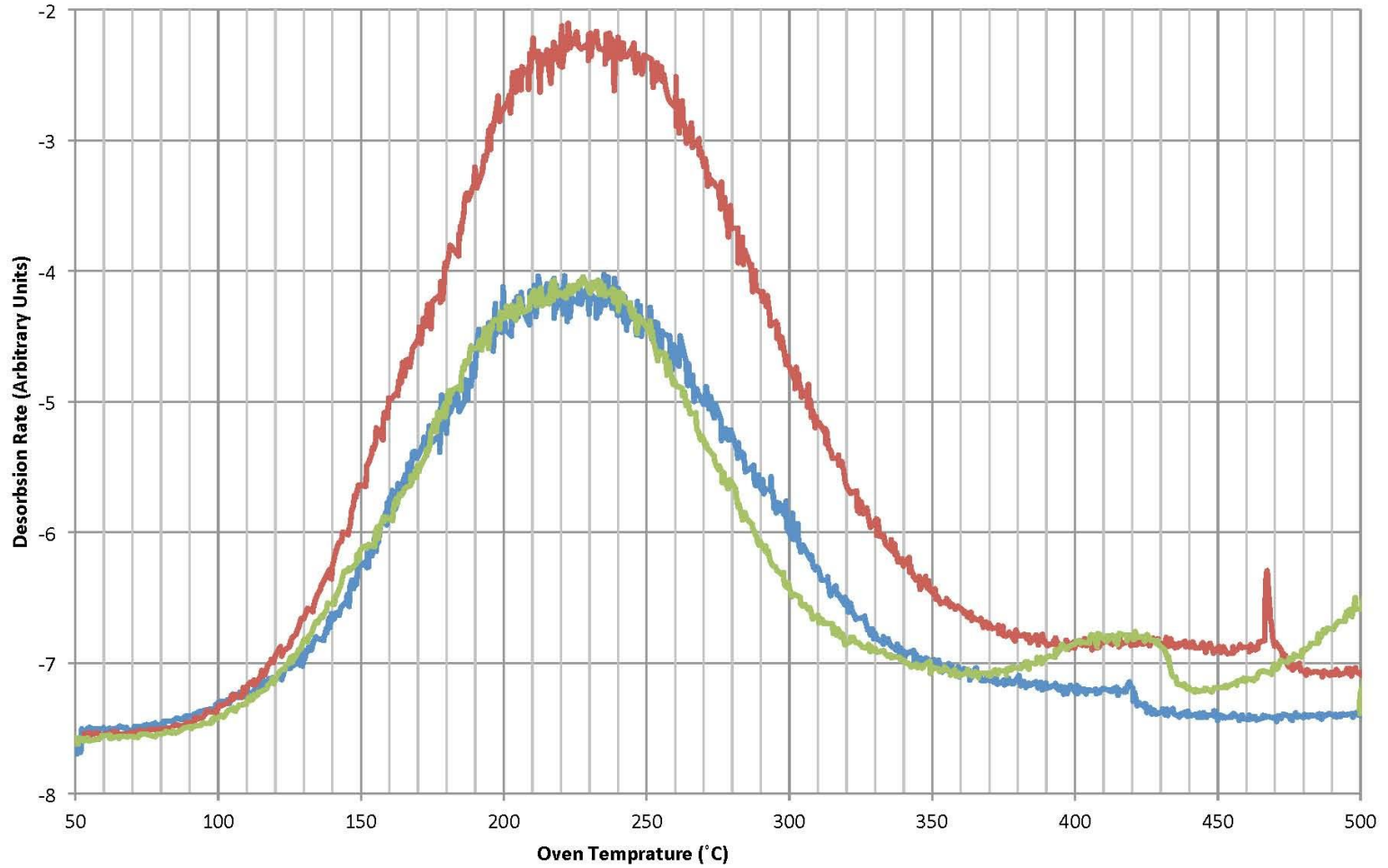
Flight #4: 30 August 2011



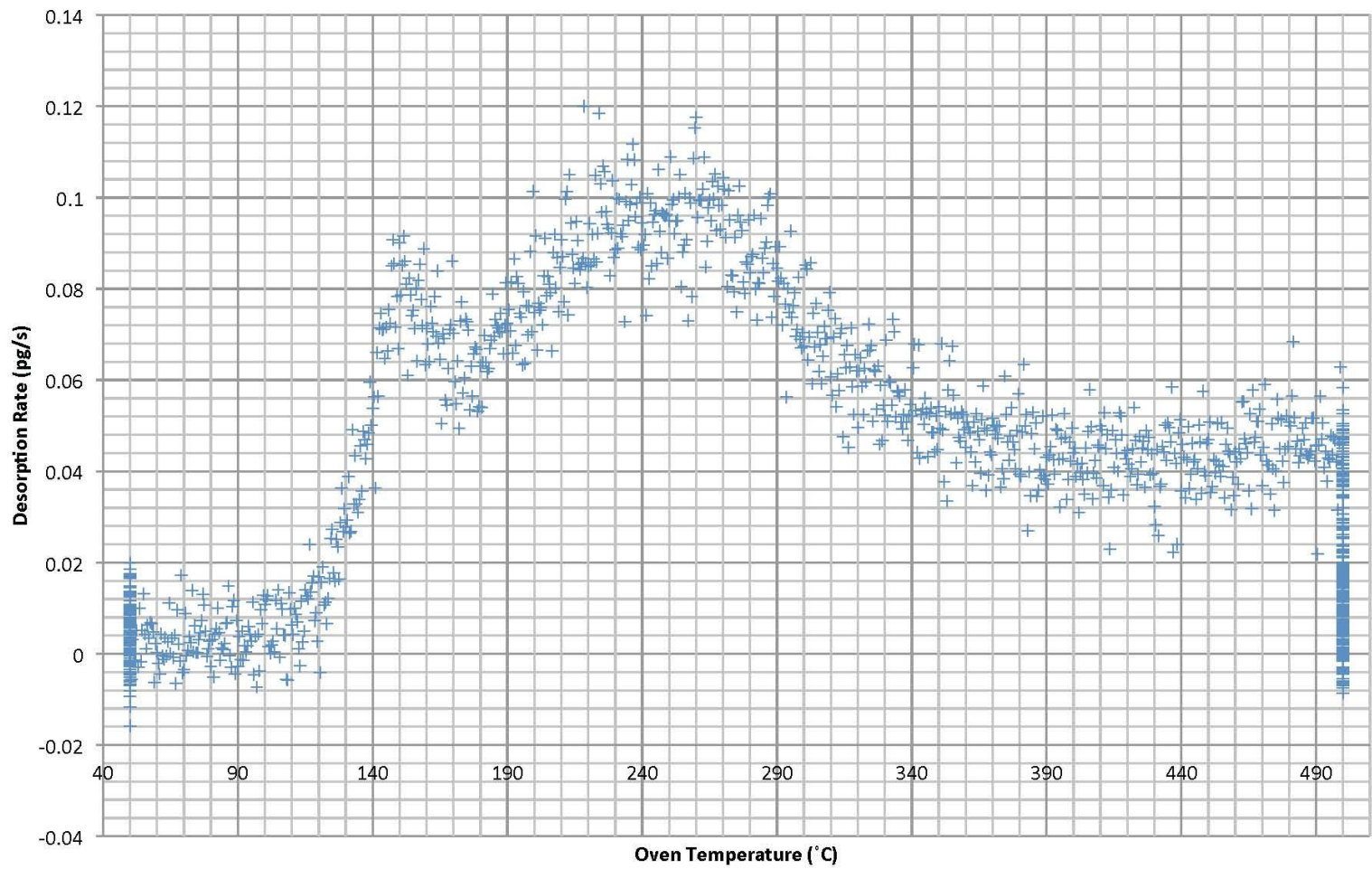
110430 Mass Summary



