

TOGA/ Fast Gas Chromatograph Mass Spectrometer (FGCMS) – prototype for HIAPER

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Gas chromatograph/Mass Spectrometer

Measurement campaigns

TRACE-P
IDEAS
MIRAGE
INTEX-B
ARCTAS
OASIS
MIRAGE-Shanghai

Trace Organic Gas Analyzer (TOGA)



- High sensitivity VOCs/OVOCs to ppt
- High selectivity 2-D GC/MS
- Rapid 2 minutes
- Altitude independent 0-50,000 feet
- semi – Autonomous

Eric Apel

Alan Hills

Rebecca Hornbrook

Dan Riemer

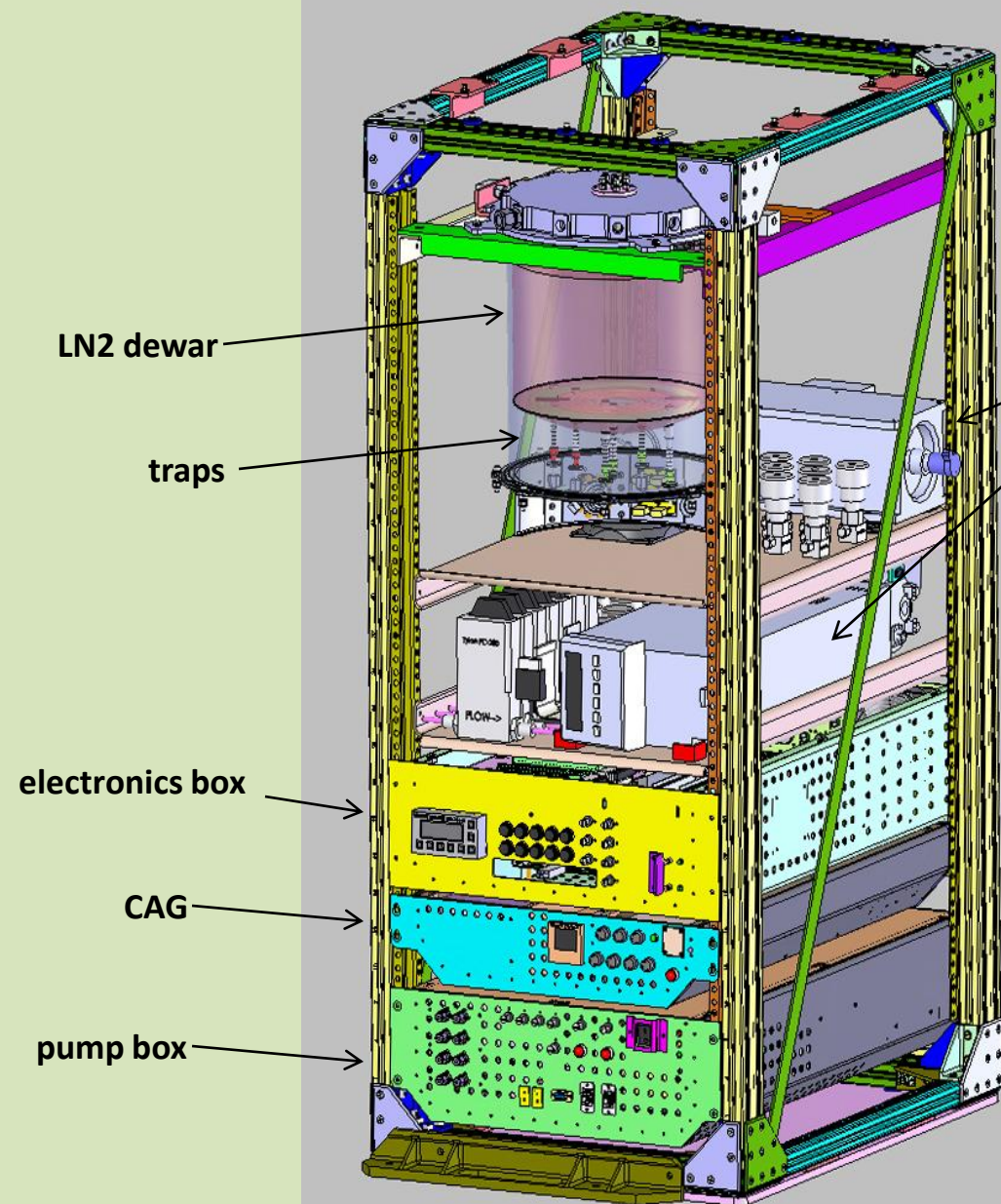
Steve Gabbard

Roger Hendershot

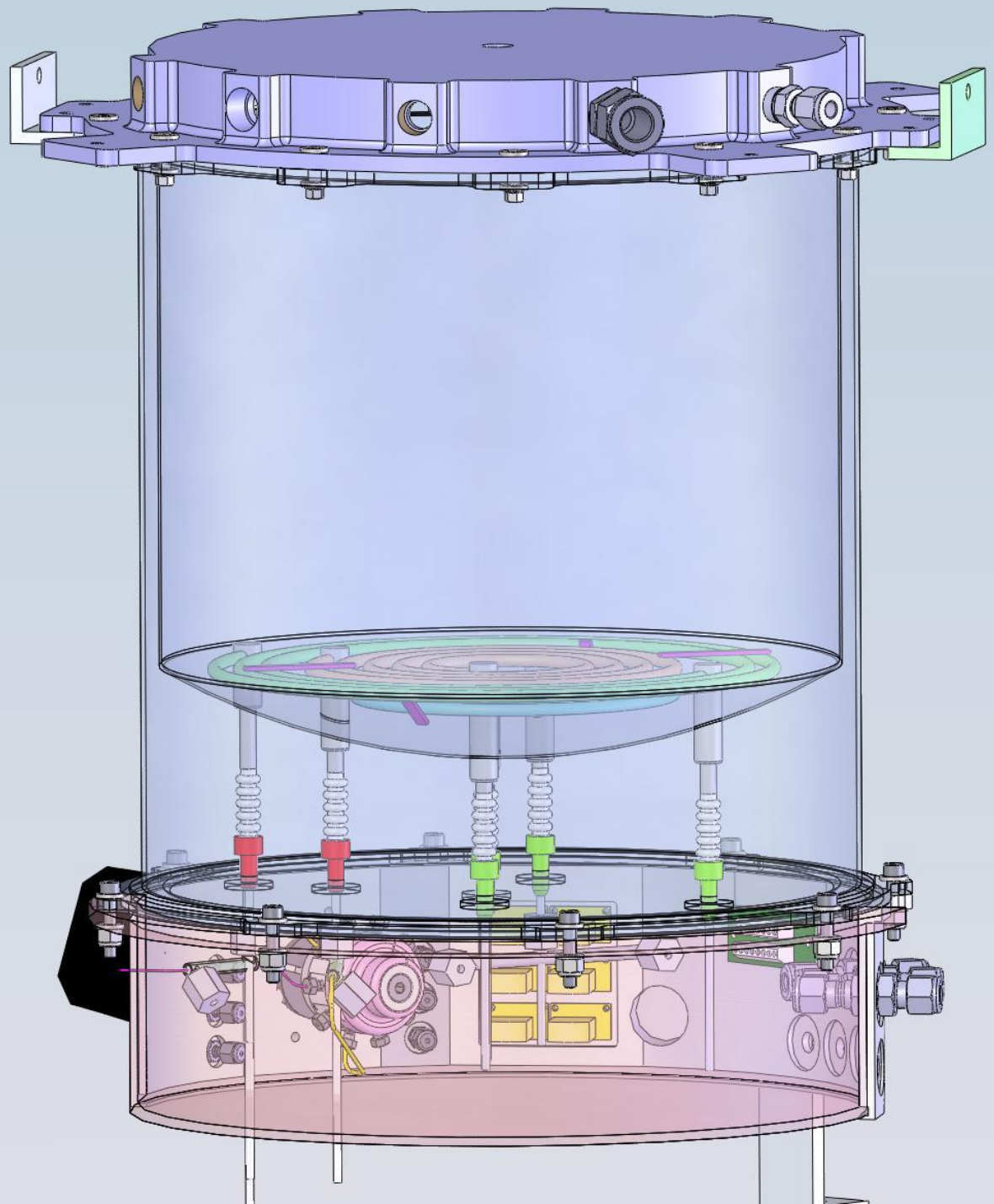
Rudy Montoya



Gulfstream G-V