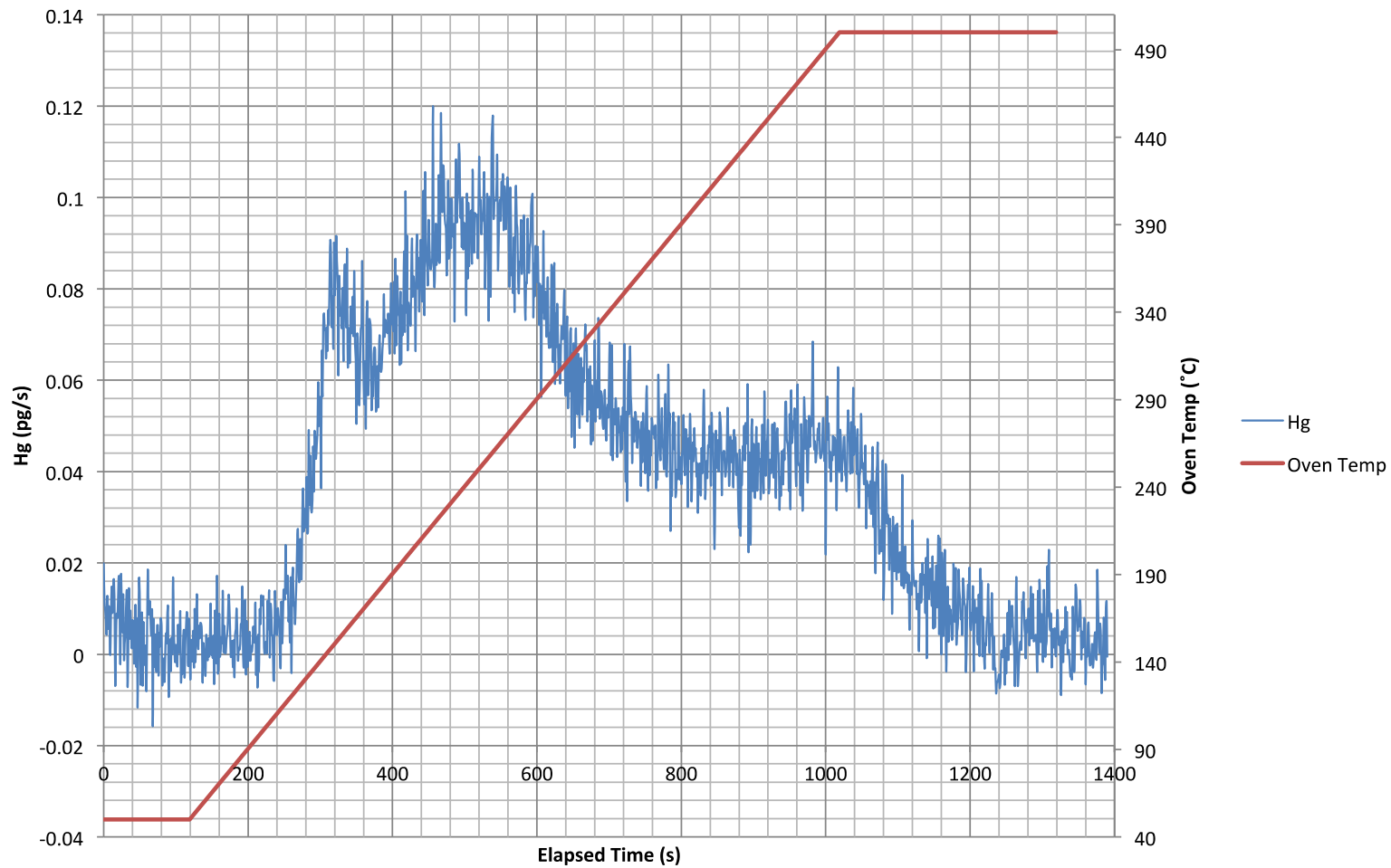
HgCl₂ Desorption (Std. Oven Program)



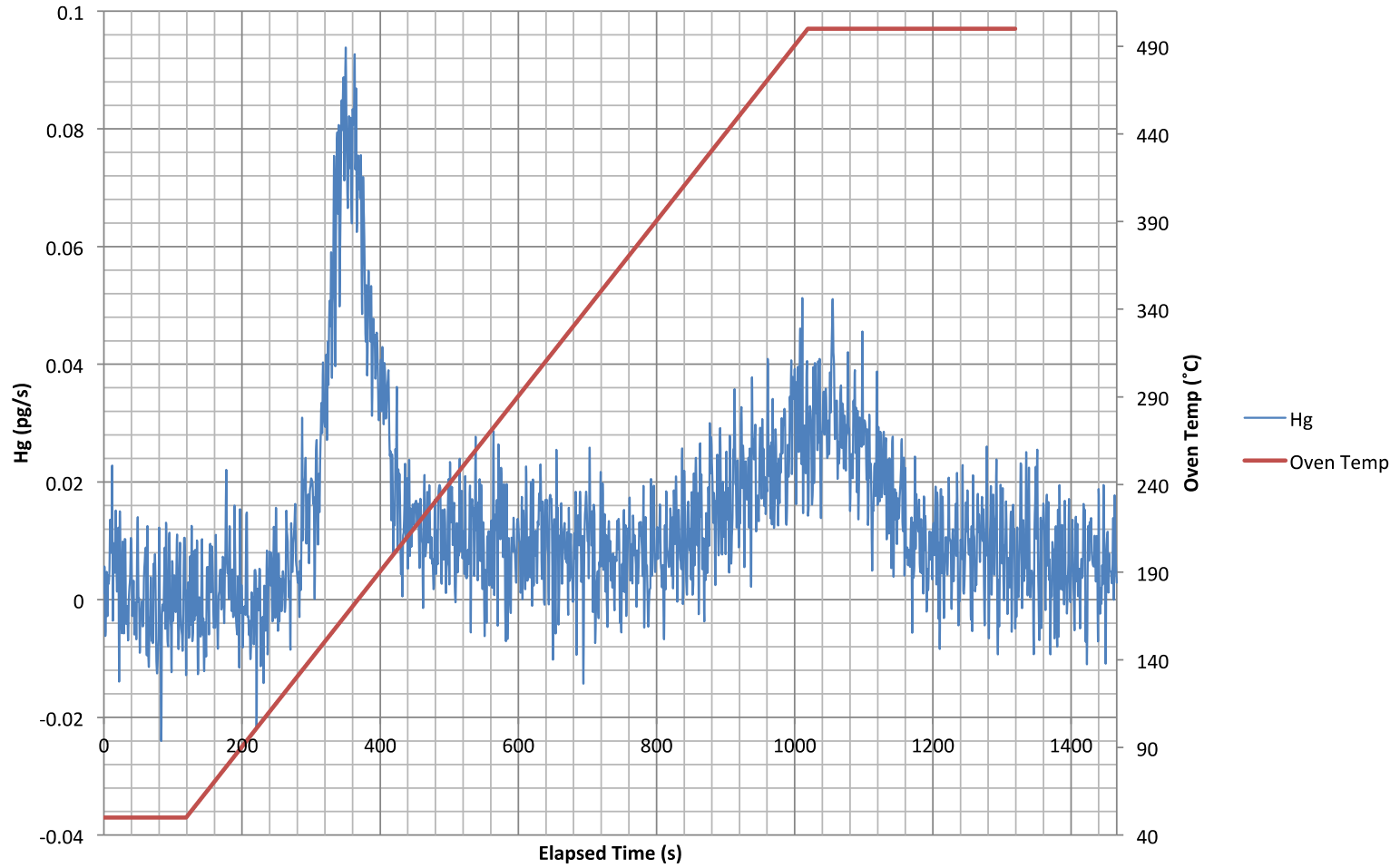
110430_D5: 50.6 pg (High Alt.)



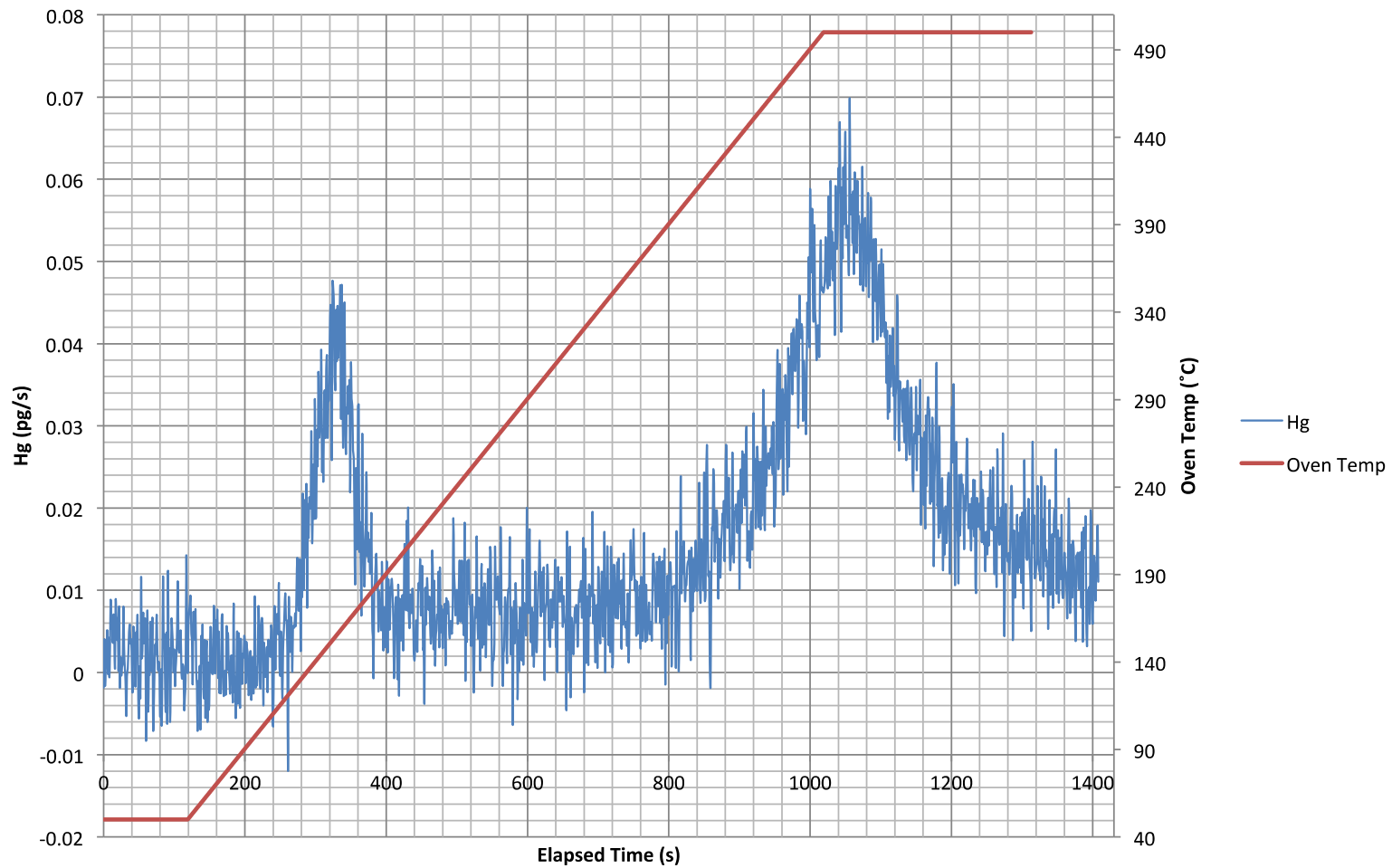
110430_D5: 50.6 pg (High Alt.)



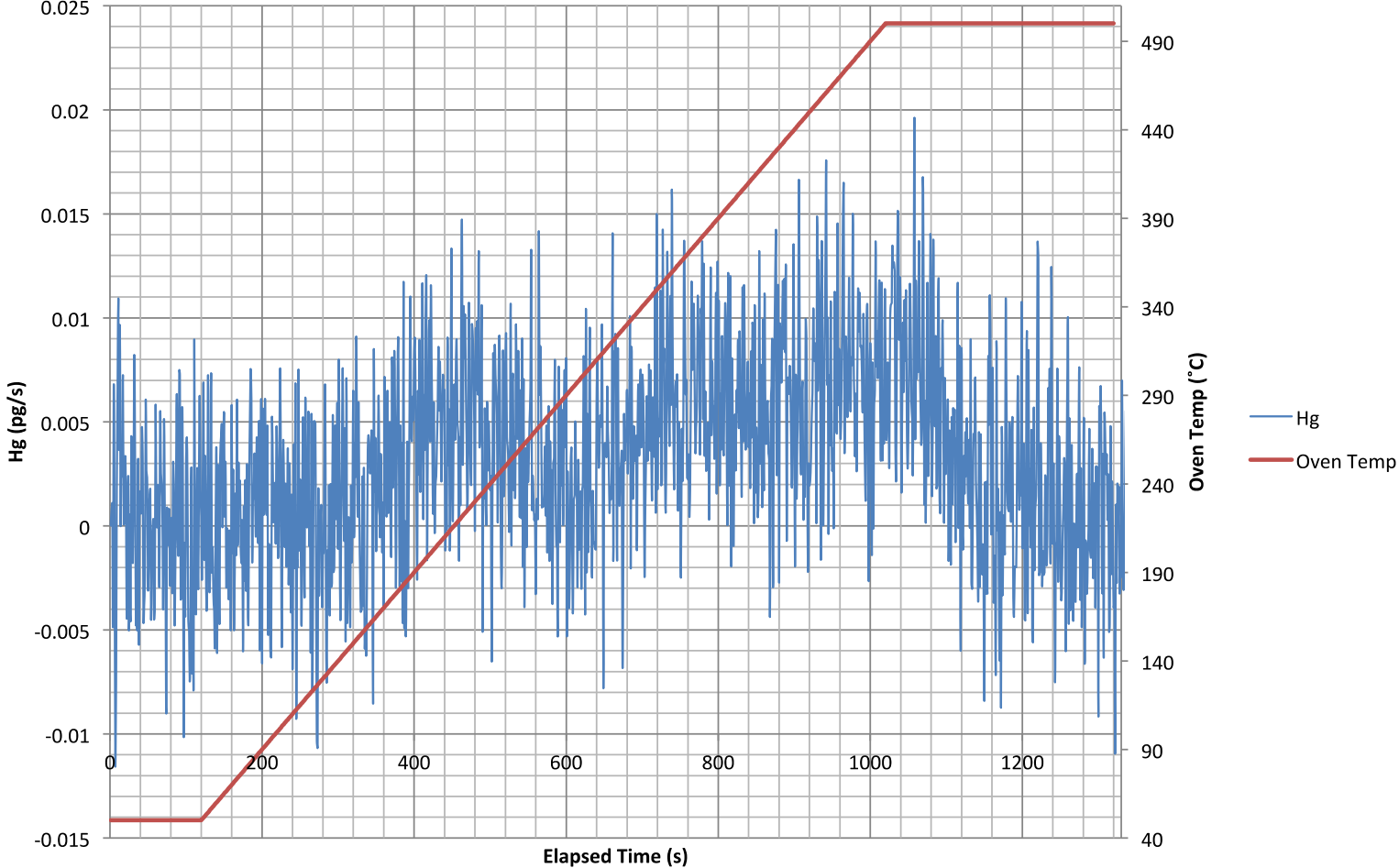
110430_D6: 18.5 pg (Mid Alt.)



110430_D7: 21.4 pg (Low Alt.)