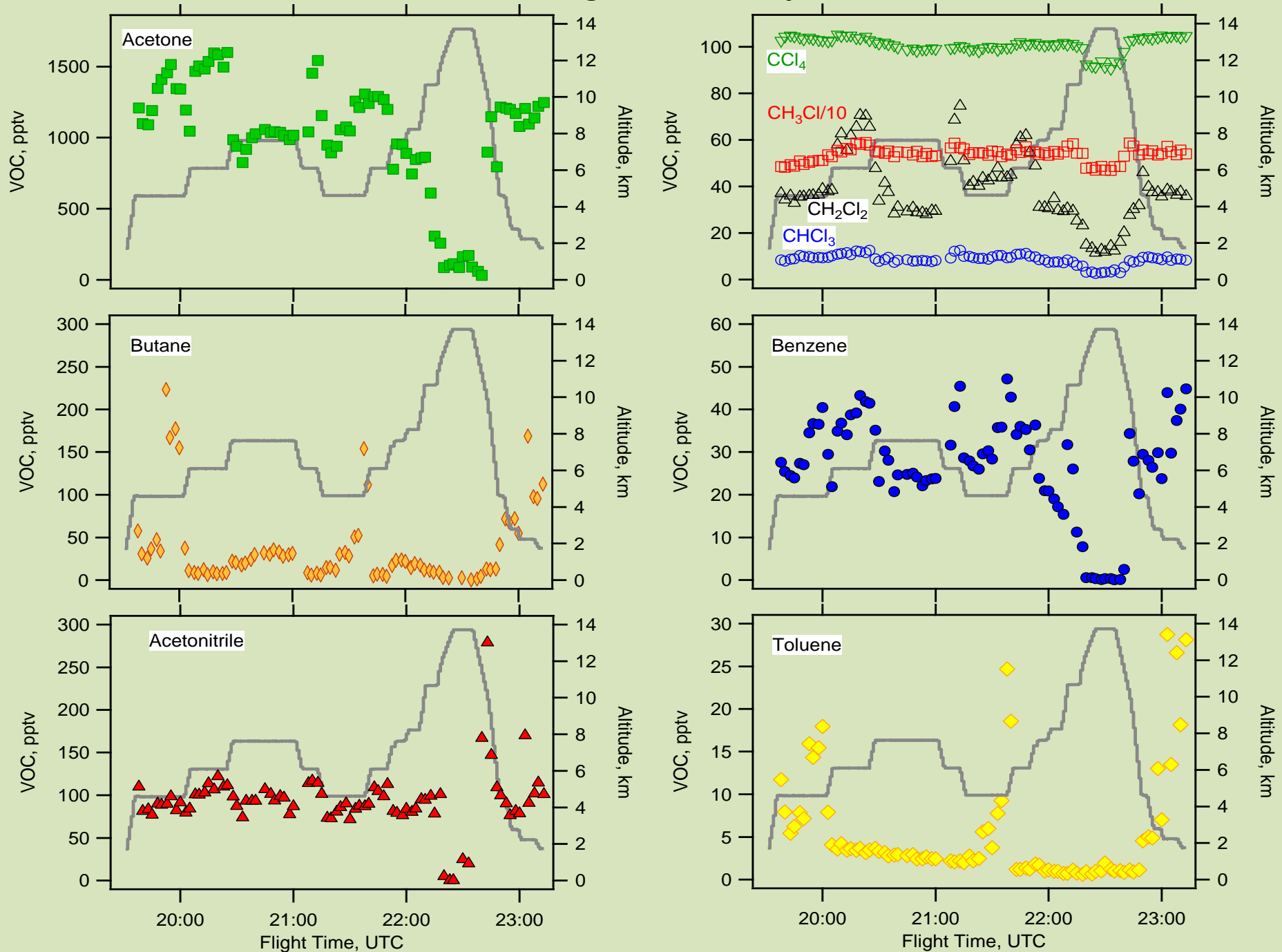
**10-L
Cryogenic LN2 dewar**



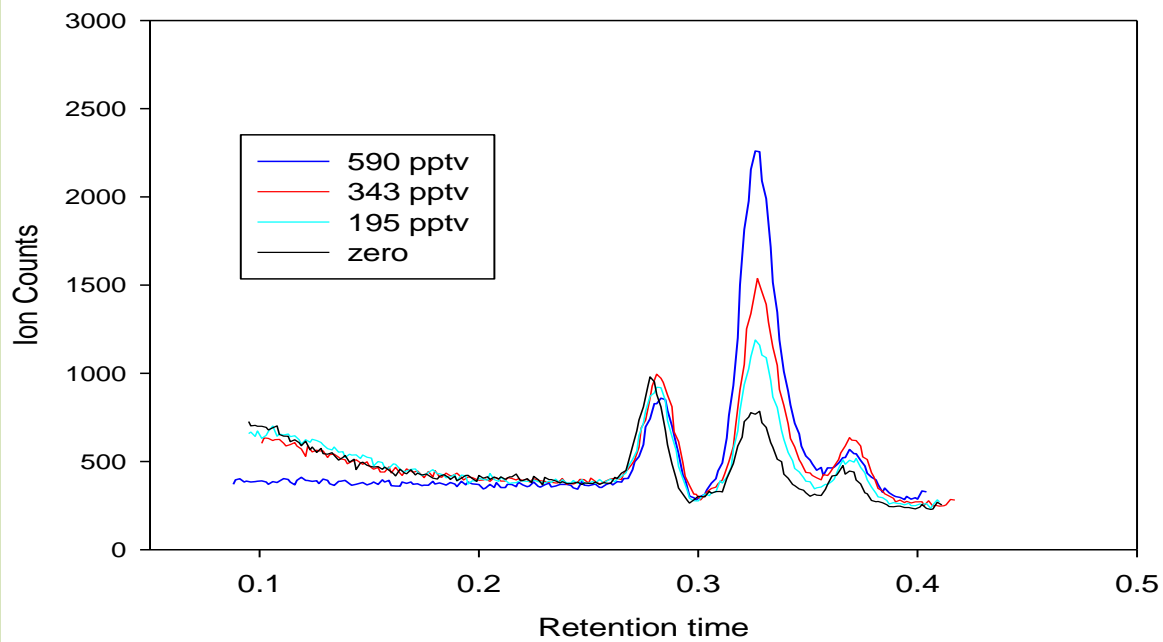
TOGA compounds (may do a subset or switch subsets on the fly)

<p align="center">Hydrocarbons</p>	<p>Propane 1-Butene <i>i</i>-Butene Butane <i>i</i>-Butane Benzene Toluene Ethyl Benzene <i>t</i>-2-Butene <i>c</i>-2-Butene Pentane 1,3-Butadiene Limonene</p>	<p>Isoprene <i>t</i>-2-Pentene <i>c</i>-2-Pentene <i>i</i>-Pentane <i>o</i>-Xylene <i>m/p</i>-Xylene 1,3,5-Trimethylbenzene 1,2,4-Trimethylbenzene α-Pinene β-Pinene Camphene Myrcene</p>	
<p align="center">Oxygenates</p>	<p>Formaldehyde Acetaldehyde Propanal Butanal Pentanal Methacrolein Methyl Vinyl Ketone Methyl Butenol</p>	<p>Methanol Ethanol Acetone Butanone 2-Pentanone 3-Pentanone Methyl t-Butyl Ether</p>	
<p align="center">Halocarbons</p>	<p>Chloroform (CHCl₃) Methylene chloride (CH₂Cl₂) Methyl chloride (CH₃Cl) Methyl bromide (CH₃Br) Tetrachloroethane (CH₂Cl₄) Tetrachloroethylene (C₂Cl₄) Bromoform</p>	<p>Tetrachloromethane (CCl₄) CFC-113 HCFC-141b HCFC-134a 1,2-Dichloroethane (C₂H₄Cl₂) Methyl Iodide (CH₃I) iodoform</p>	<p>dibromomethane diodomethane bromocjhloromethane bromiodomethane chloriodomethane</p>
<p align="center">Nitrogen and sulfur compounds</p>	<p>Acetonitrile Dimethyl Sulfide (DMS)</p>	<p>DMSO?</p>	