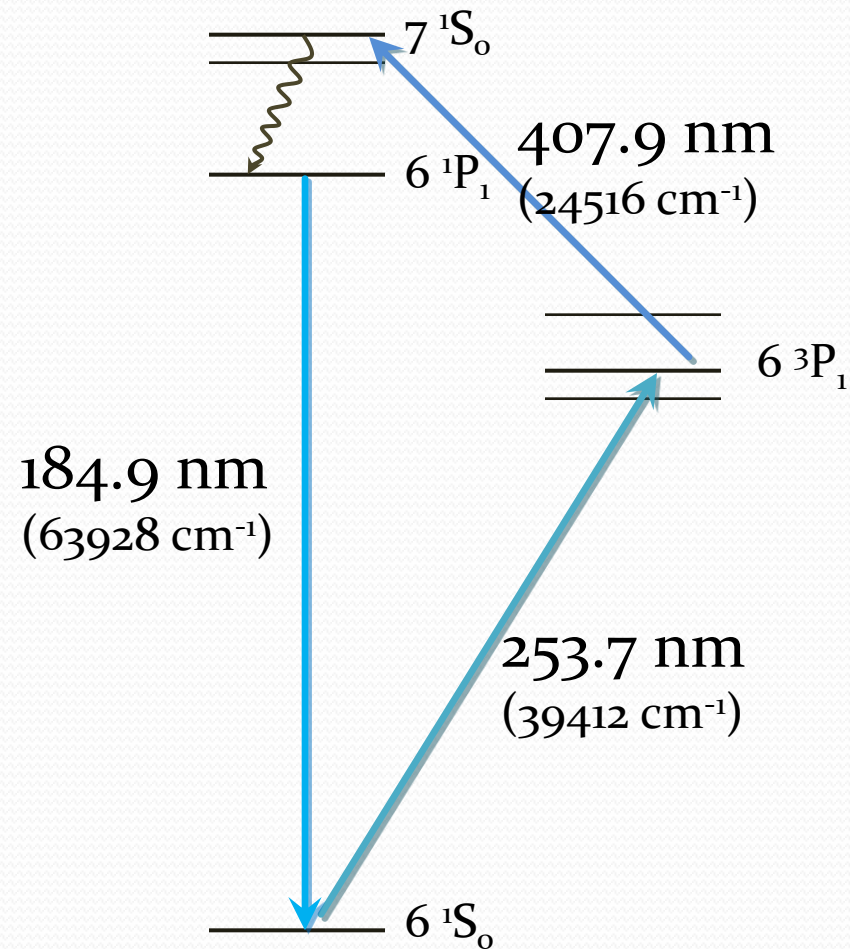


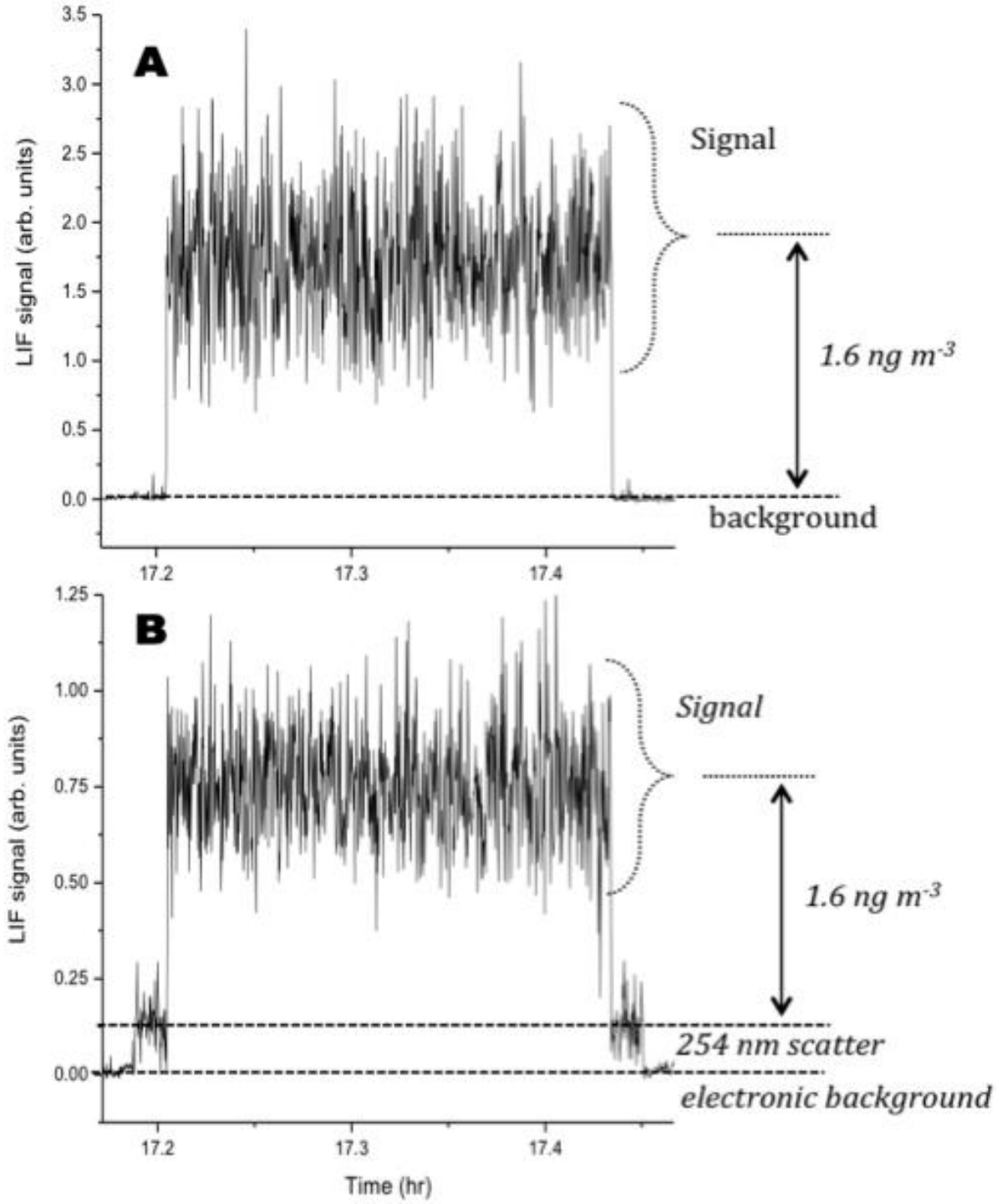
110430_D8: 3.4 pg (Blank)



In-situ measurement of GEM using 2-photon laser-induced fluorescence

- Differentiation between probe lasers and fluorescence signal





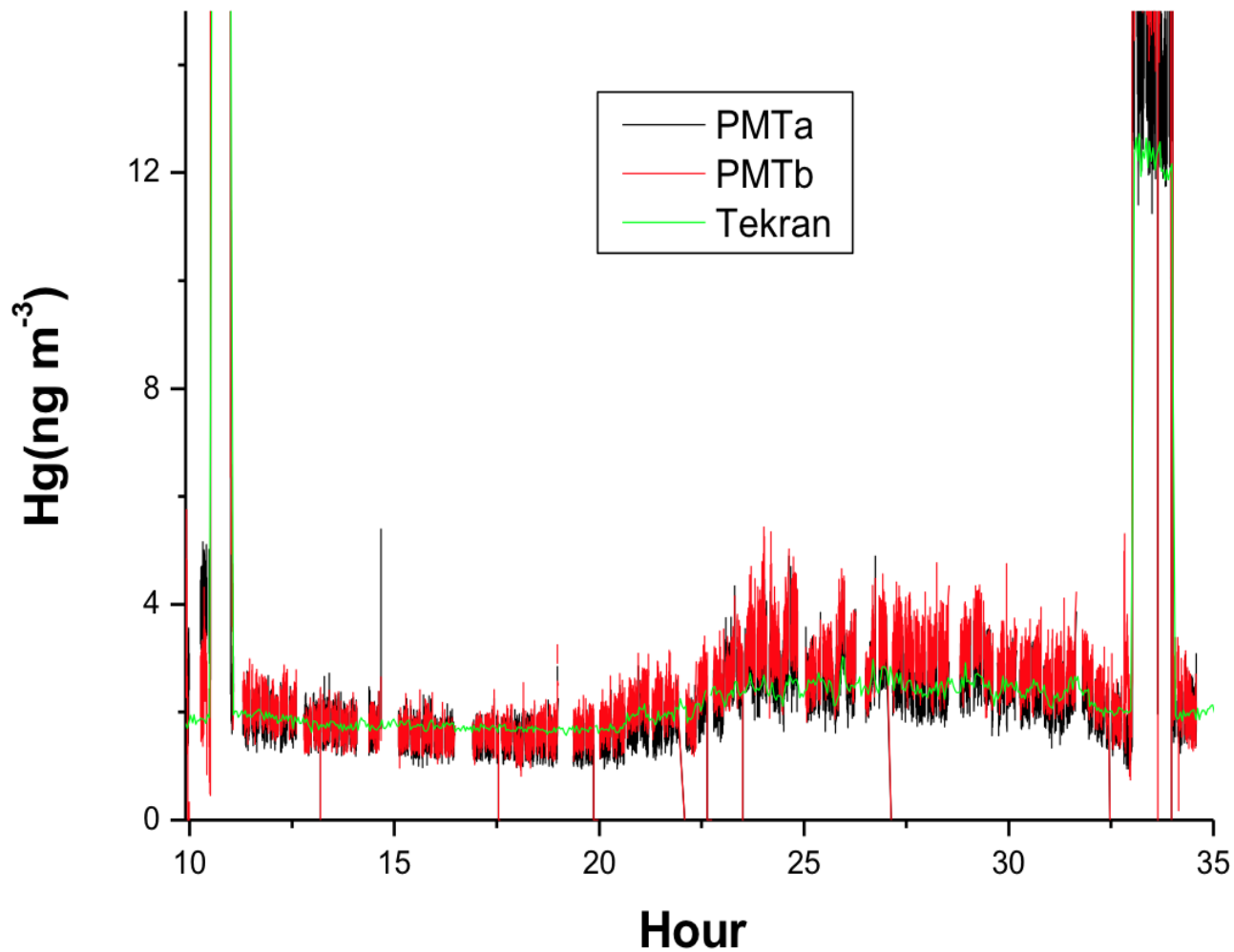


Fig.2 LIF and Tekran signals over a 25 hour sampling period on Sept ½, expanded concentration scale.

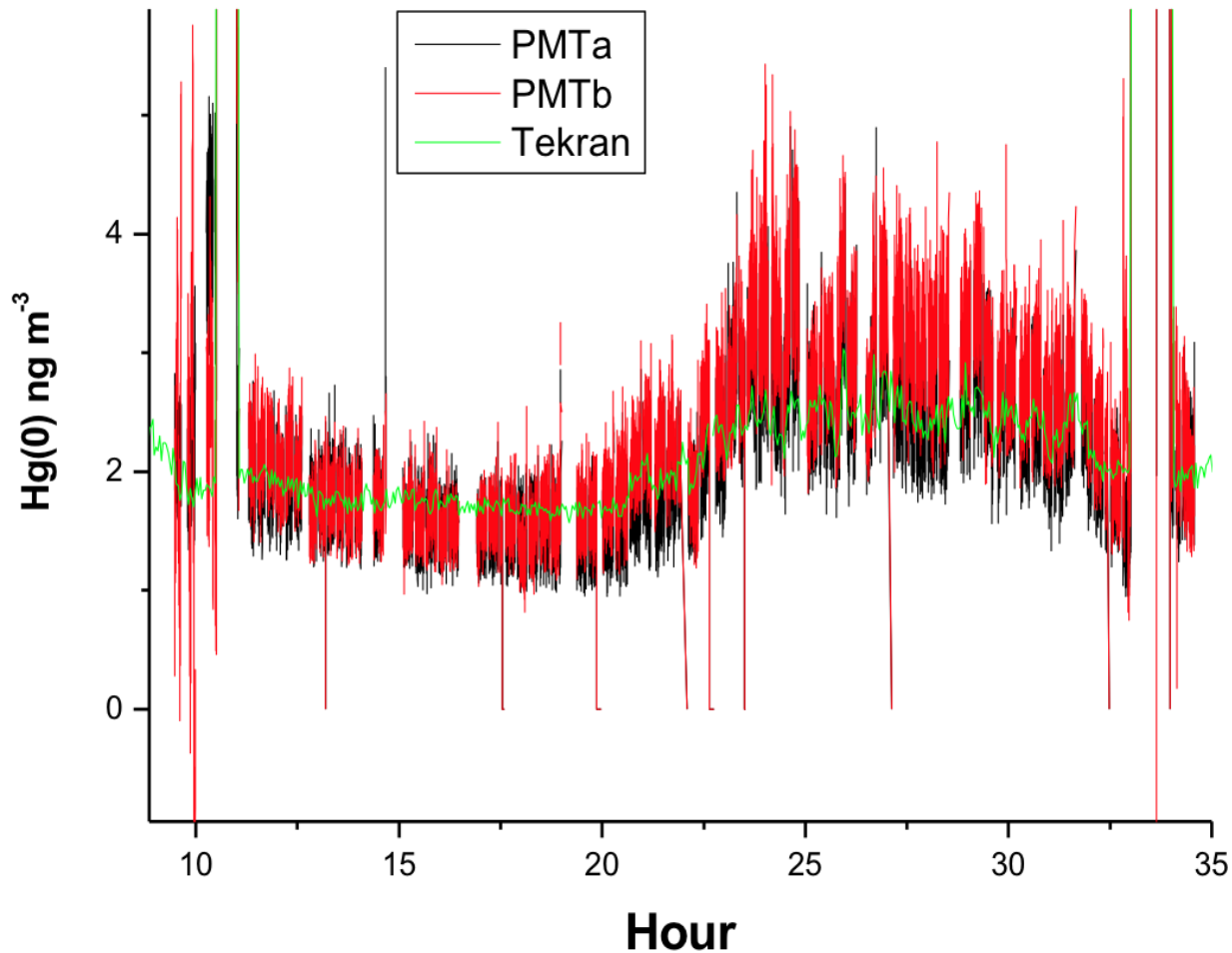
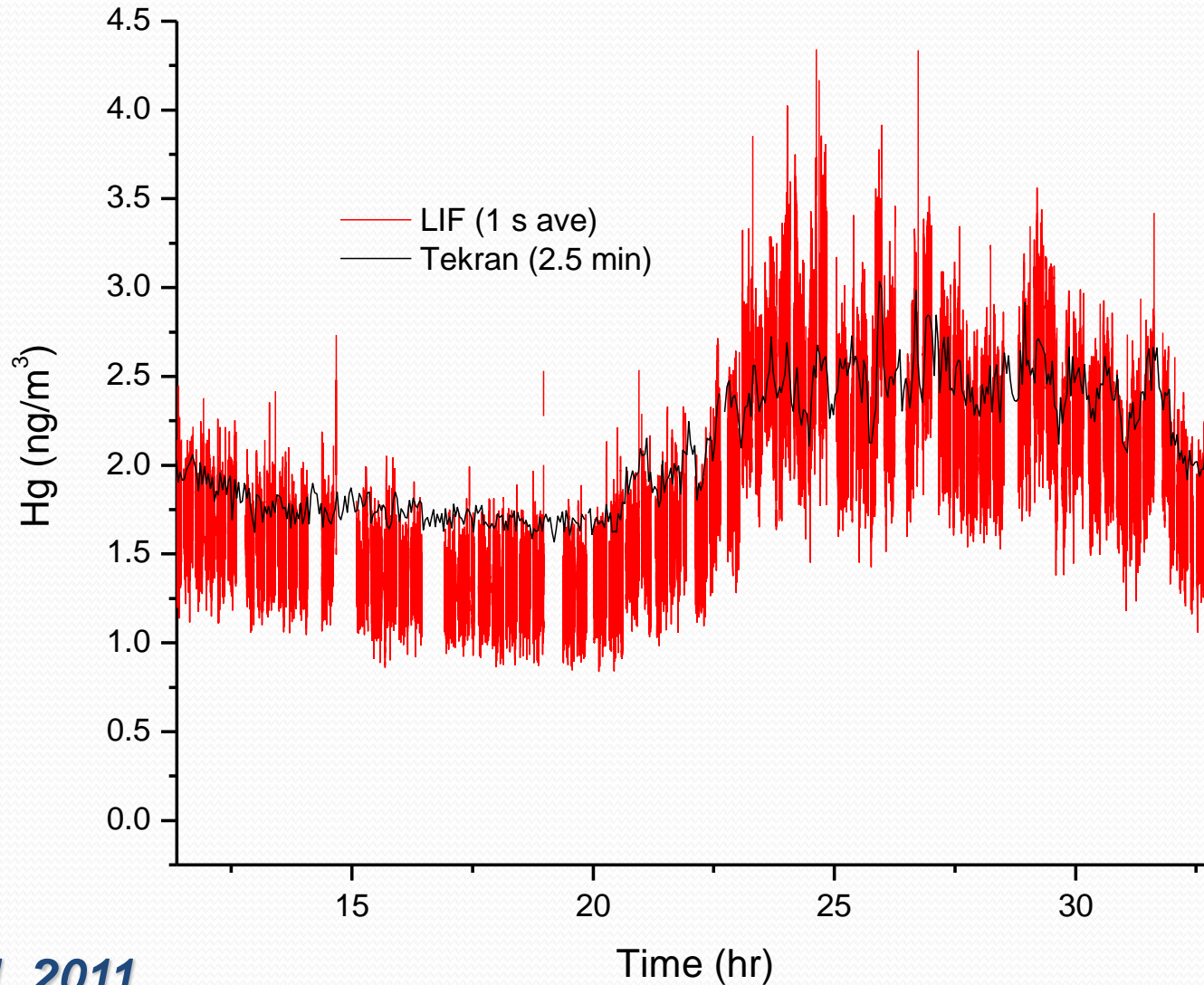