TOGA flight data, 6 May 2011

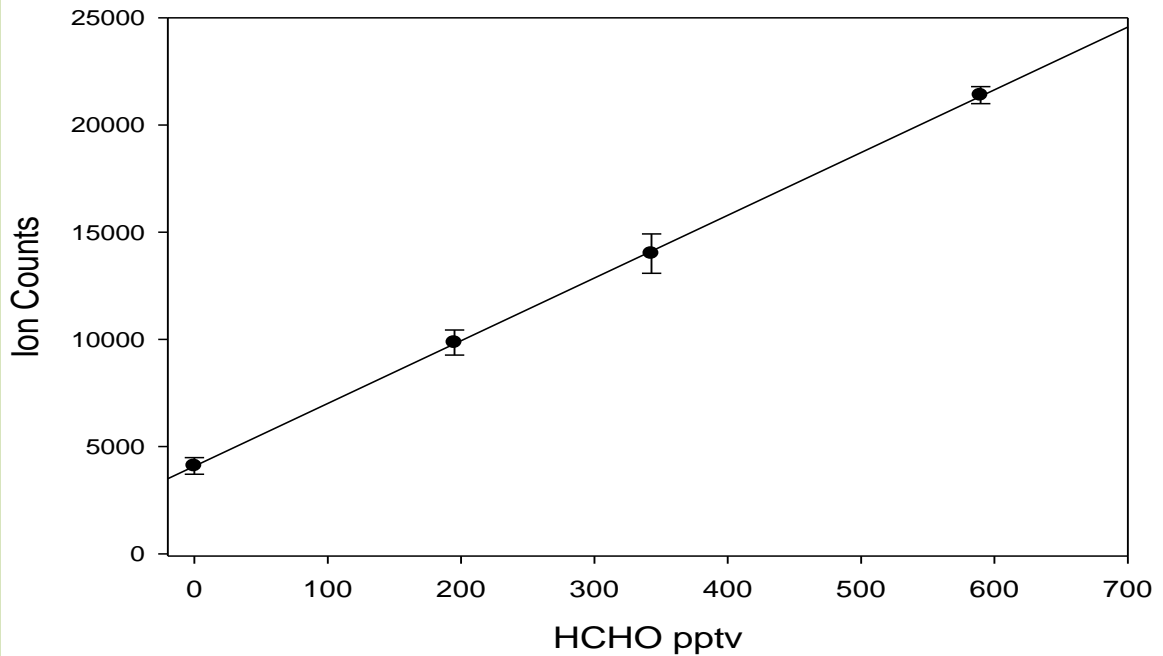


TOGA Formaldehyde Lab Data



Signal-to-noise (S/N) of 195 peak = 35
LOD for 5 x S/N = $195 * 5/35 = 28$ pptv, but because the peak we use for the S/N calculation contains some background, the peak used for the S/N calculation is actually ~ 2x larger than it should be relative to the noise so the detection limit at 5x is closer to 50-60 pptv.

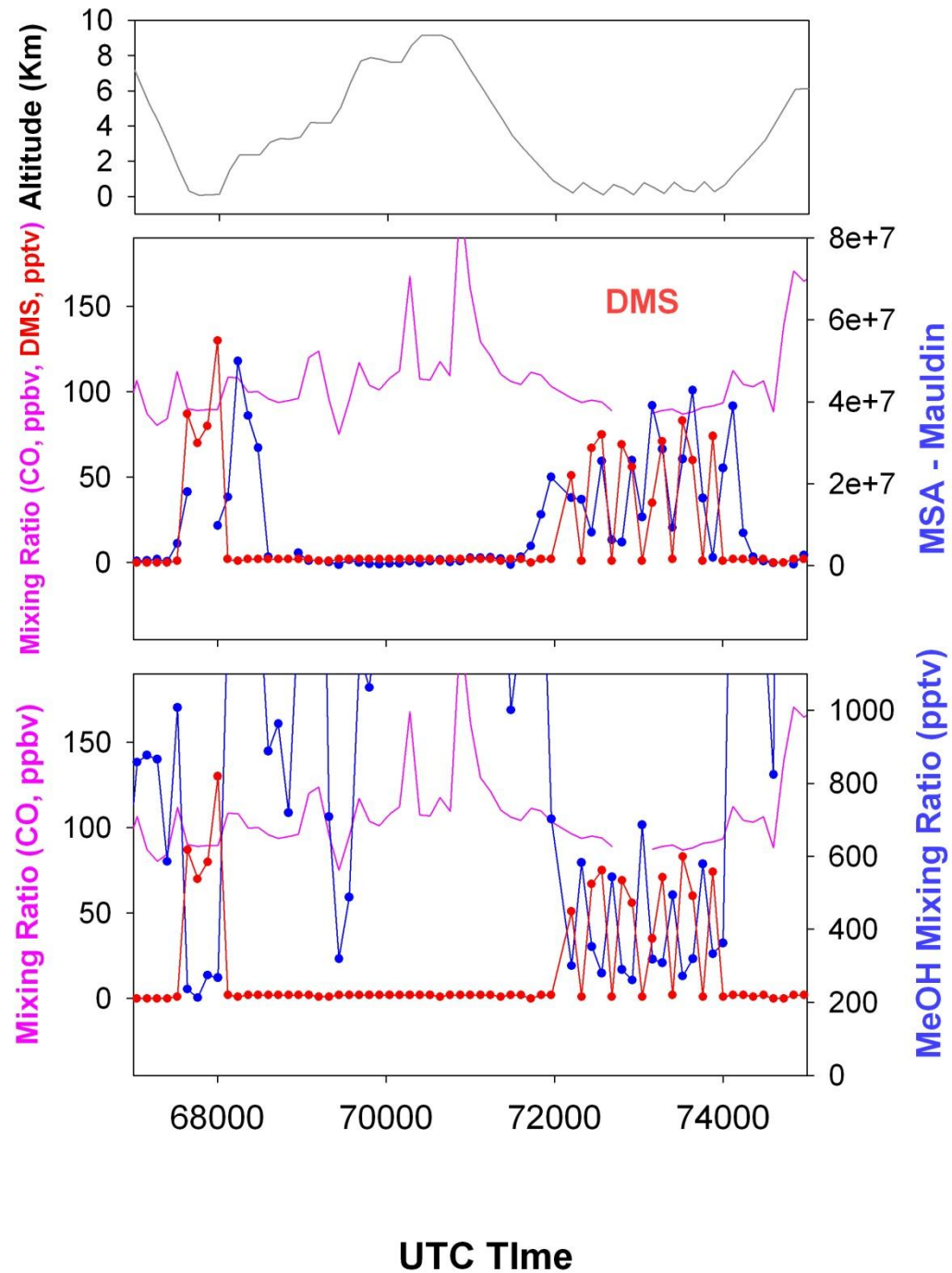
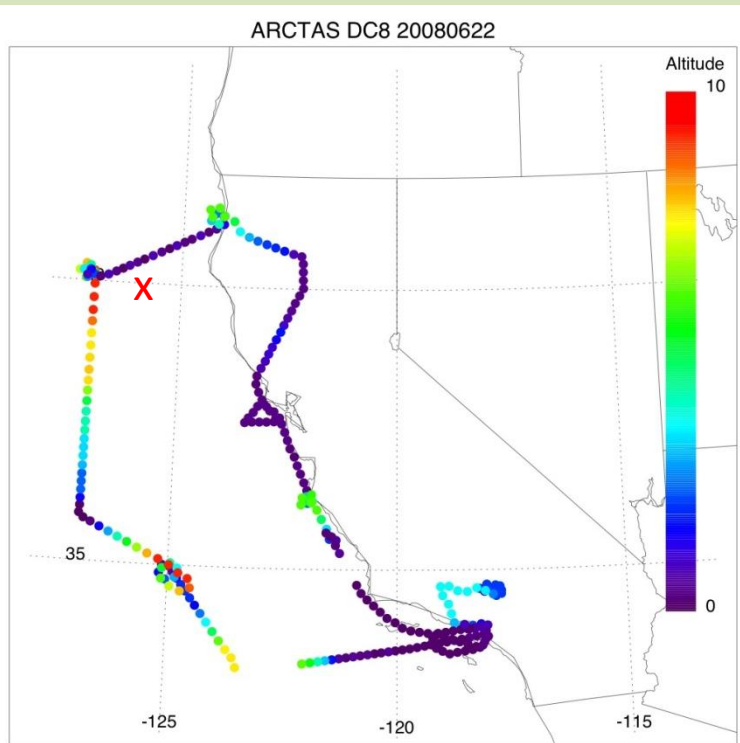
Formaldehyde low concentration study



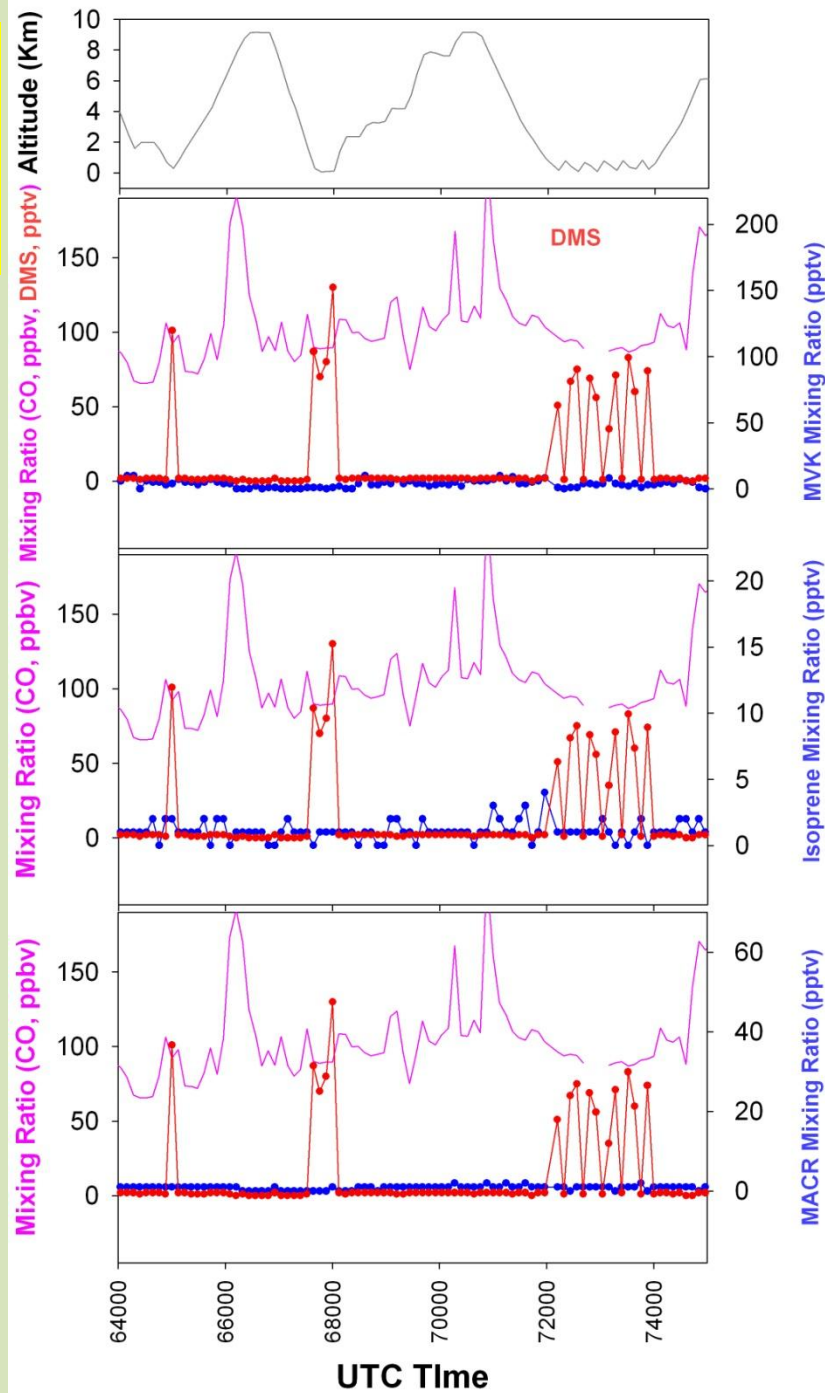
Calibration curve for the TOGA low concentration formaldehyde study. Five measurements were obtained for each data point using formaldehyde standards introduced to TOGA. The standards were generated using the permeation system described in Figure 2 coupled to the dilution apparatus described in Figure 4.