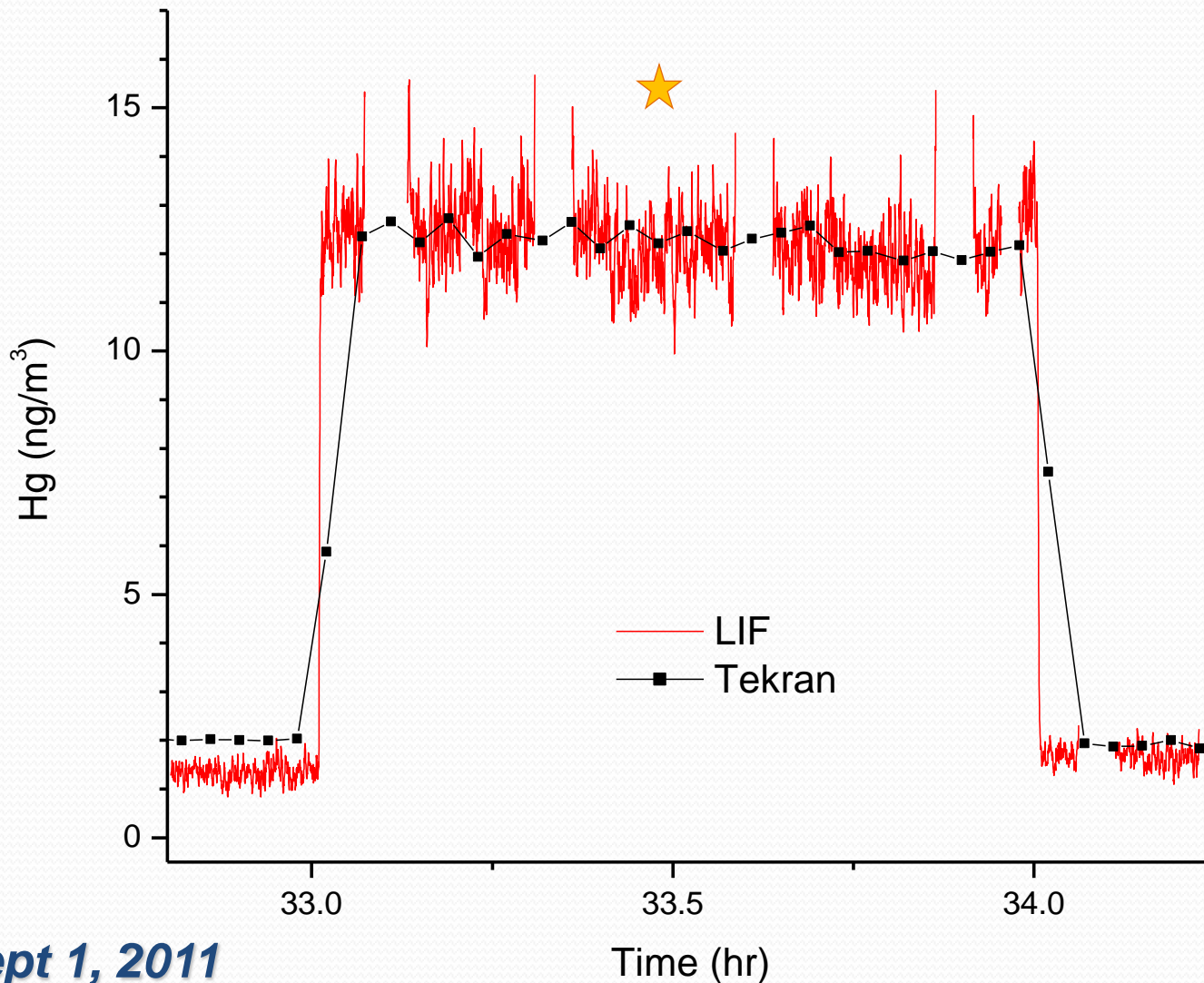
Fig.3 LIF and Tekran signals over a 25 hour sampling period on Sept ½, further expanded concentration scale.

GEM Measurement



Sept 1, 2011

Time Resolution



Rise Time

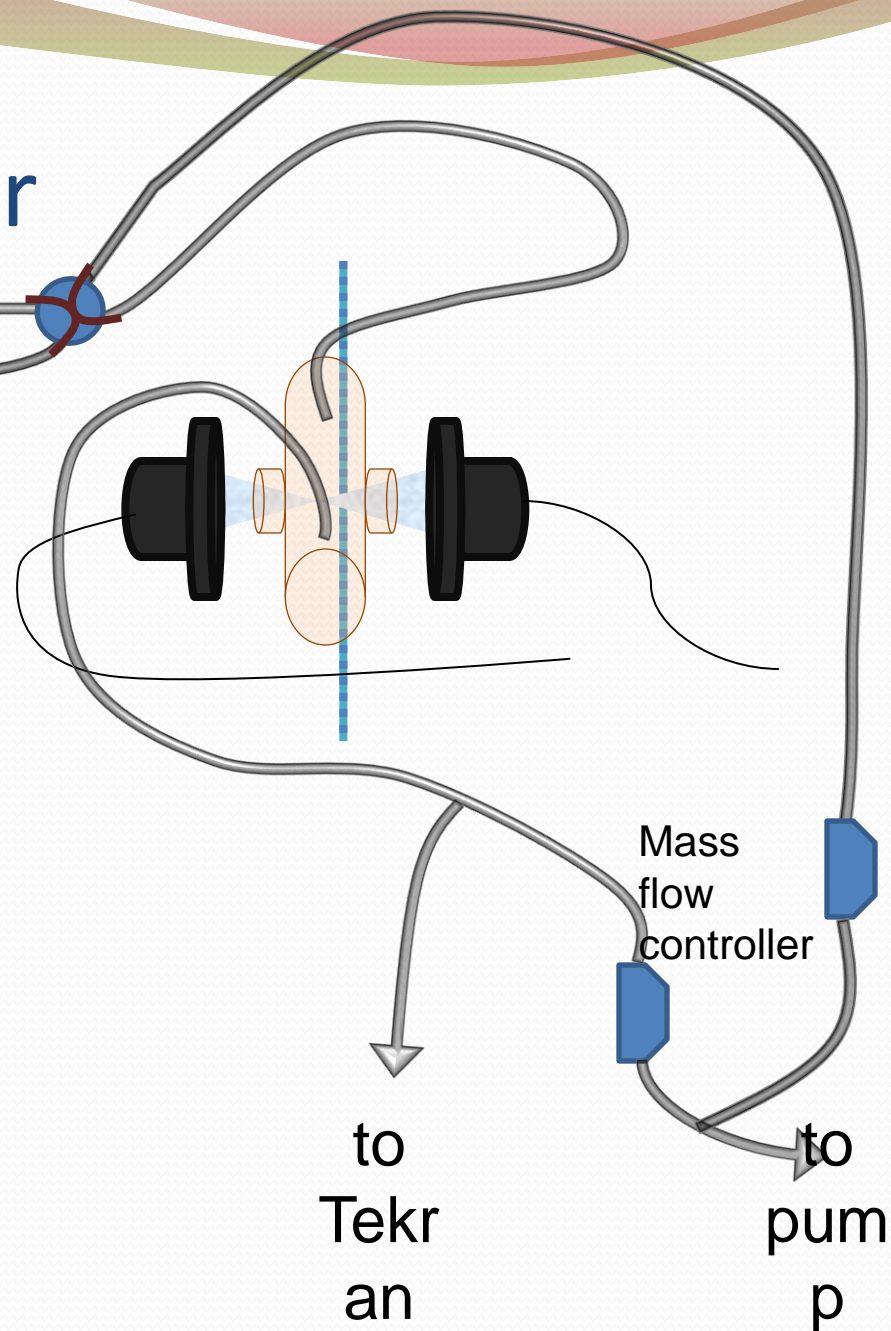
LIF : ≤ 1 s

Tekran : 5 min

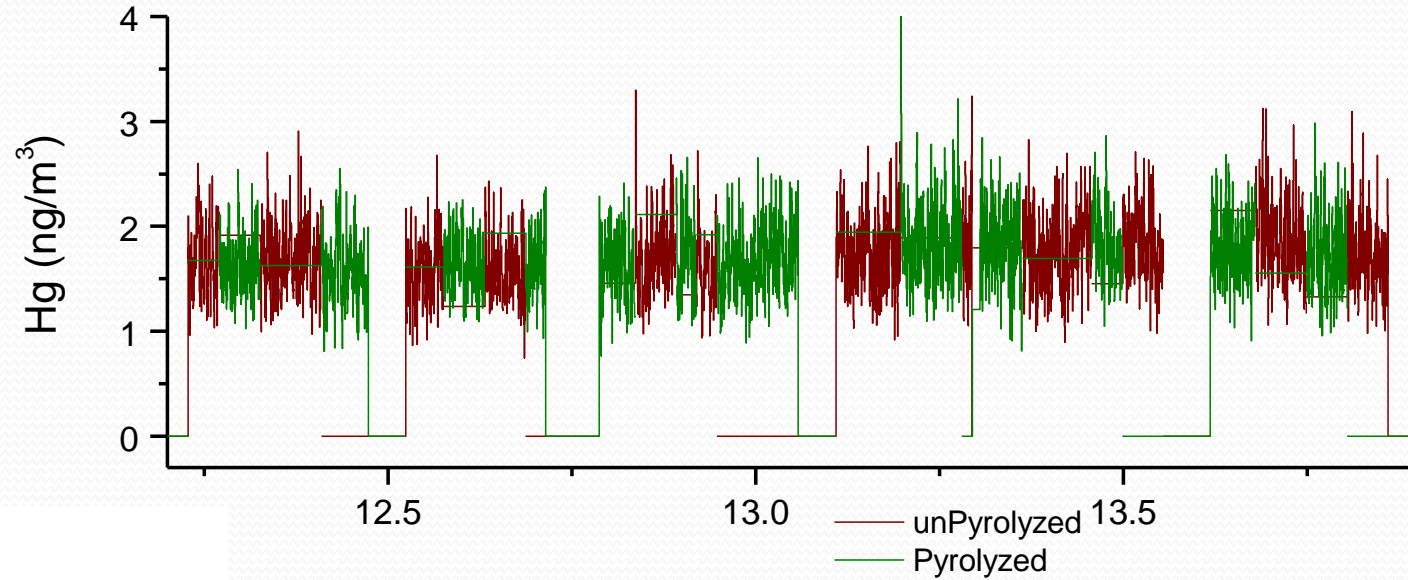
Air Flow - pyrolyzer

from
Manifold
pyrolyzer

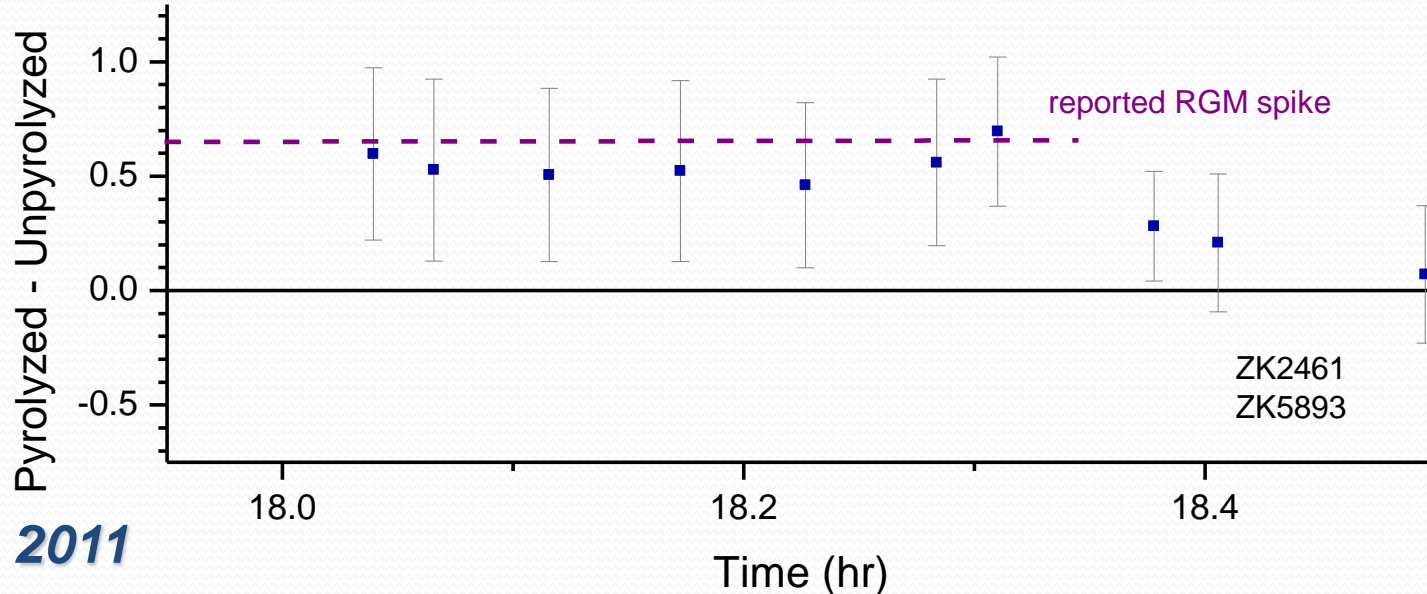
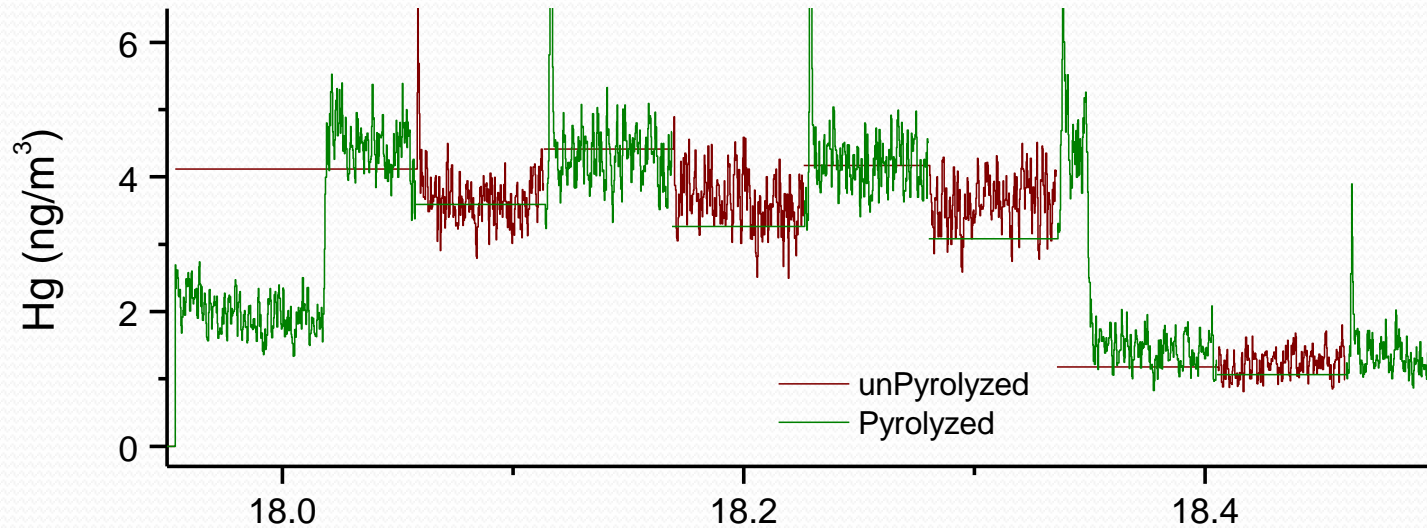
- ~15 cm fused Si tubing (3/8" id) partially filled with Si wool,
- ~ 8 cm wrapped with NiCr wire with constant voltage applied
- Continuous air flow through both pathways