DMS
MSA

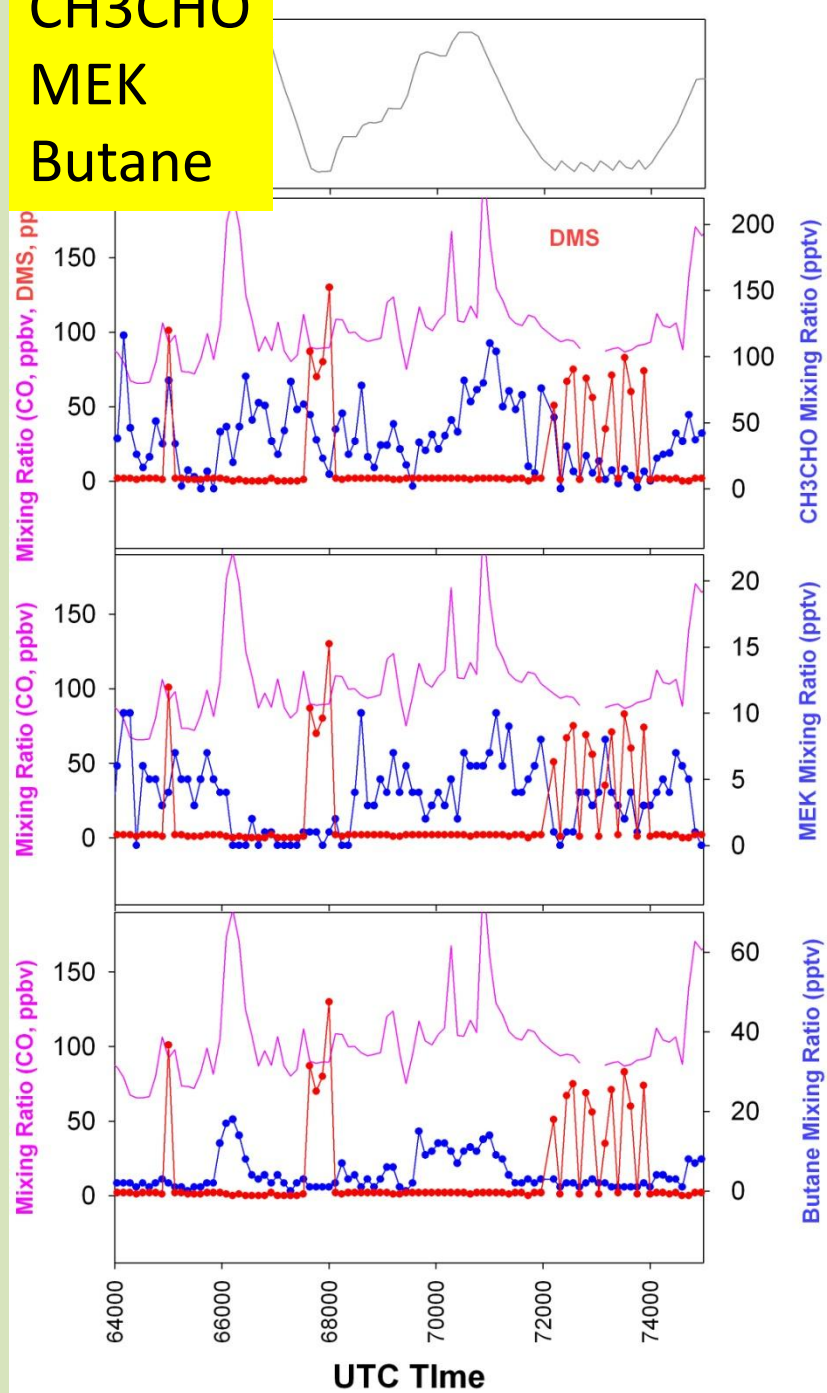
ARCTAS-CARB



ISO
MVK
MACR

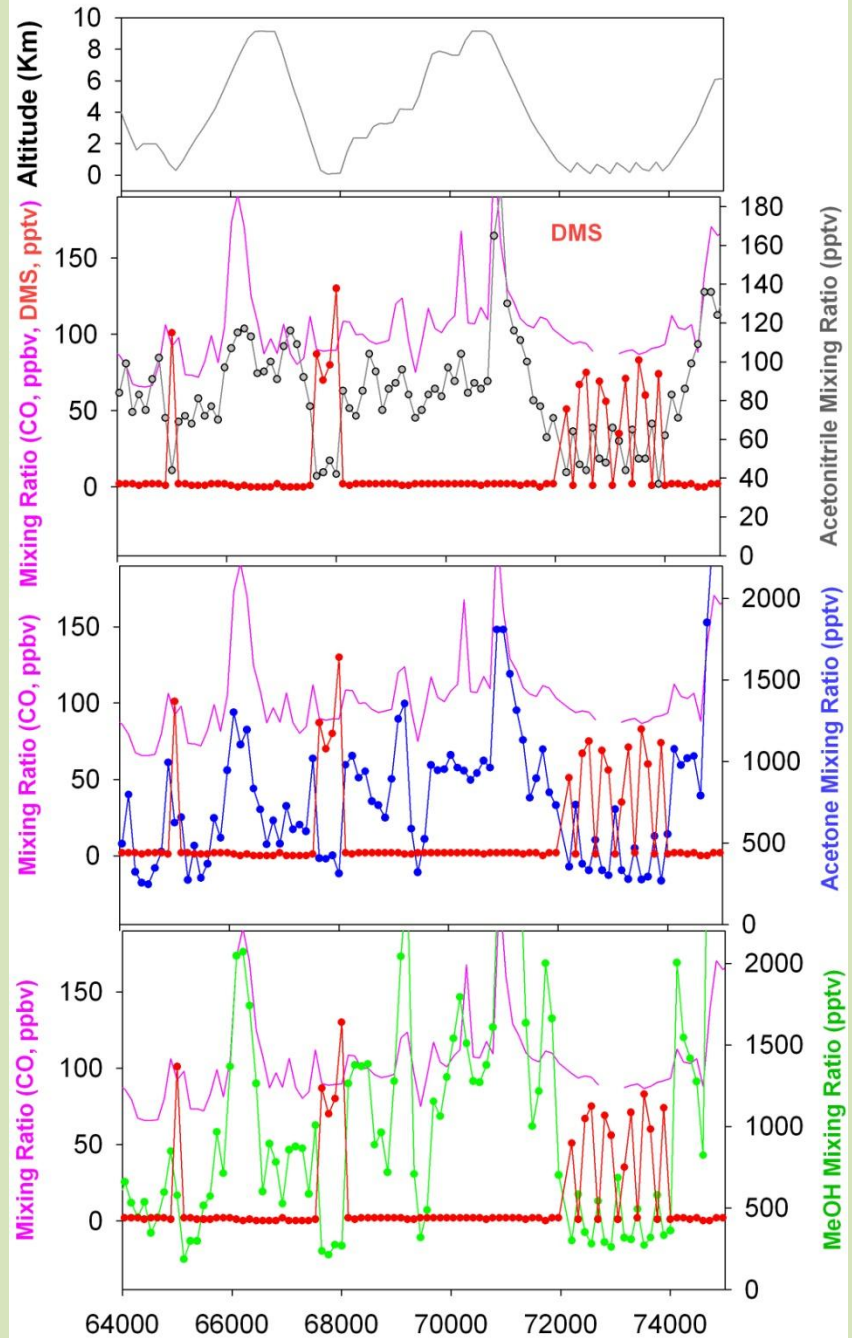


CH3CHO
MEK
Butane



CH₃CN
Acetone
MeOH

All species
depleted in MBL

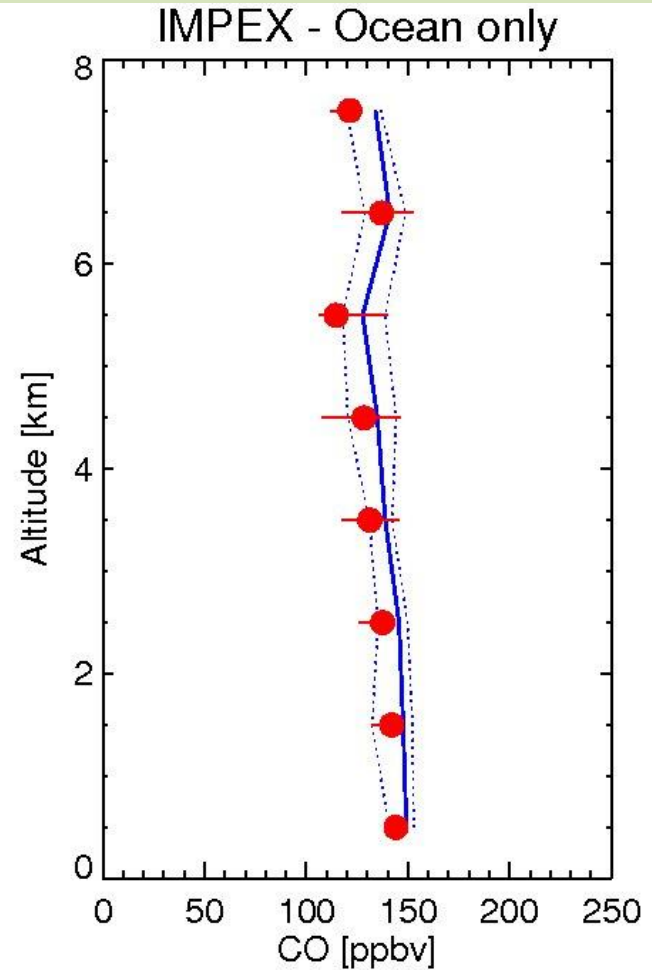
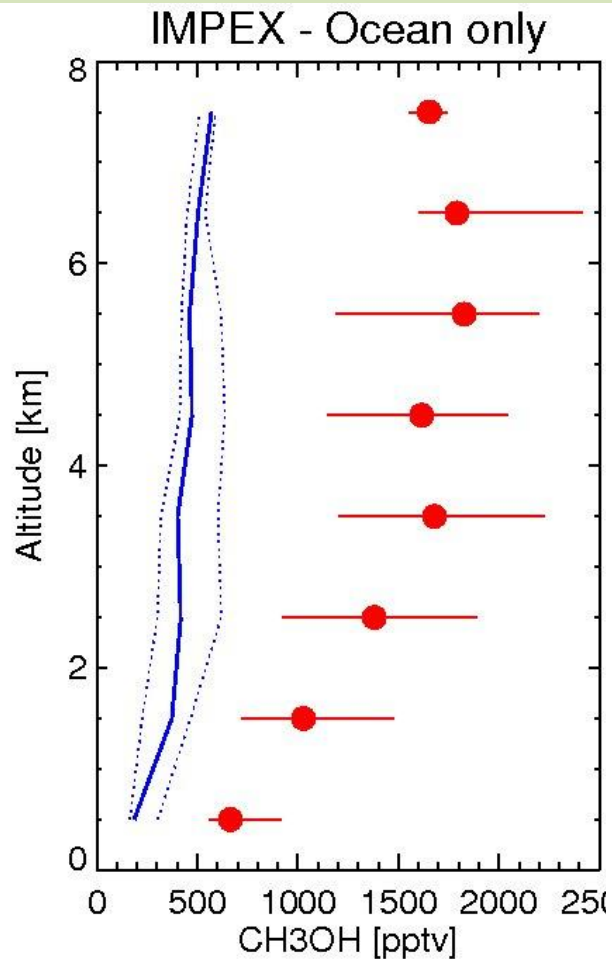
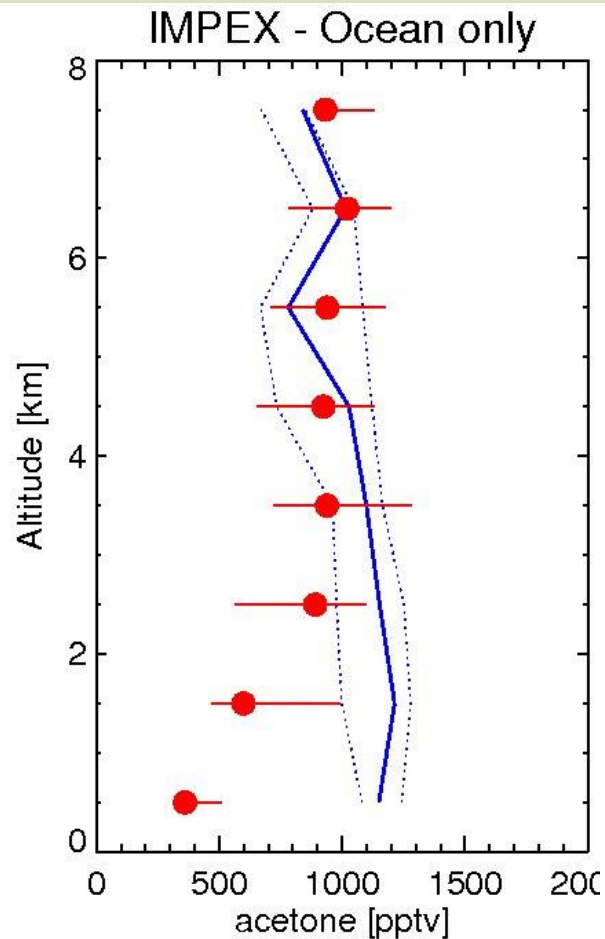


TRACE Organic Gas Analyzer - TOGA - VOC measurements INTEX-B

Role of Ocean in Global Atmospheric Budgets

Oxygenates: acetone, methanol

CO



● Median Values - TOGA

— MOZART

Acetone $-4.7 \text{ um m}^2 \text{ day}^{-1}$

Methanol $-3.3 \text{ um m}^2 \text{ day}^{-1}$

ROLE OF OCEAN IN ATMOSPHERIC BUDGETS OF OVOCs

SINK?

Physical uptake
limited by both gas –
and aqueous –
phase transfer

$$F_{\text{top}} = V * \Delta C$$

$h\nu$

Organic
microlayer

VOCs



$$F_{\text{MBL}} = \text{prod} - \text{loss}$$

MBL

Sea salt spray

$h\nu$

Organic microlayer

$$F_{\text{ocean}} = -F_{\text{mbl}} - F_{\text{top}}$$

OCEAN

Previous Work:

Methanol, Acetone, etc.. – large ocean source – Nature
Singh et al, 2001

– small ocean sink (Singh et al, 2003)

Acetone: Marandino et al (Saltzman (2005)) – Fluxes of Acetone

- large ocean sink

- Gradient fluxes of $\sim -3 -12 \text{ } \mu\text{m m}^2 \text{ day}^{-1}$ (Eddy Co-variance)

Preliminary results (back of envelope) here:

Flux (Assume MBL - MBLP = 0)

Acetone -5.5 $\mu\text{m m}^2 \text{ day}^{-1}$ (Saltzman - ~ - 9 $\mu\text{m m}^2 \text{ day}^{-1}$)

Methanol-8.32 $\mu\text{m m}^2 \text{ day}^{-1}$

CH₃CN ran out of time this morning..

Preliminary results w/ losses:

Flux (without boundary layer loss but not production)

Acetone -4.7 $\mu\text{m m}^2 \text{ day}^{-1}$

Methanol-3.3 $\mu\text{m m}^2 \text{ day}^{-1}$

CH₃CN ran out of time this morning..