TGM Detection

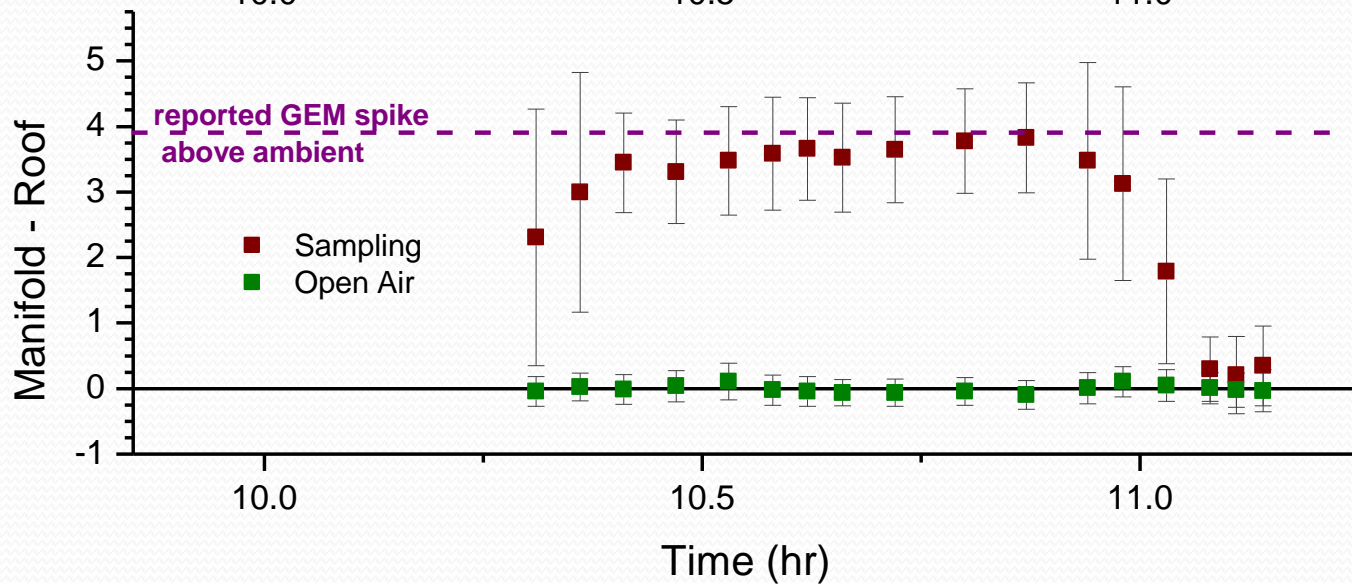
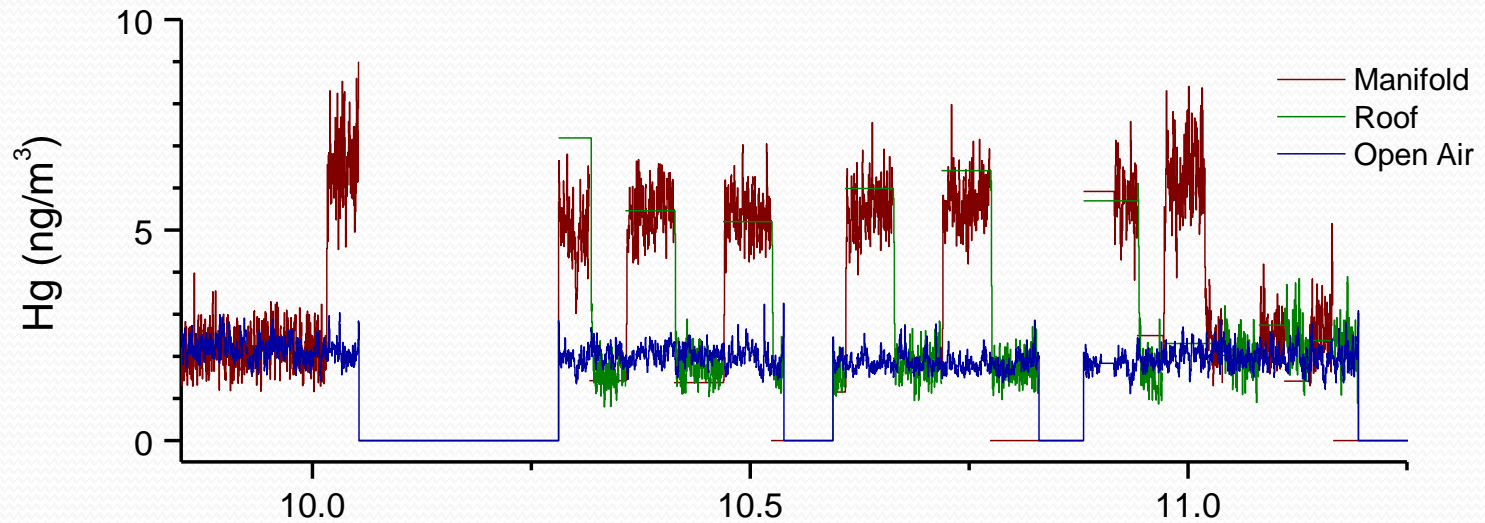


TGM Detection



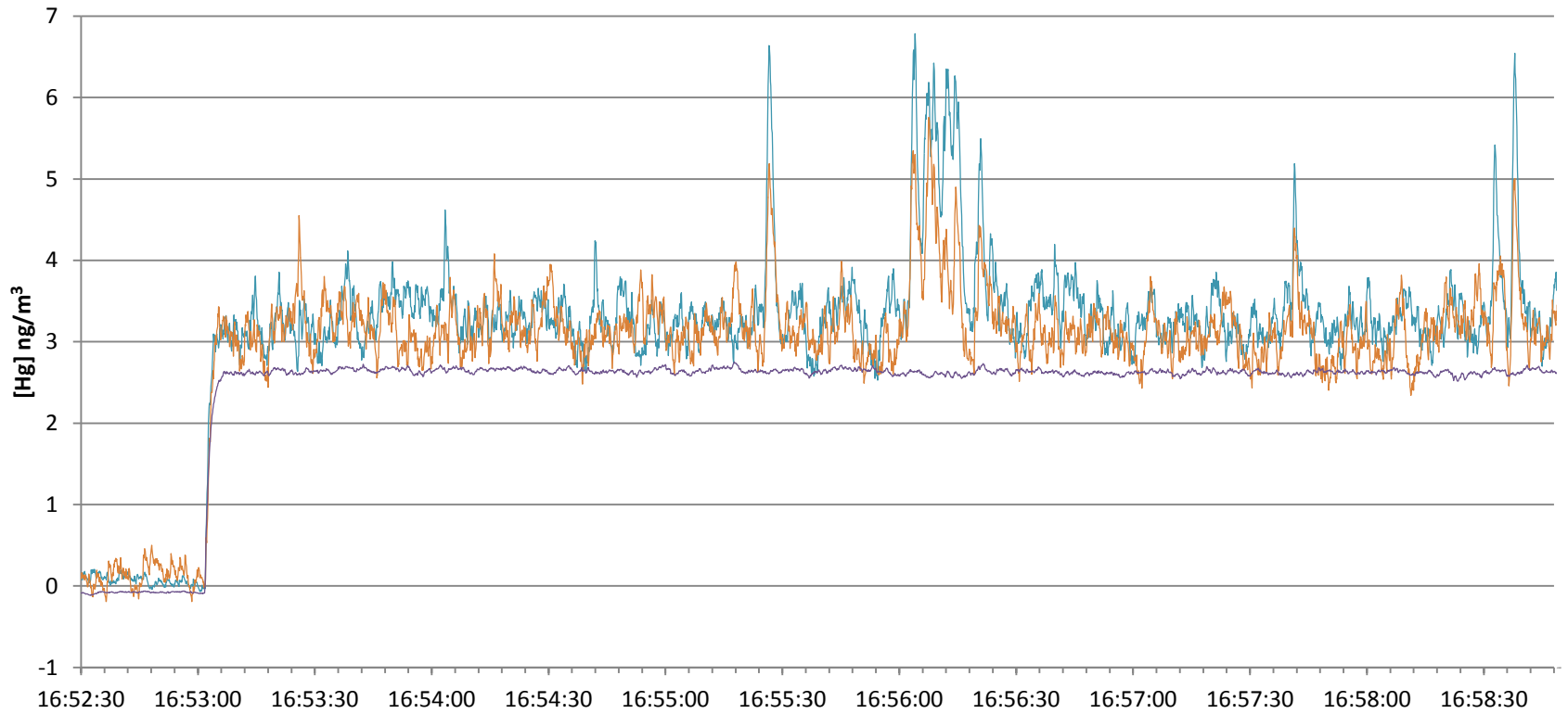
Sept 14, 2011

Open Air Sampling



Boxcar 10 pulse

PMT 1 PMT 2 Ref Ave



Boxcar 10 pulse average

PMT 1 PMT 2 Ref Ave

